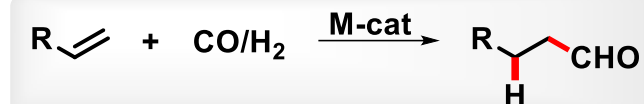
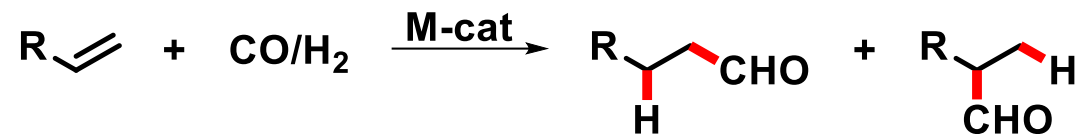


➤ Hydroformylation



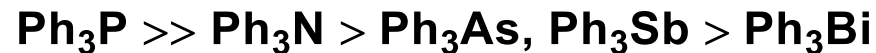
- General scheme



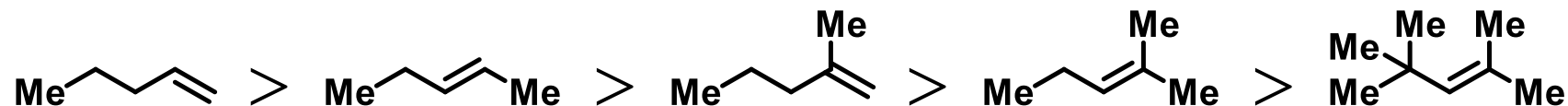
- Catalyst activity



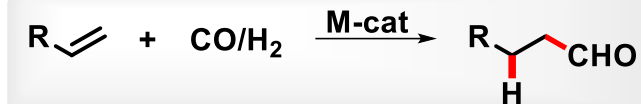
- Ligand activity for Rhodium catalysts



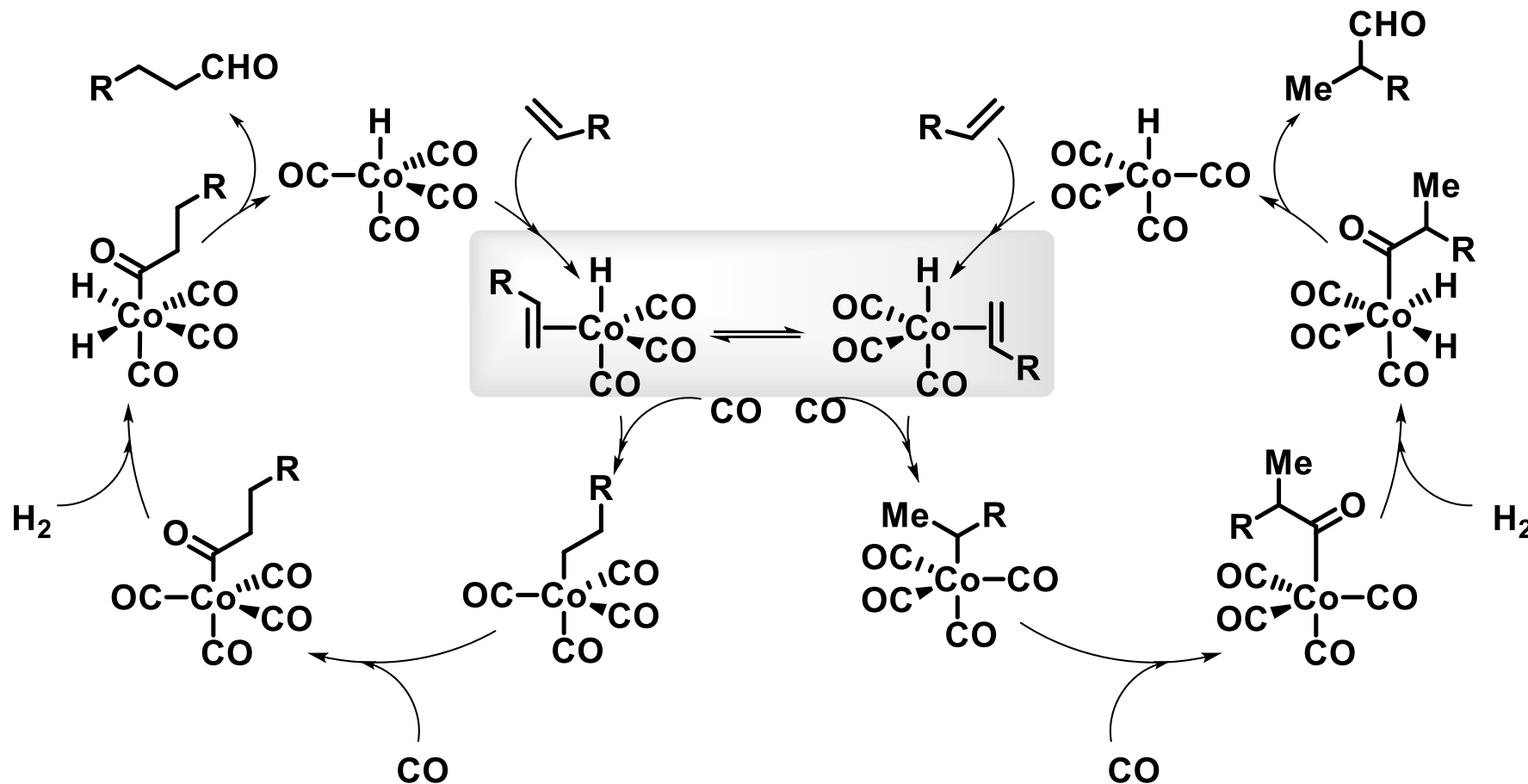
- Rate of the alkene hydroformylation



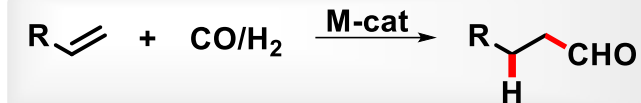
➤ Hydroformylation



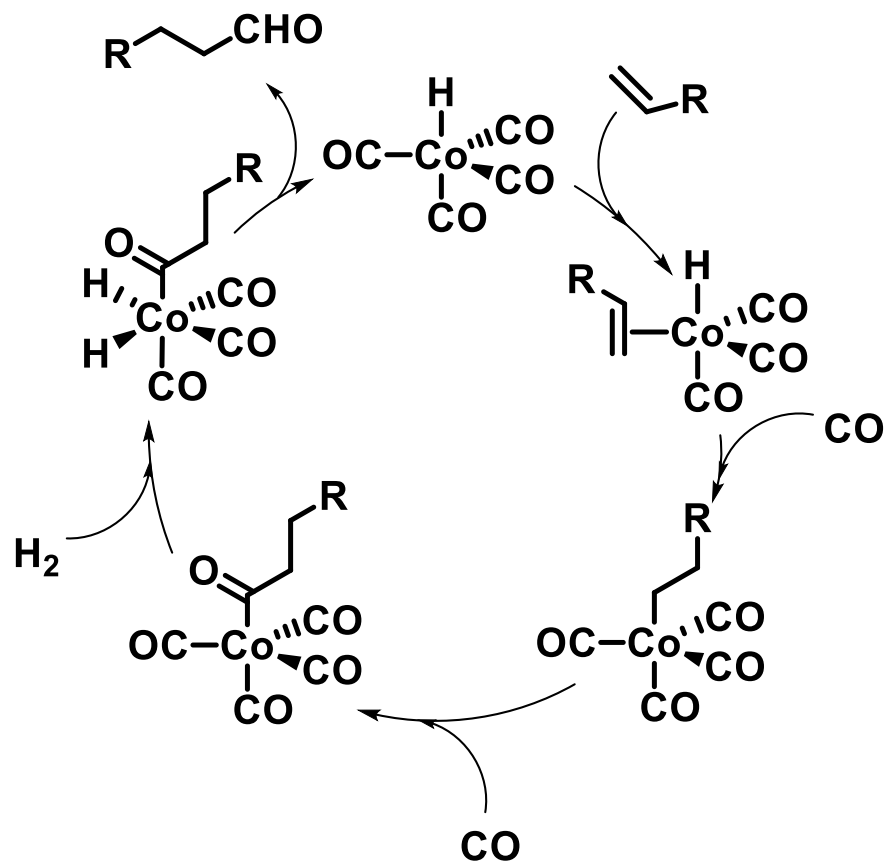
- Widely accepted mechanism for $\text{HCo}(\text{CO})_4$ catalyzed hydroformylation



