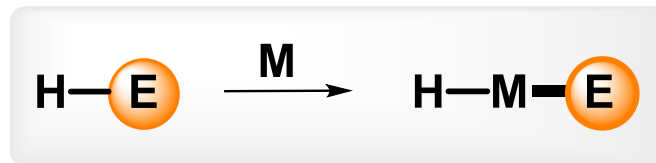


## ➤ C–H activation

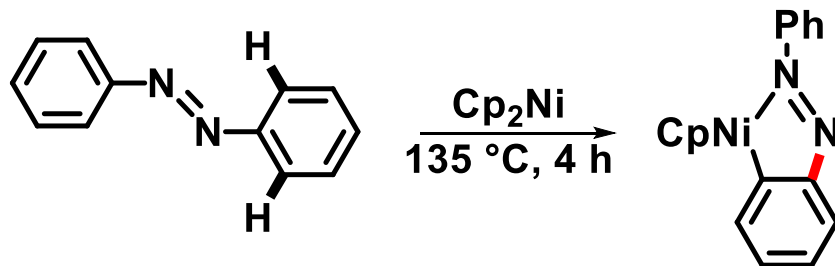


- Transition-metal-free C–H activation:

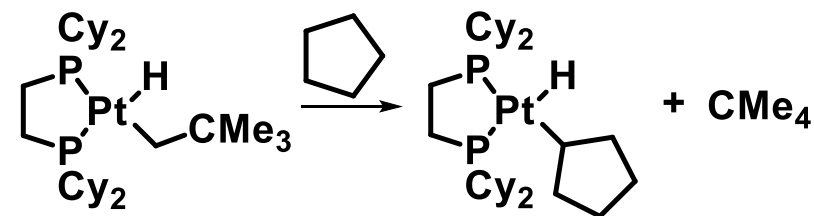


- Transition metal mediated C–H activation

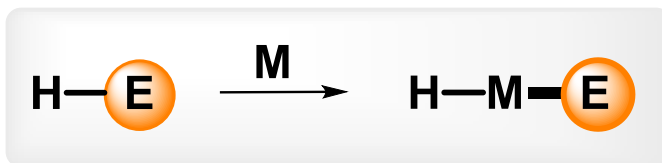
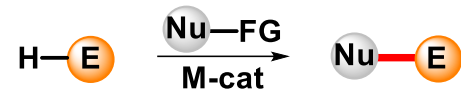
- Intramolecular C–H activation (cyclometallation):