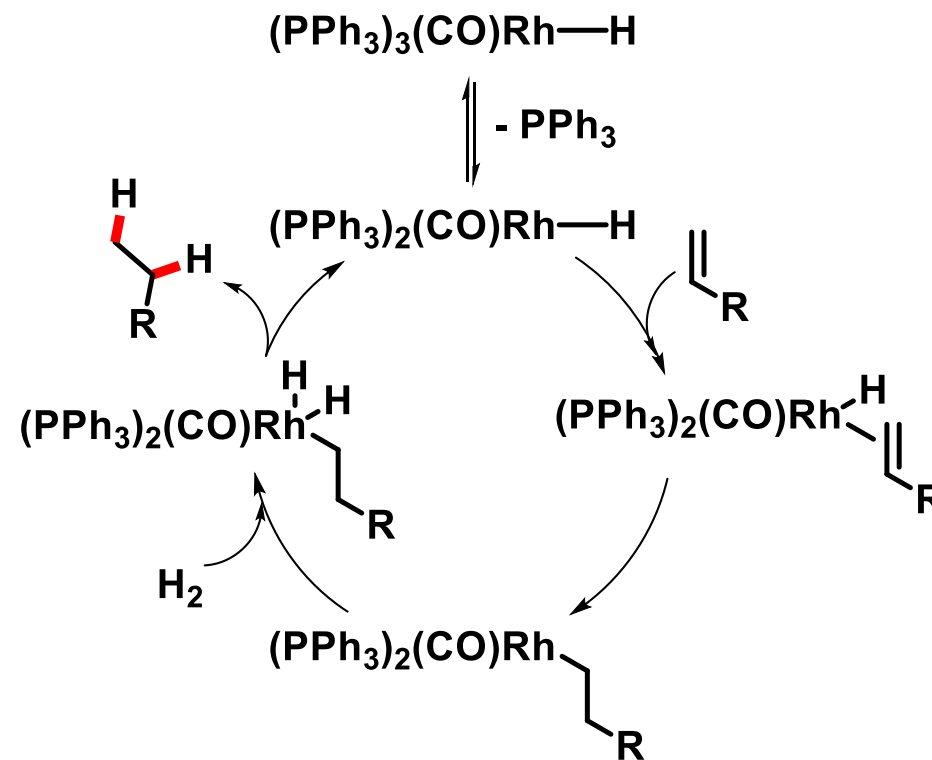
➤ Hydroformylation



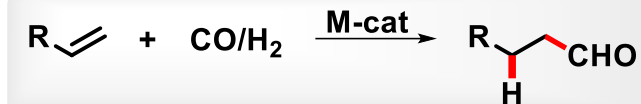
- Mechanism of hydroformylation by [Co(H)CO]



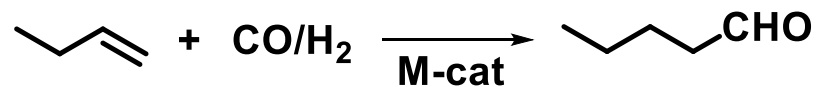
- Mechanism of hydrogenation by [Rh(H)CO]



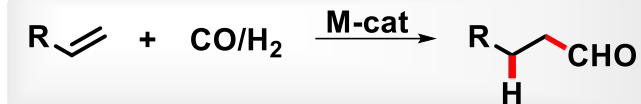
➤ Hydroformylation



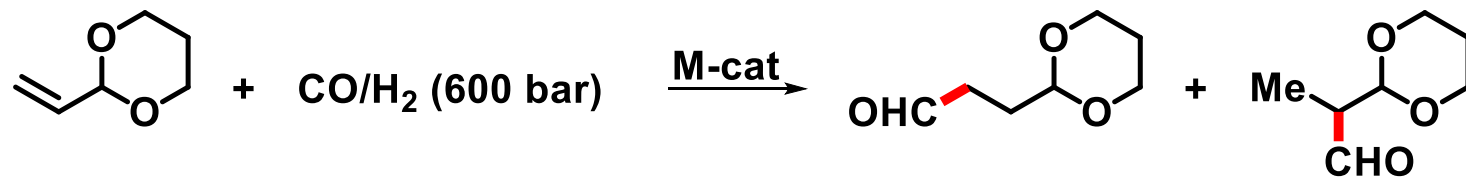
- Industrial production of aldehydes (BASF, Oxea, Eastman, Dow Chemical)



➤ Hydroformylation



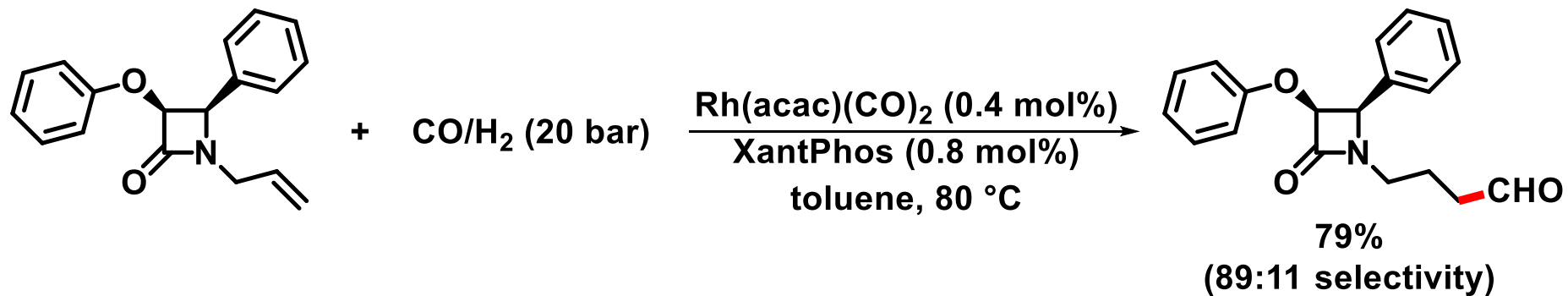
- Selected examples of hydroformylation reaction



Rh-cat, CO/H₂ (600 bar), 100 °C 25 :75

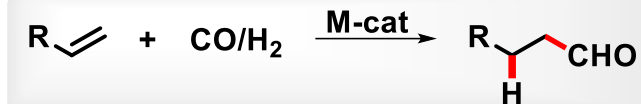
Rh-PPh₃, CO/H₂ (20 bar), 140 °C 70:30

German Patent DE 2111116, 1972; U.S. Patent 4017550, 1977

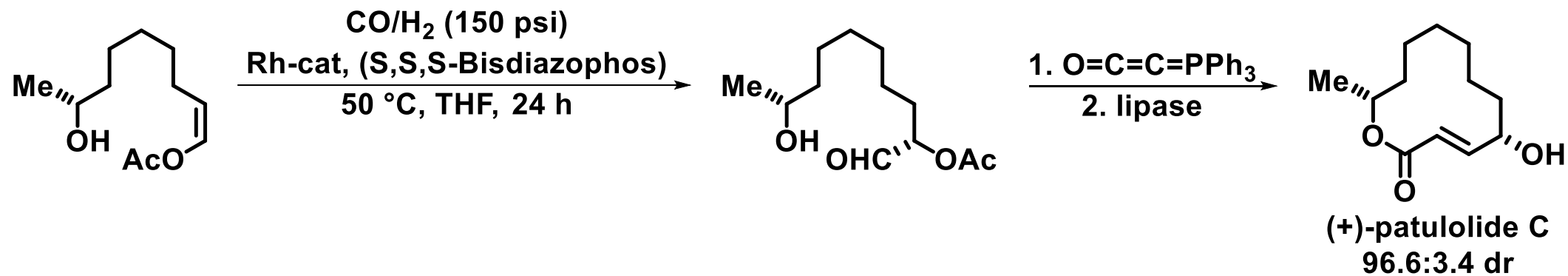


New. J. Chem. 2010, 34, 1079

➤ Hydroformylation



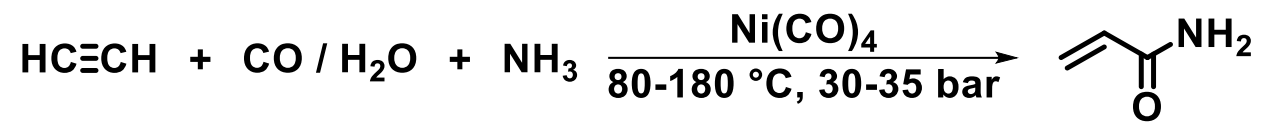
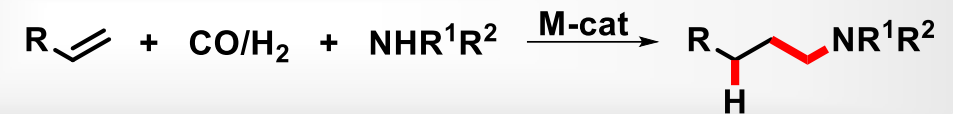
- Asymmetric hydroformylation



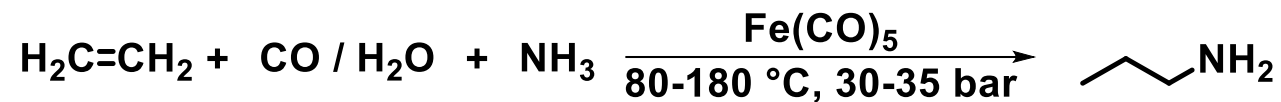
Org. Lett. 2012, 14, 1180

➤ Hydroaminomethylation

- Early examples



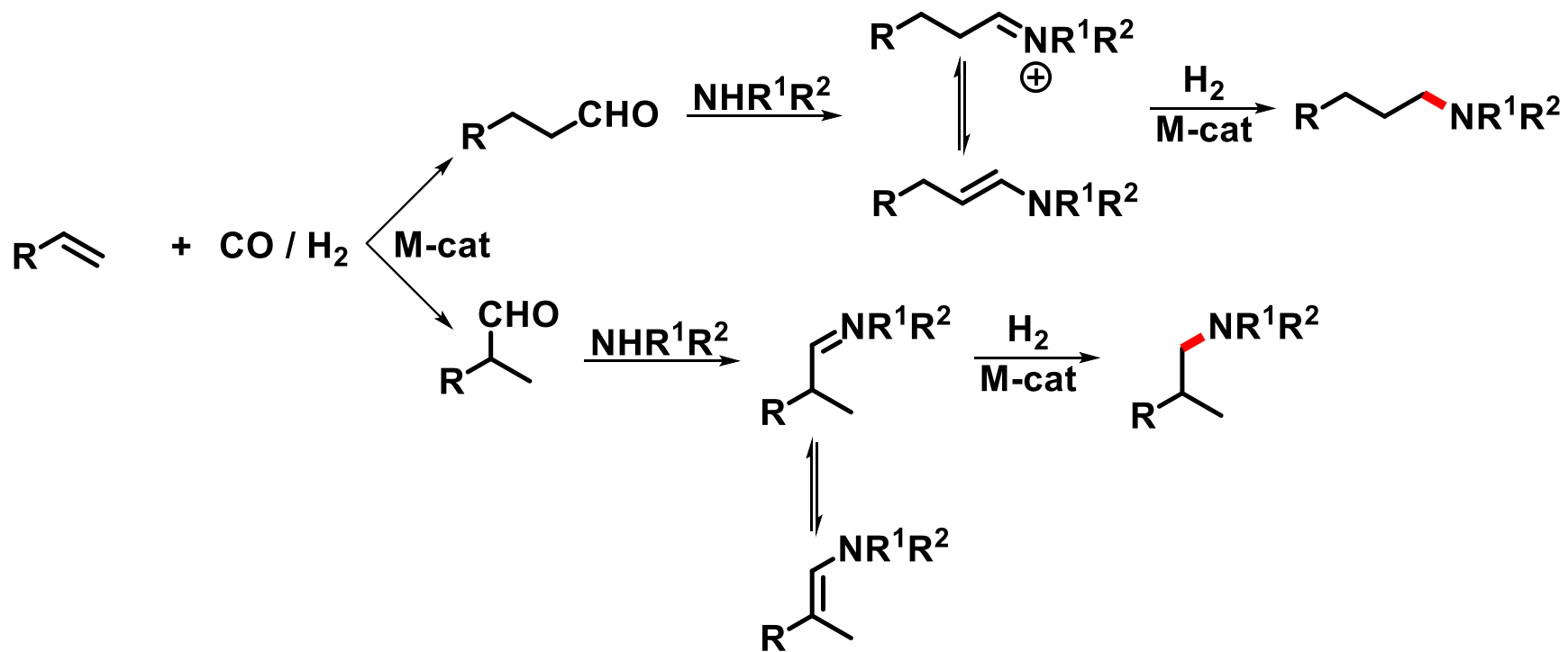
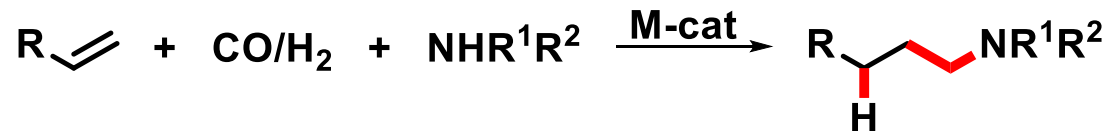
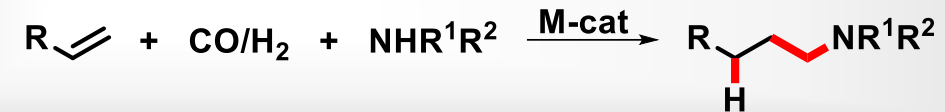
Experientia 1949, 5, 93



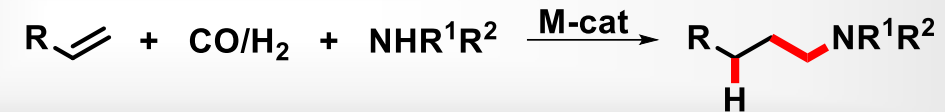
Reppe, W. Patent demand 839 800 (to I.G. Farben, May 5, 1943), then to BASF (July 8, 1949) 1943, allowed on May 26 1952

➤ Hydroaminomethylation

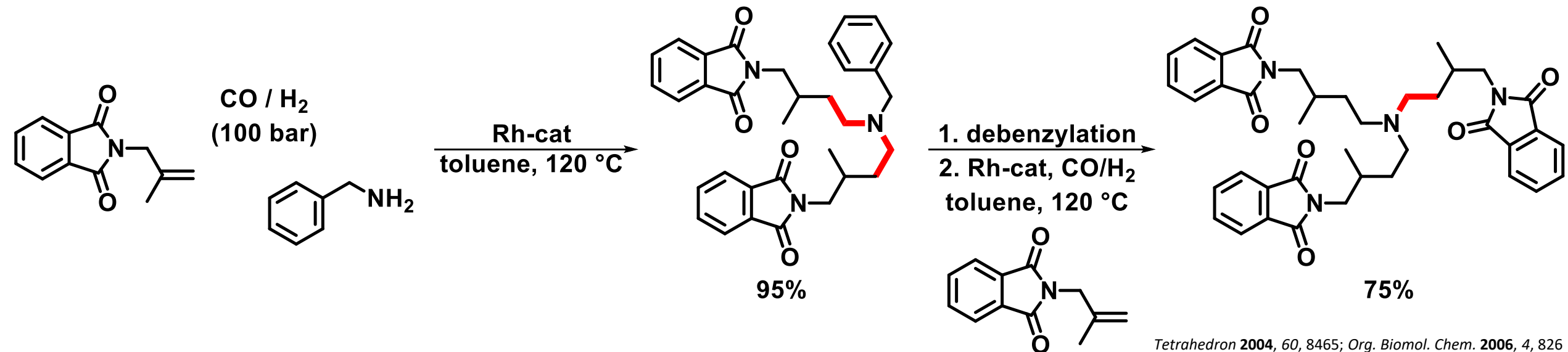
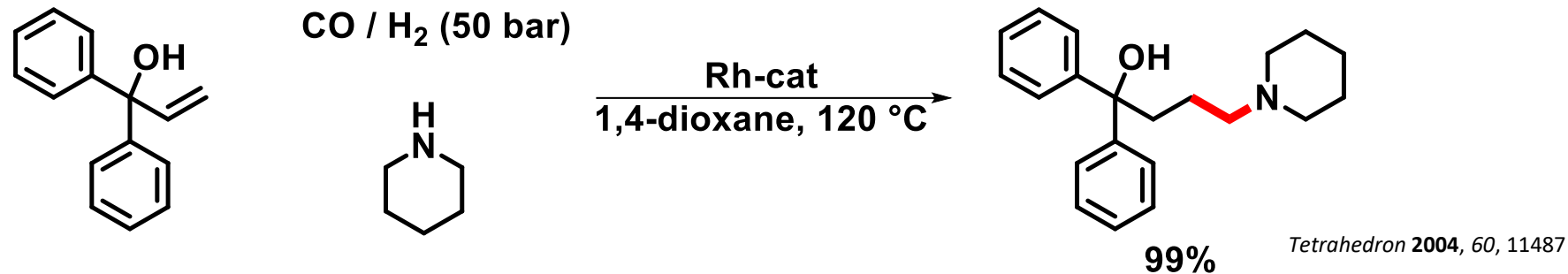
- Mechanistic consideration