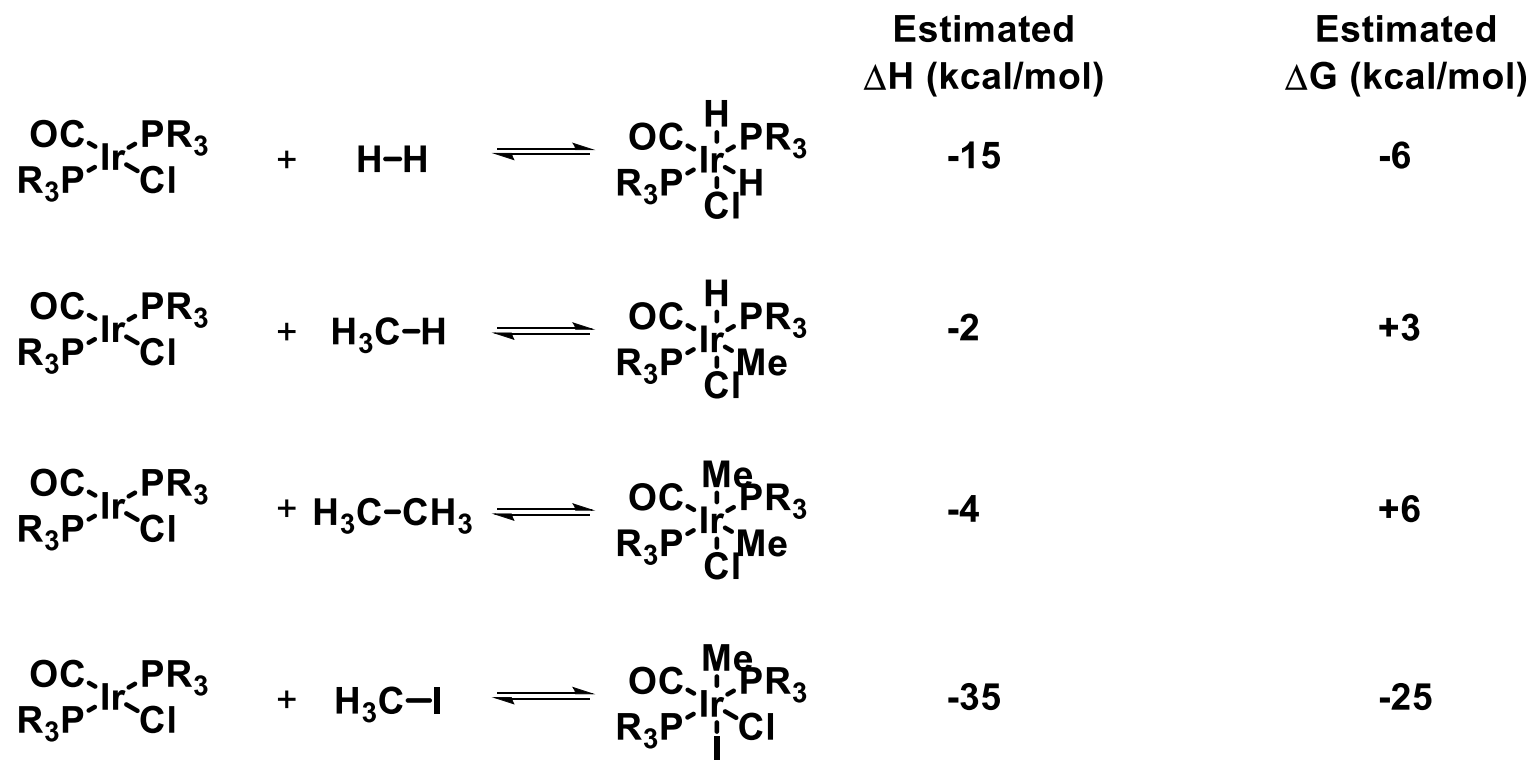
- Intermolecular C–H activation:



## ➤ C–H activation



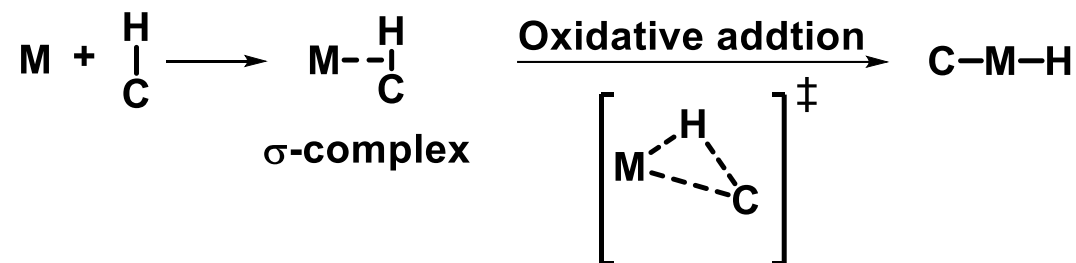
- Thermodynamics of oxidative addition for Vaska's complex