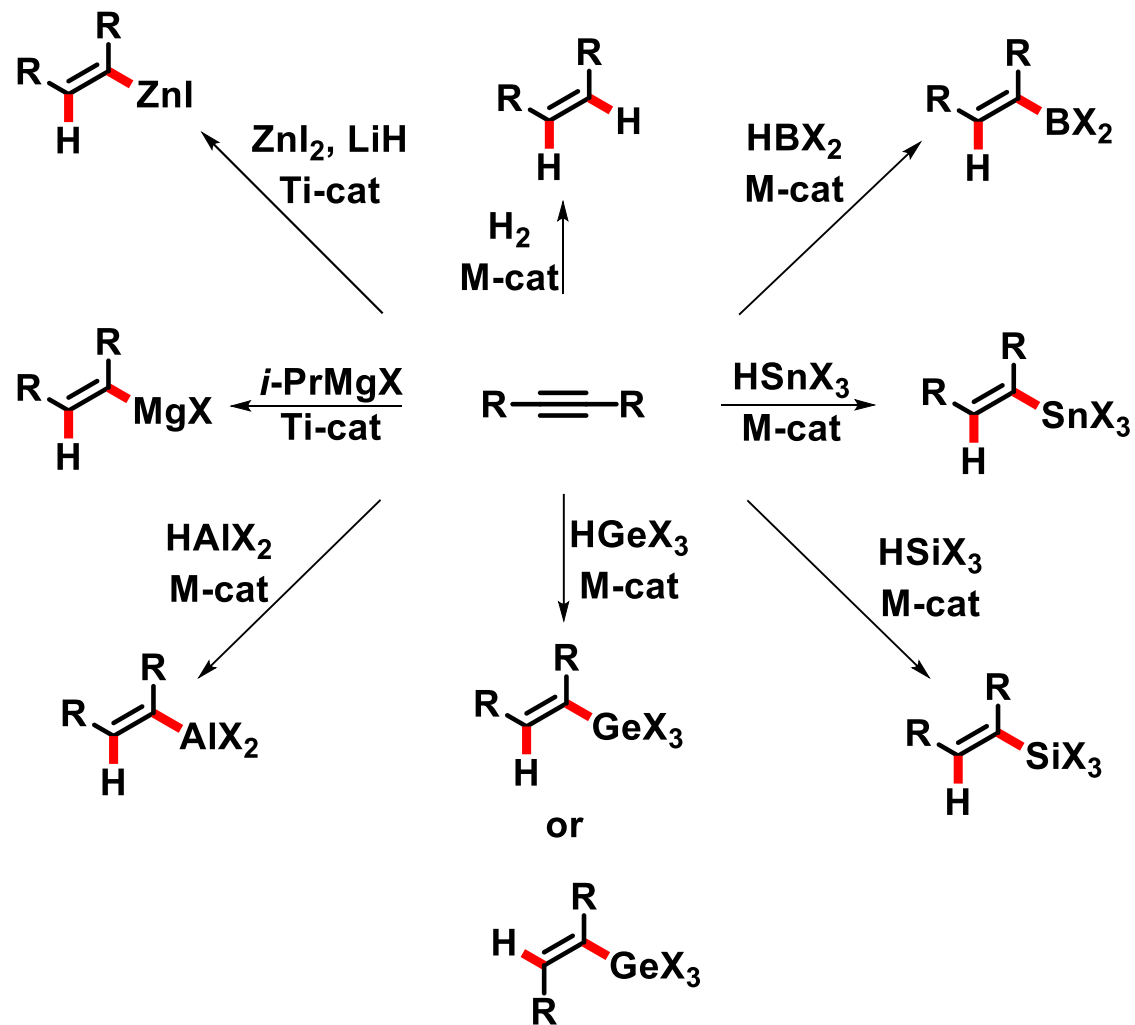
➤ Hydroaminomethylation



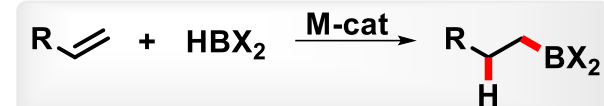
- Recent examples



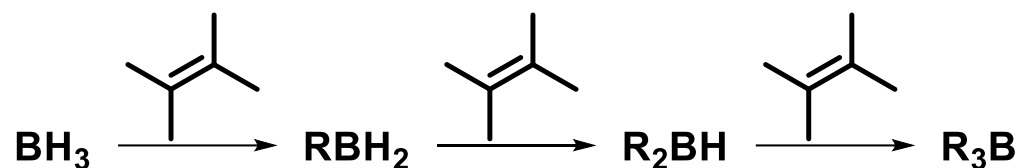
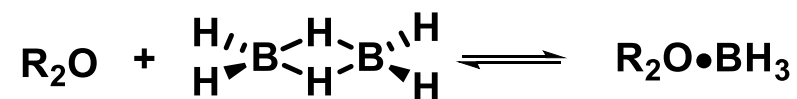
➤ Transition-metal-catalyzed hydrometallations



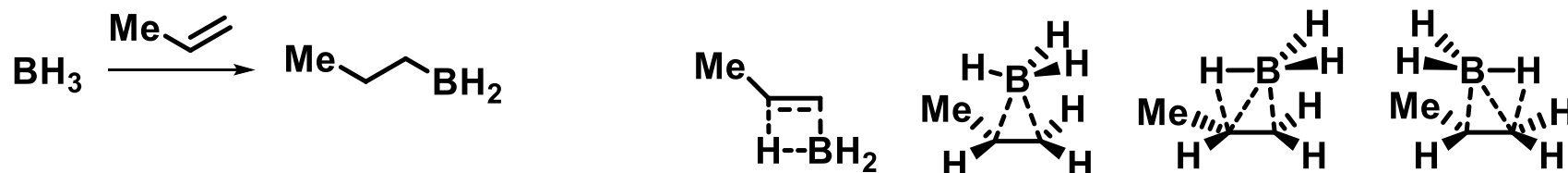
➤ Transition-metal-catalyzed hydrometallations



- Hydroboration



- Proposed transition states for the borane addition to 1-propene



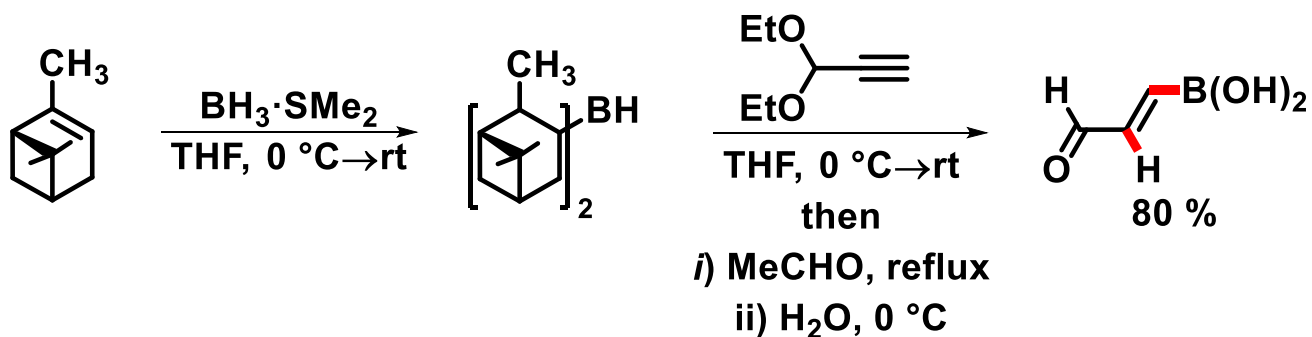
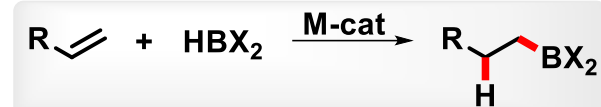
J. Am. Chem. Soc. **2009**, *131*, 3130

- Why transition-metal-catalyzed hydroboration?

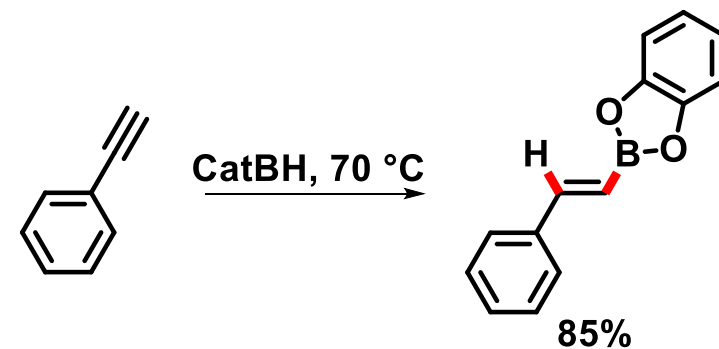
➤ Transition-metal-catalyzed hydrometallations

- Hydroboration

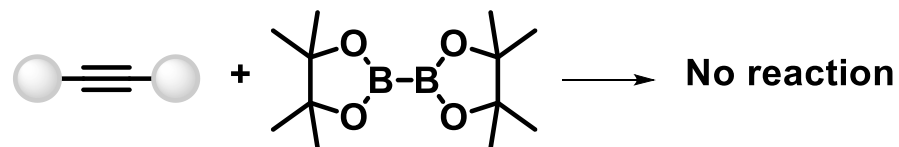
- Why transition-metal-catalyzed hydroboration?



Chem. Eur. J. 2003, 9, 466

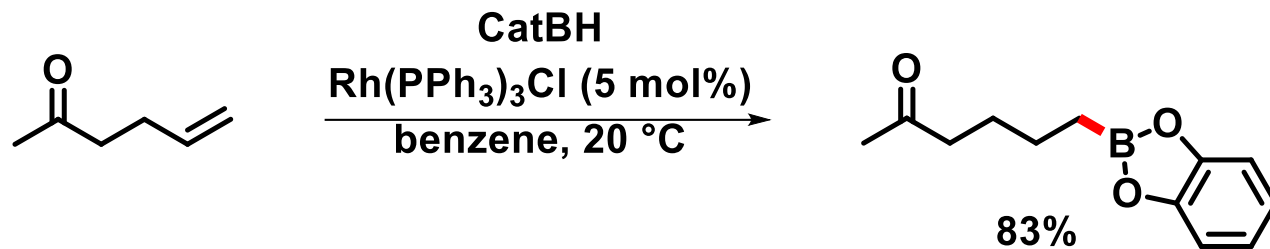
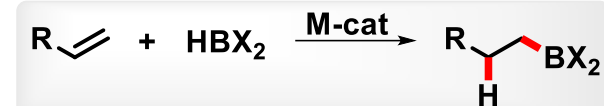


J. Org. Chem. 1975, 97, 5249

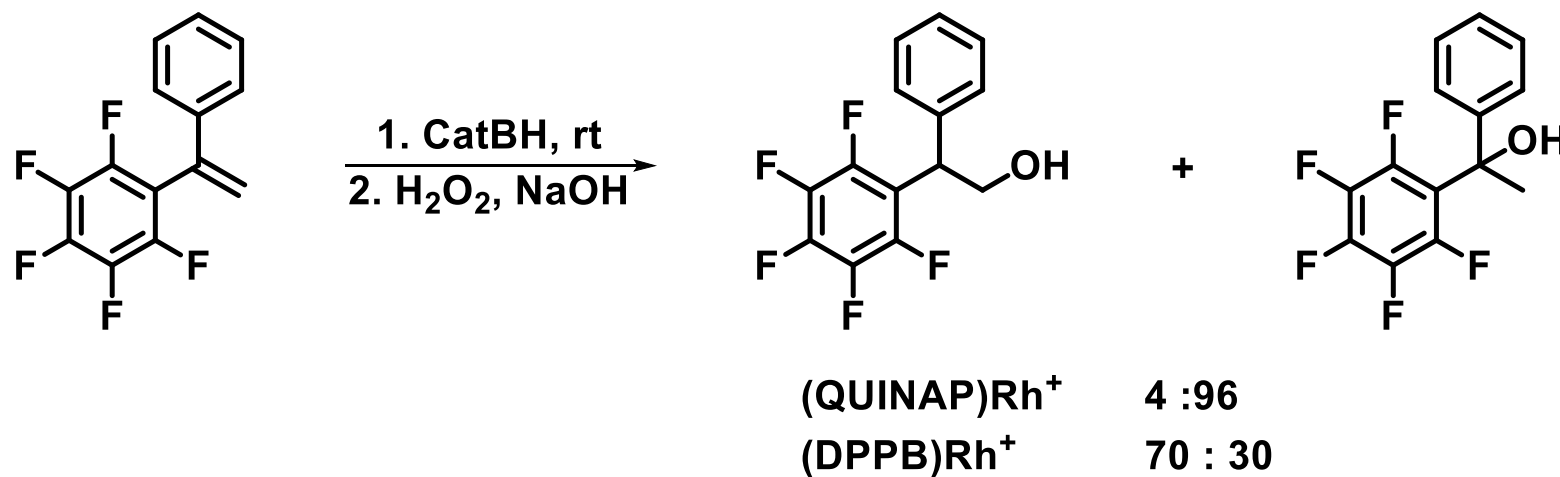


➤ Transition-metal-catalyzed hydrometallations

- Transition-metal-catalyzed hydroboration (Cu, Ni, Co, Fe ...)

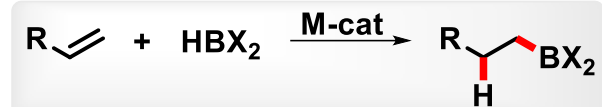


Angew.Chem. Int. Ed. 1985, 24, 878

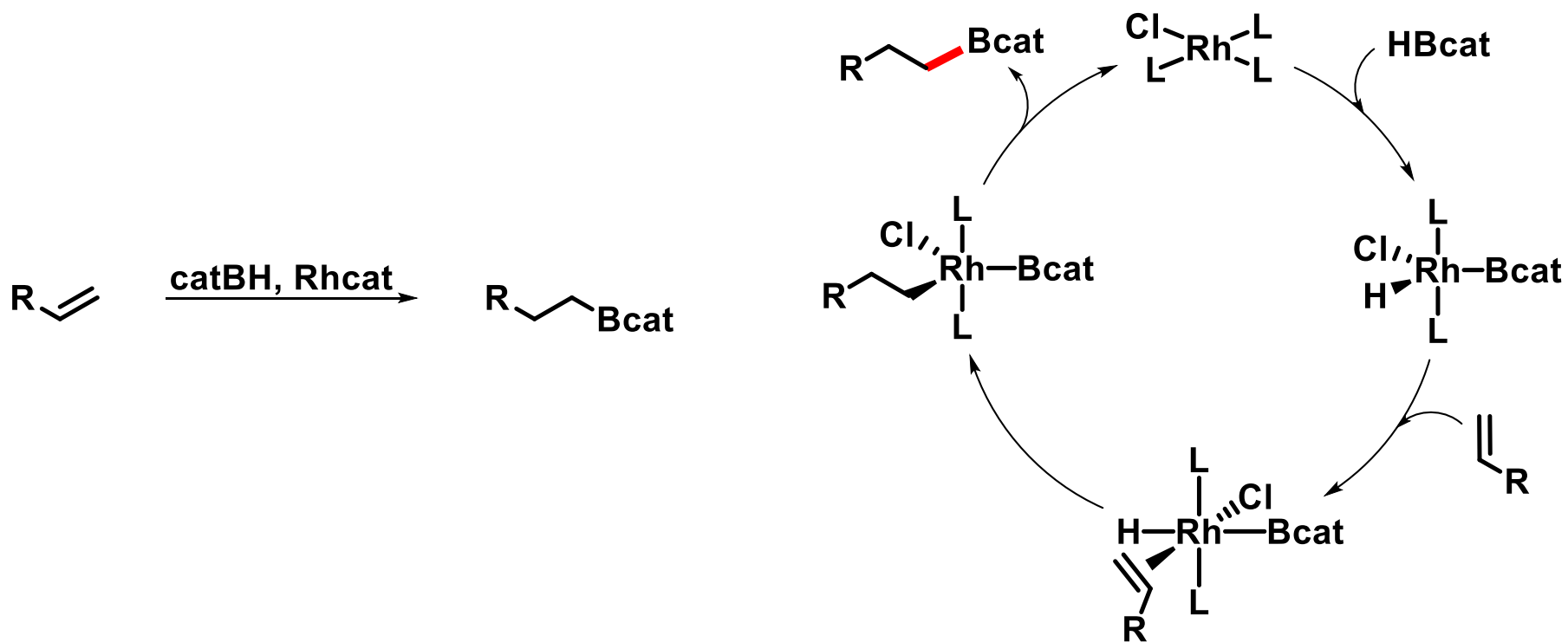


Chem. Commun. 2005, 5284

➤ Transition-metal-catalyzed hydrometallations

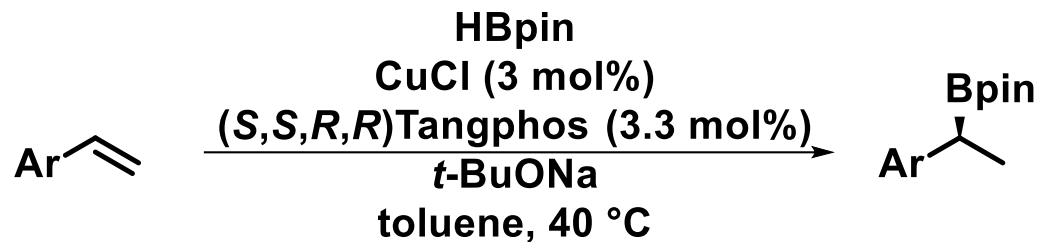
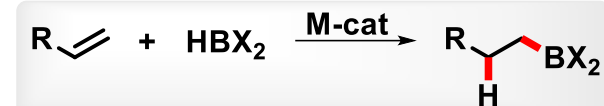


- Mechanism of rhodium-catalyzed hydroboration

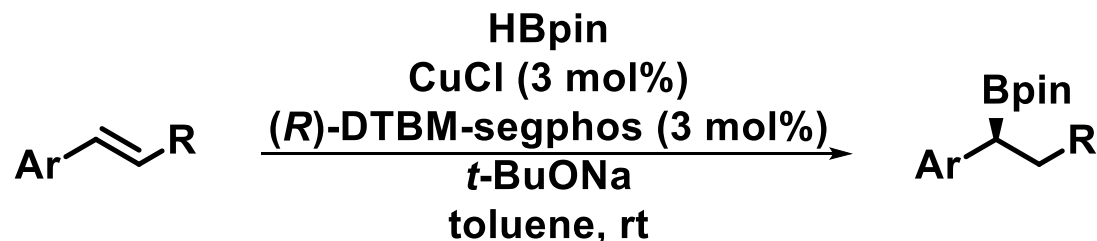


➤ Transition-metal-catalyzed hydrometallations

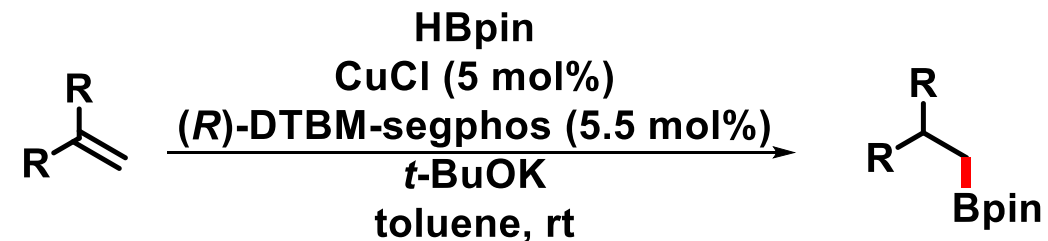
- Transition-metal-catalyzed hydroboration (Cu, Ni, Co, Fe ...)