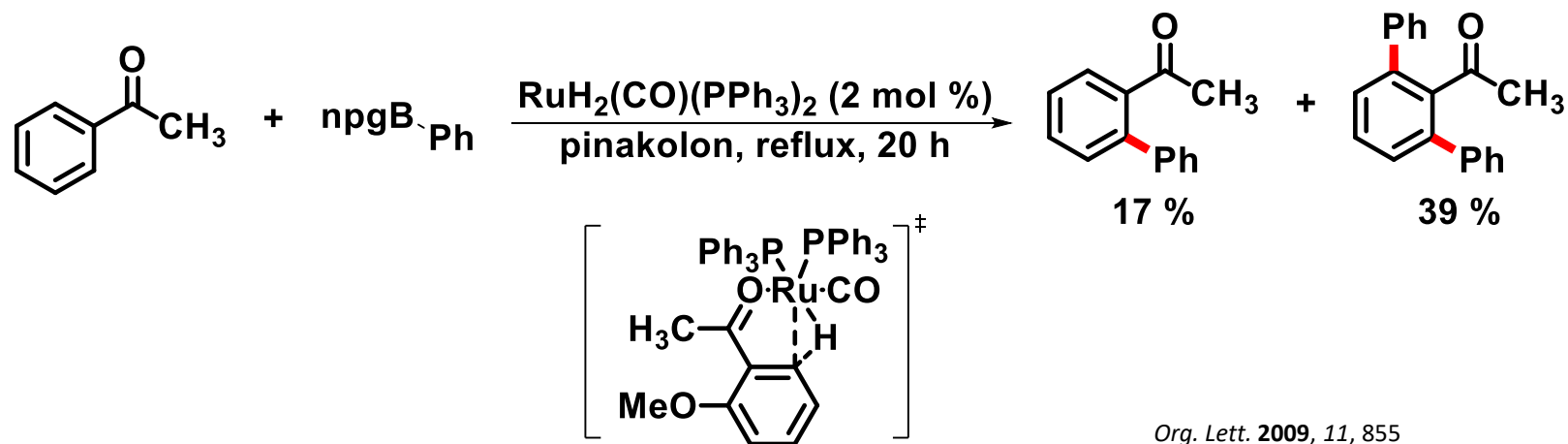
## ➤ C–H activation

- Mechanism of C–H activation

- Concerted mechanism

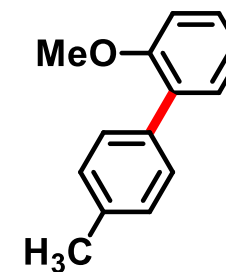
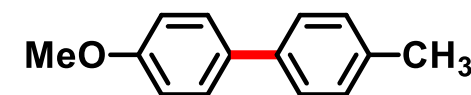
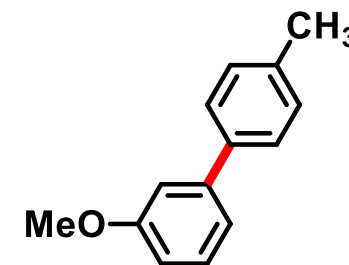
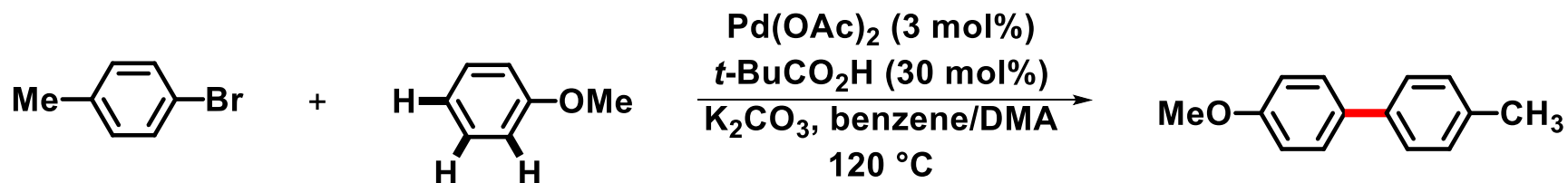
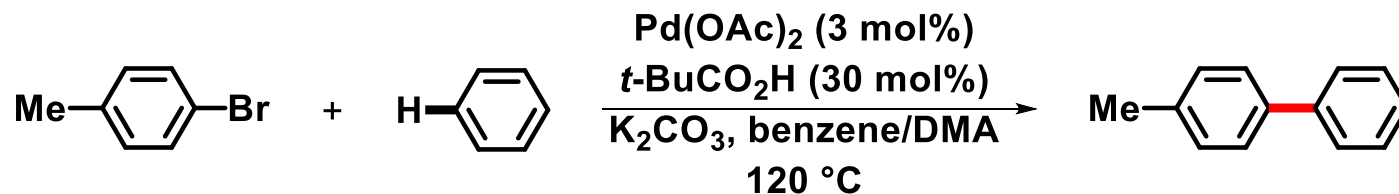
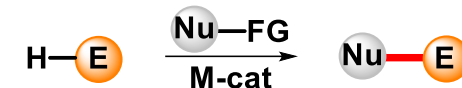