Angew. Chem. Int. Ed. **2009**, *48*, 6062



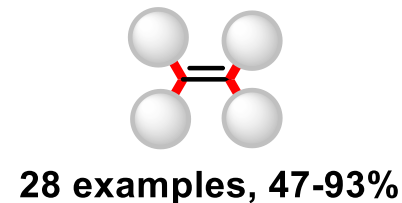
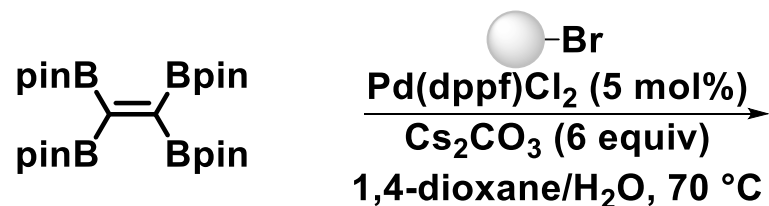
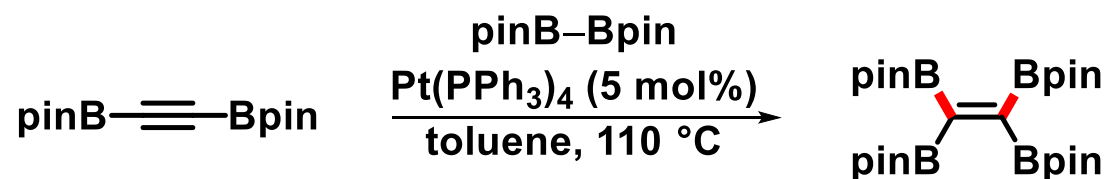
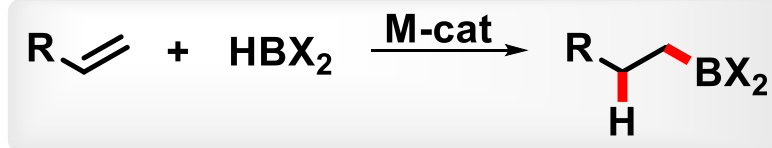
Angew. Chem. Int. Ed. **2013**, *52*, 3989; *Chem. Asian J.* **2011**, *6*, 1967



J. Am. Chem. Soc. **2017**, *139*, 13660

➤ Transition-metal-catalyzed hydrometallations

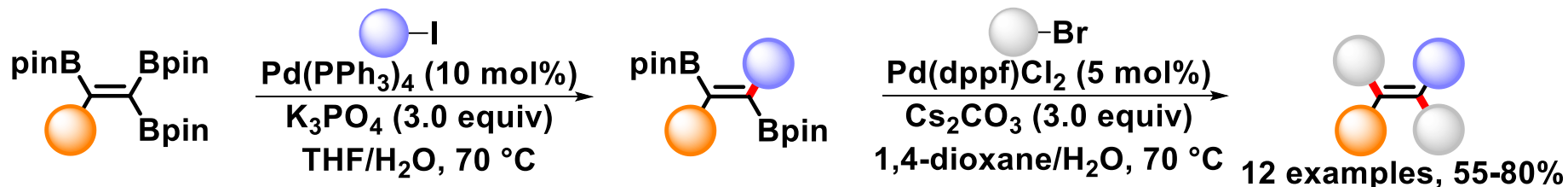
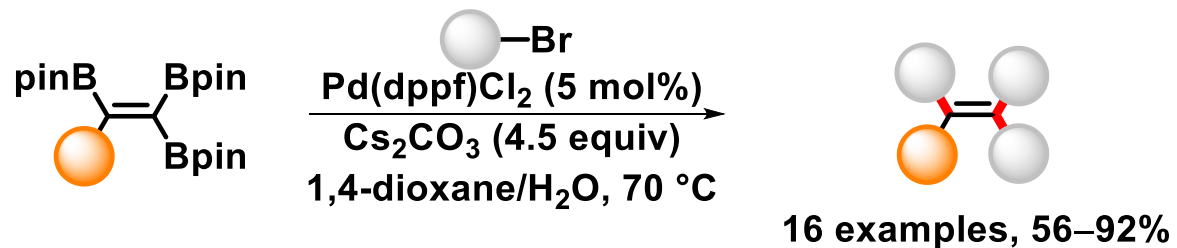
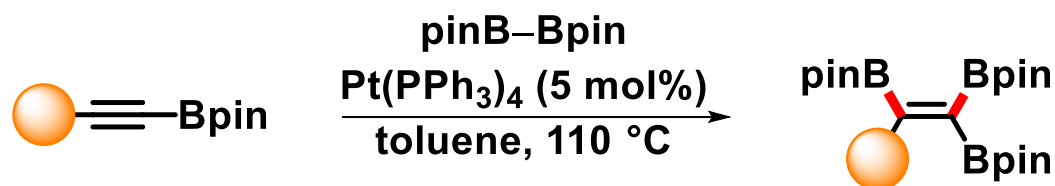
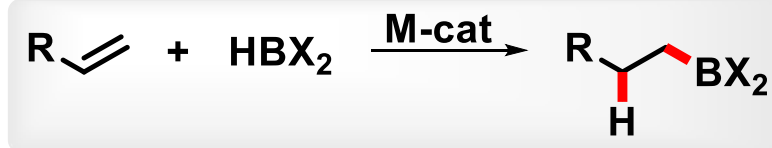
- Transition-metal-catalyzed hydroboration (Cu, Ni, Co, Fe ...)
- Lego-based approach to tetrasubstituted alkenes



Angew. Chem. Int. Ed. 2020, 59, 20090

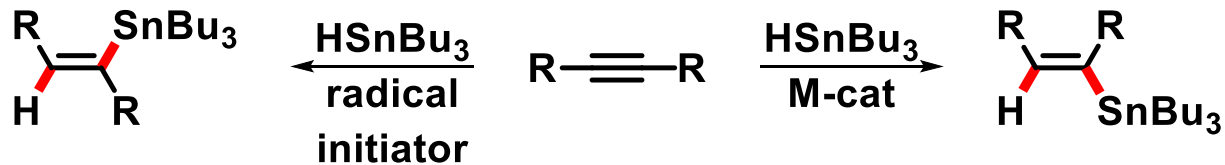
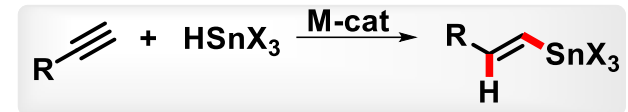
➤ Transition-metal-catalyzed hydrometallations

- Transition-metal-catalyzed hydroboration (Cu, Ni, Co, Fe ...)
- Lego-based approach to tetrasubstituted alkenes