- Selected example



*Org. Lett.* 2009, 11, 855

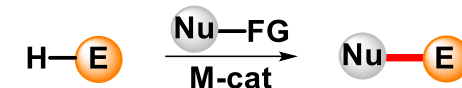
## ➤ C–H activation



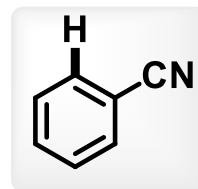
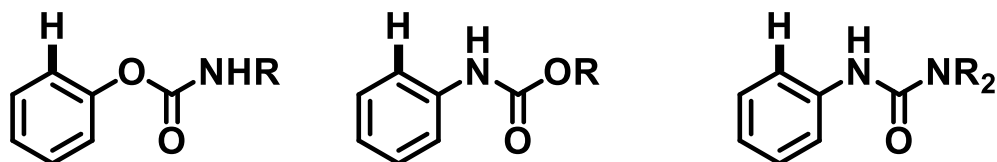
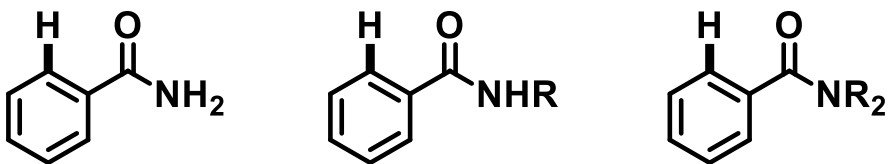
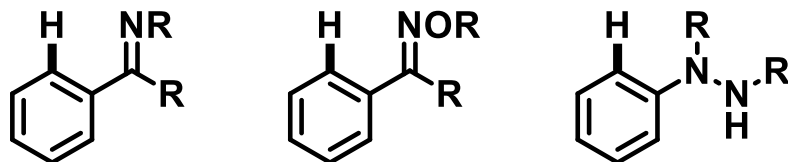
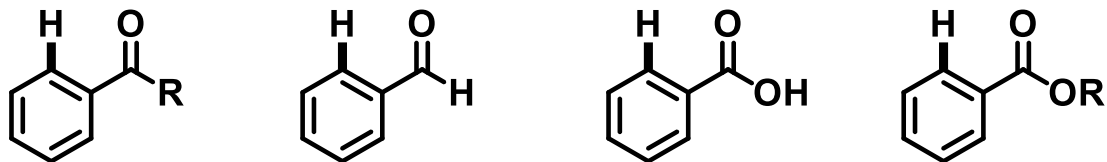
*o:m:p* = 22:53:25

## ➤ C–H activation

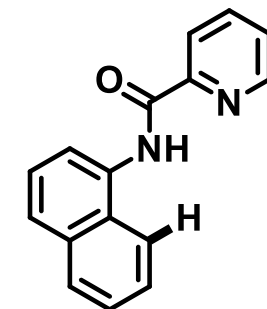
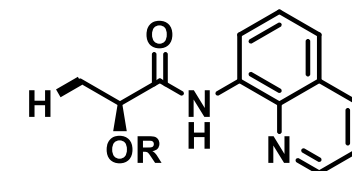
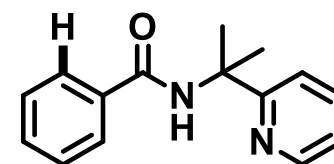
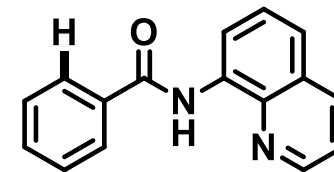
- Directing groups