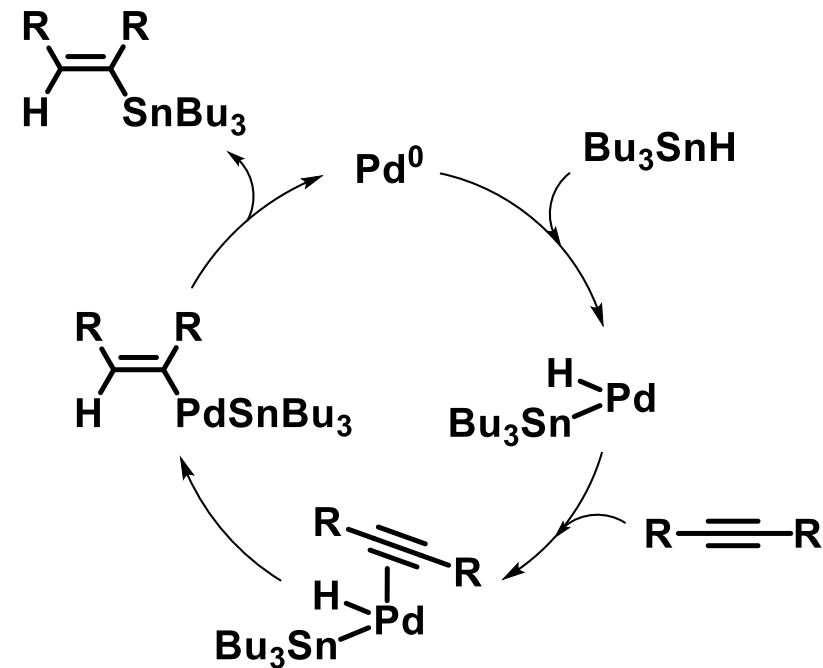


➤ Transition-metal-catalyzed hydrometallations

- Transition-metal-catalyzed hydrostannylation

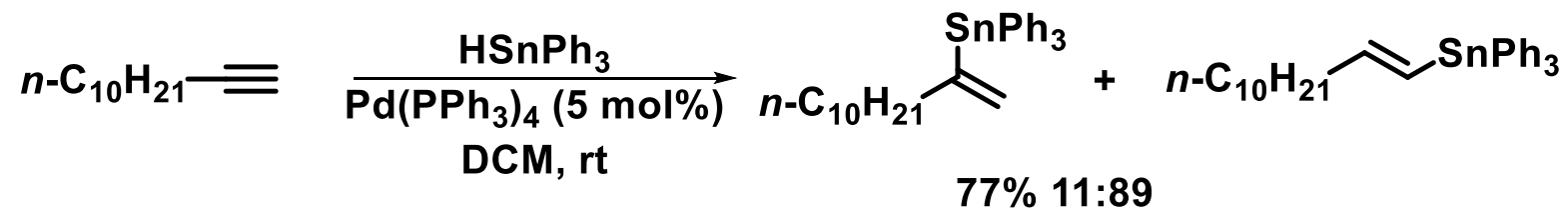
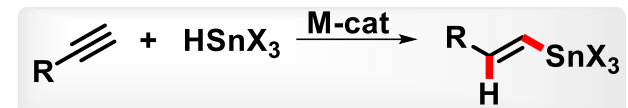


○ Mechanism of palladium-catalyzed hydrostannylation

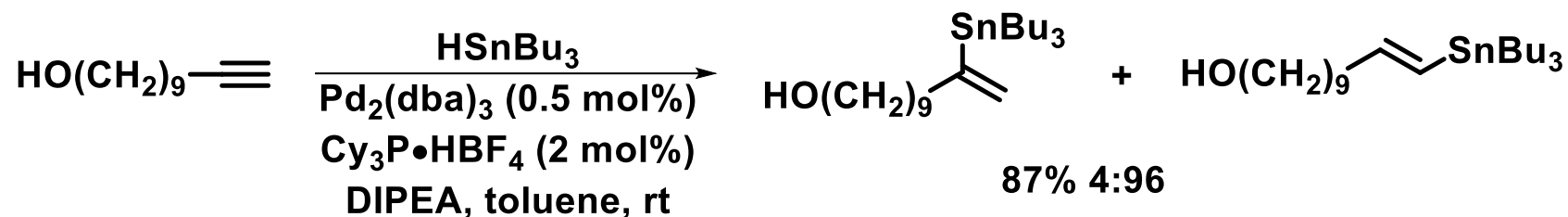


➤ Transition-metal-catalyzed hydrometallations

- Transition-metal-catalyzed hydrostannylation

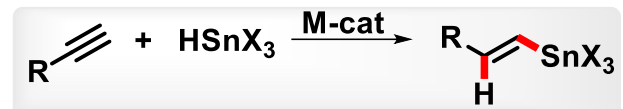


Bull. Chem. Soc. Jpn. **1987**, *60*, 3468



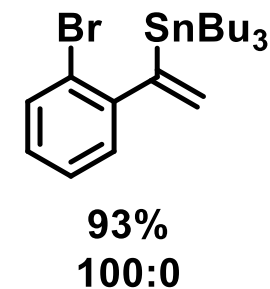
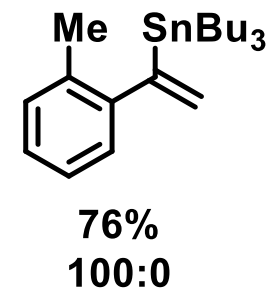
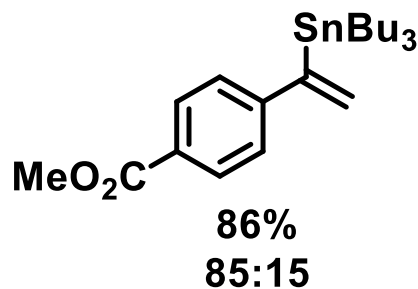
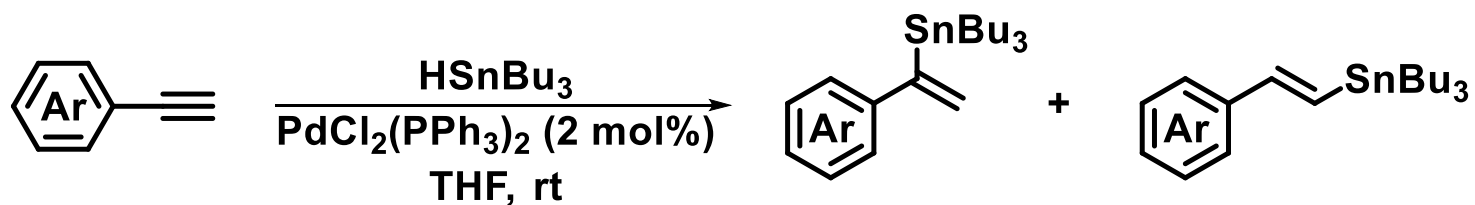
Org. Lett. **2008**, *10*, 861

➤ Transition-metal-catalyzed hydrometallations



- Transition-metal-catalyzed hydrostannylation

- Hydrostannylation of *ortho*-substituted arylacetylenes

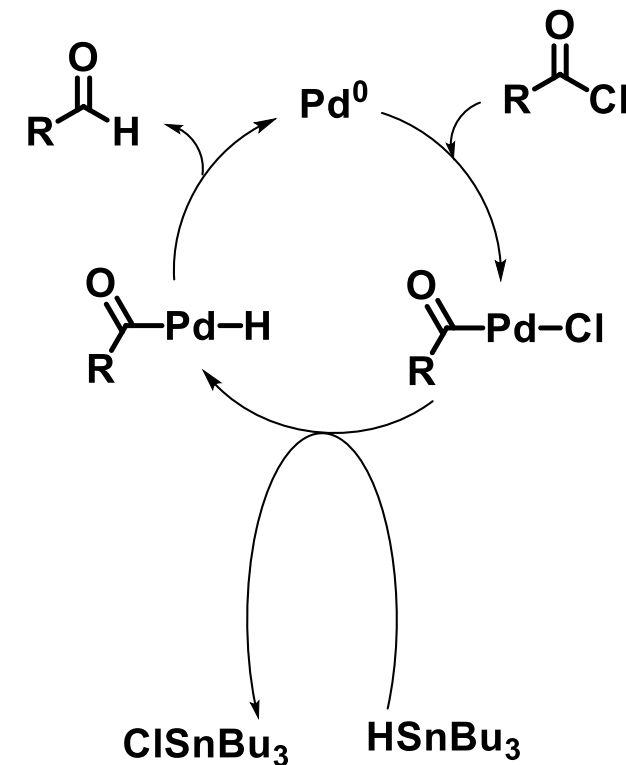
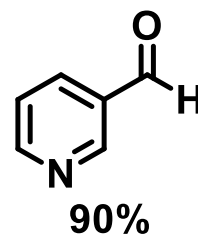
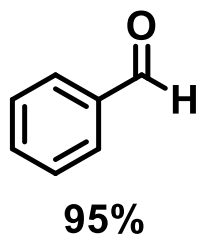
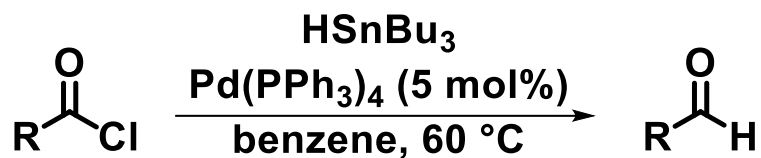


J. Org. Chem. **2009**, *74*, 1337; *Synlett* **1999**, 246

➤ Transition-metal-catalyzed hydrometallations

- Transition-metal-catalyzed hydrostannylation
 - Other applications of R_3SnH – reduction of acyl chlorides

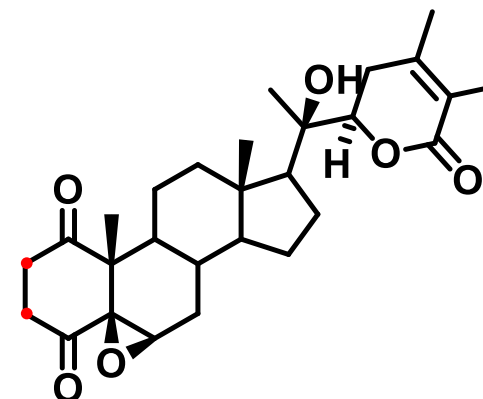
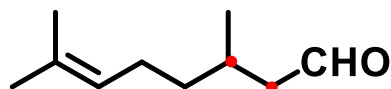
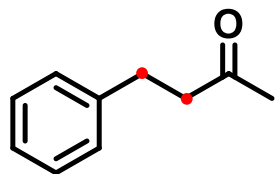
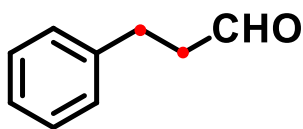
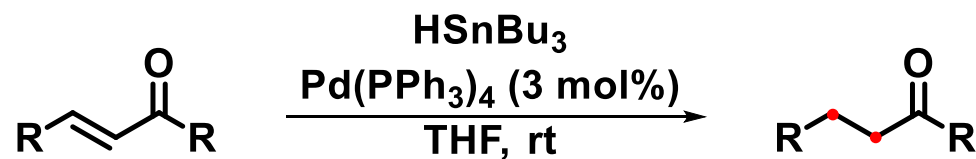
✓ Proposed mechanism



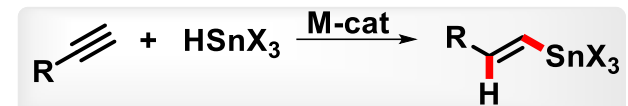
J. Org. Chem. **1981**, *46*, 4439

➤ Transition-metal-catalyzed hydrometallations

- Transition-metal-catalyzed hydrostannylation
 - Other applications of R_3SnH – reduction of α,β -unsaturated ketones

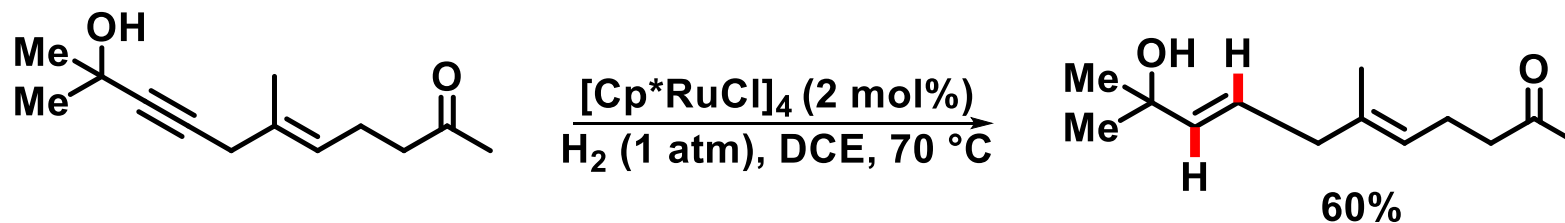
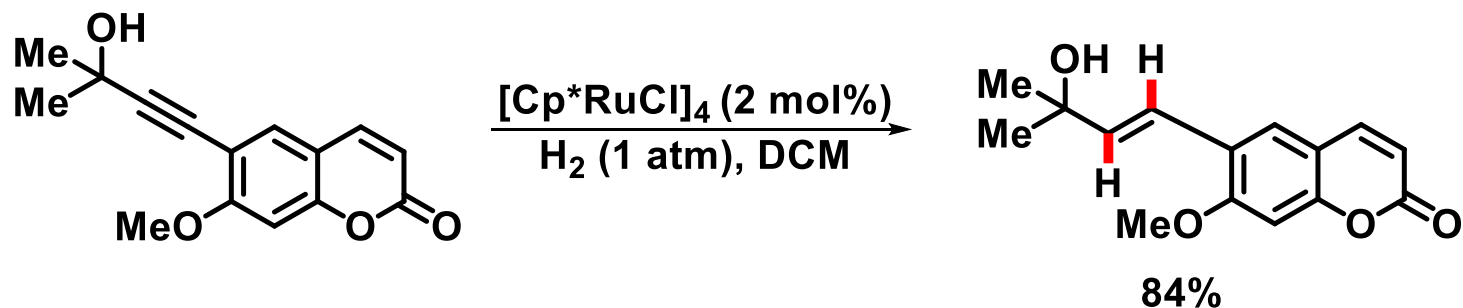


➤ *Trans* hydrogenation and *trans* hydrometallation

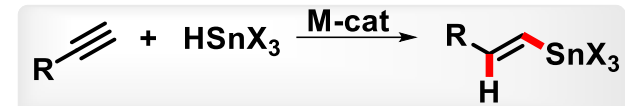


- So far, all reactions of hydrido complexes have been syn additions

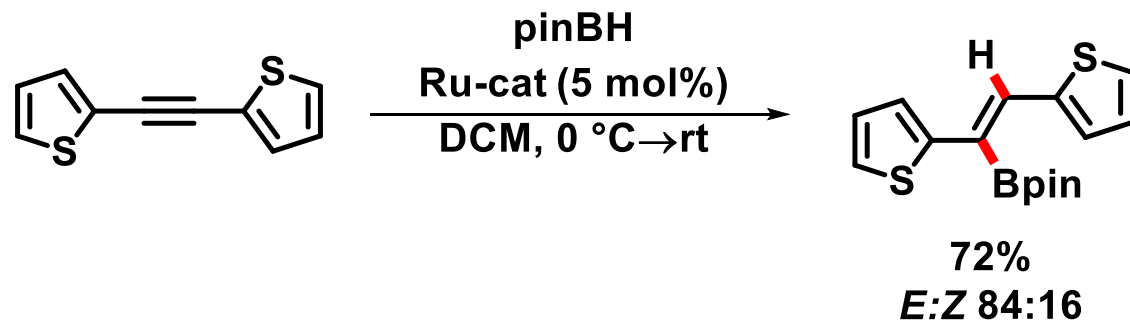
○ Selected examples



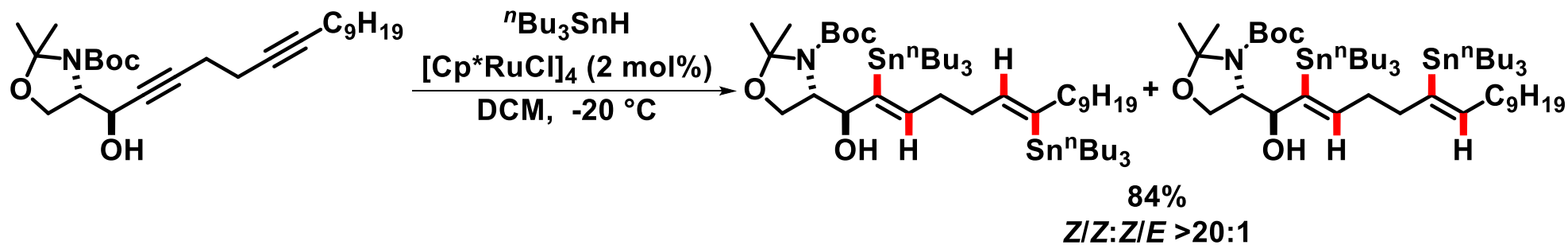
➤ *Trans* hydrogenation and *trans* hydrometallation



- Selected examples



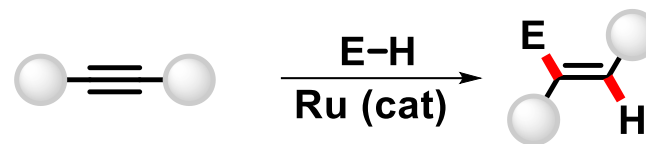
Angew. Chem. Int. Ed. **2013**, *52*, 14050



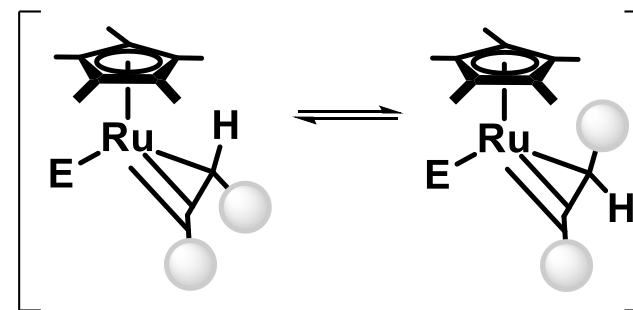
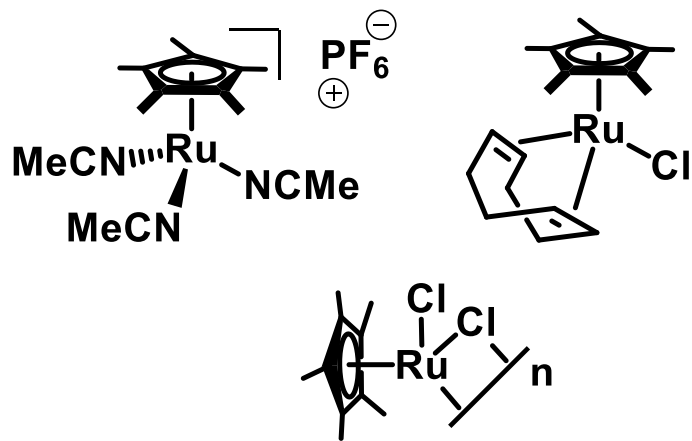
Chem. - Eur. J. **2018**, *24*, 9667

➤ *Trans* hydrogenation and *trans* hydrometallation

- Different mechanism



Ru precatalyst



➤ Suzuki Reaction (Suzuki–Miyaura reaction)

- Synthesis of boronic acids or boronic acid esters
 - Anti hydroboration (suitable for vinylboronic acids)