### ○ Monodentate directing groups

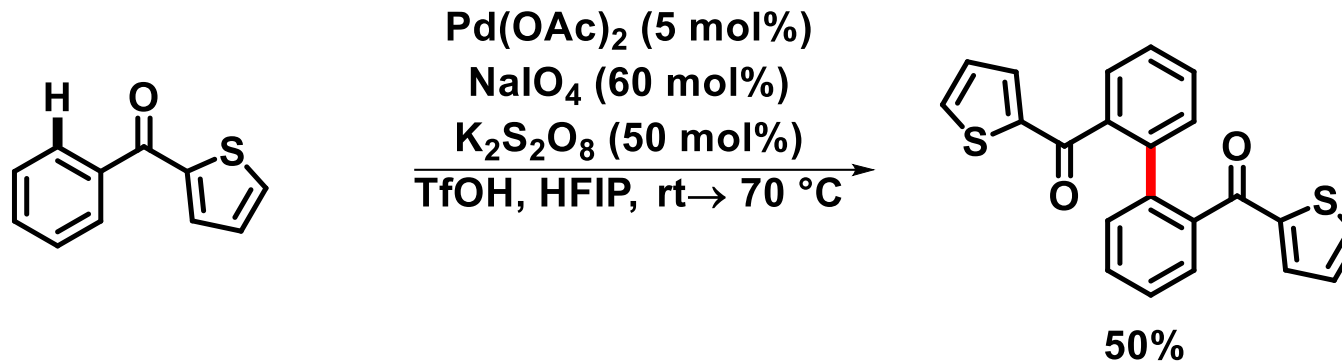
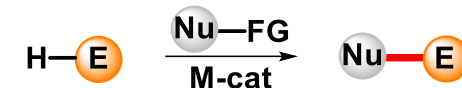


### ○ Bidentate directing groups

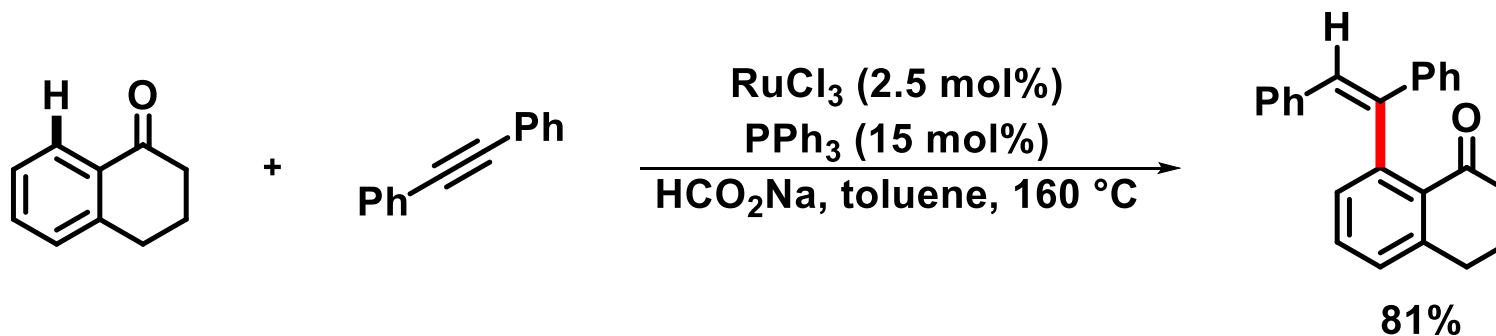


## ➤ C–H activation

- *Ortho* directing groups



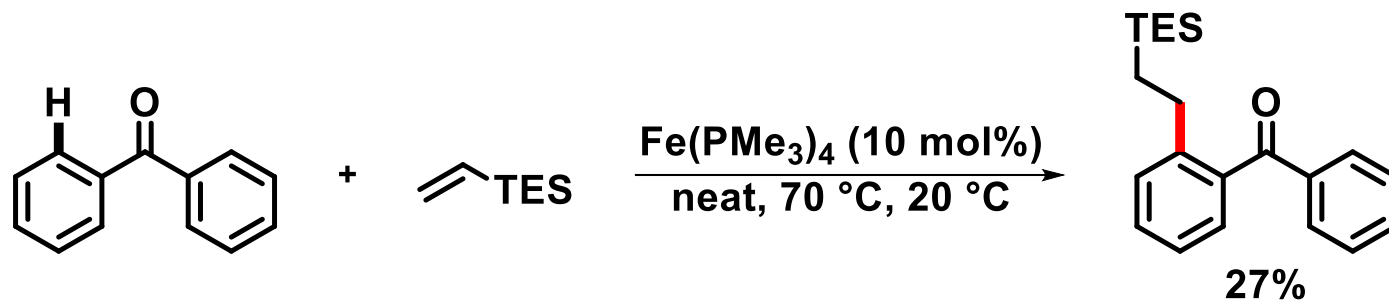
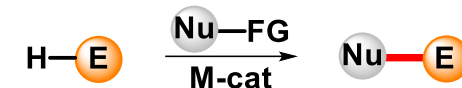
*Org. Lett.* 2015, 17, 4456



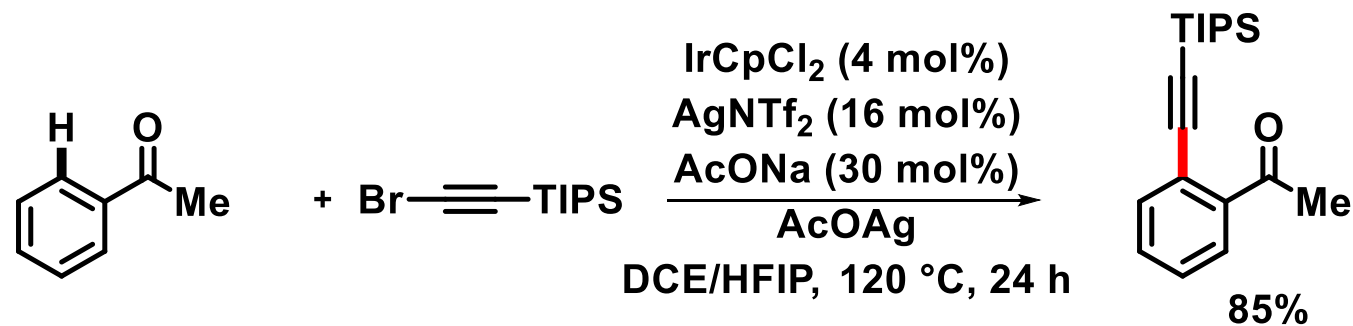
*Chem. Commun.* 2016, 52, 9715

## ➤ C–H activation

- *Ortho* directing groups



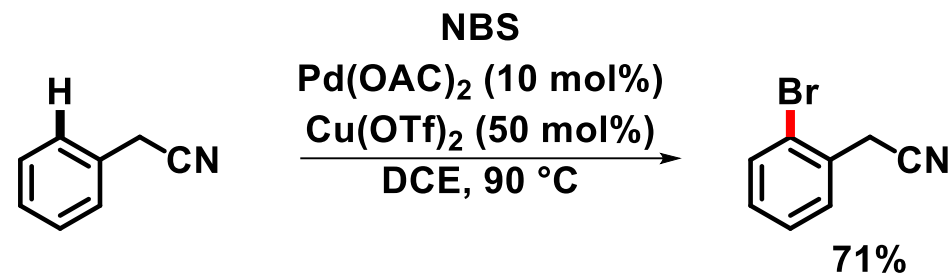
*J. Am. Chem. Soc.* **2017**, *139*, 14849



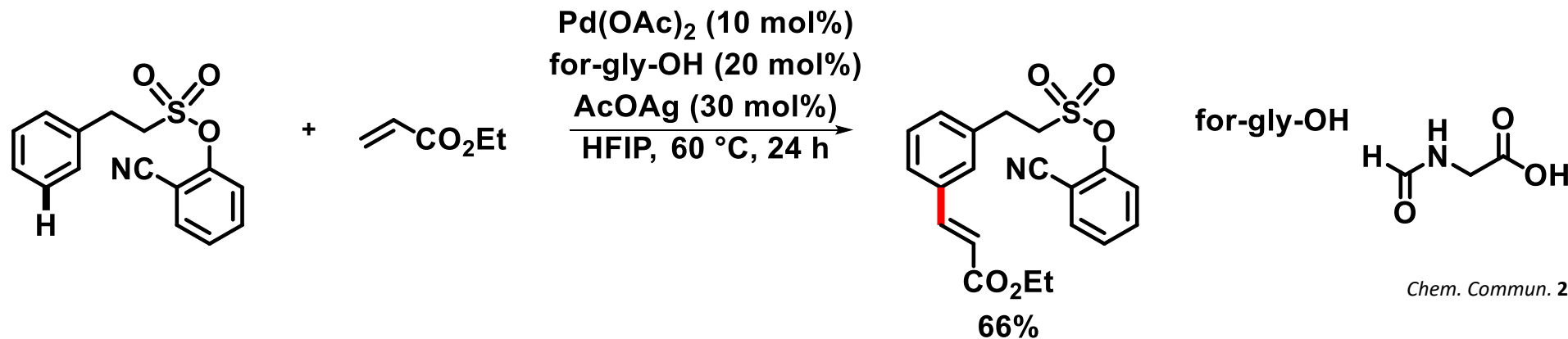
*J. Org. Chem.* **2017**, *82*, 13003

## ➤ C–H activation

- *Meta*-directing groups



*J. Org. Chem.* **2017**, *82*, 1114

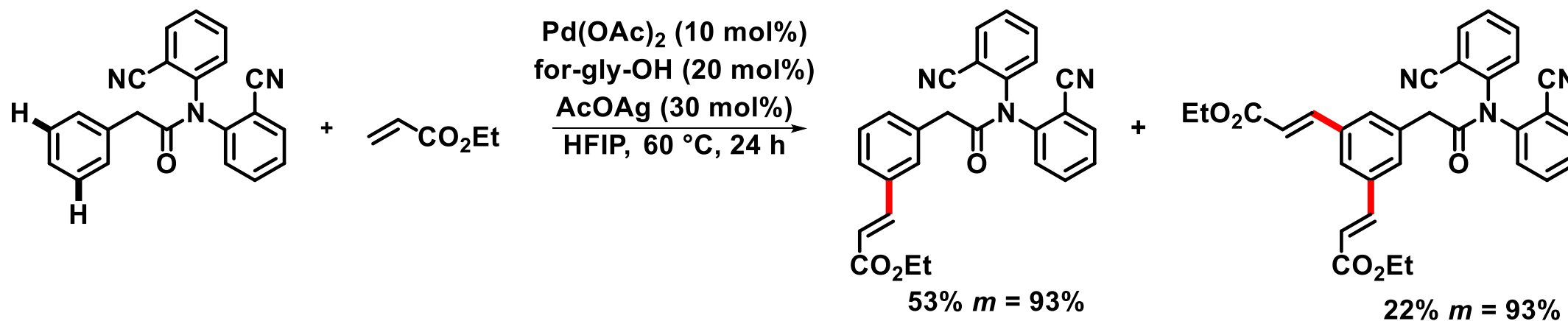
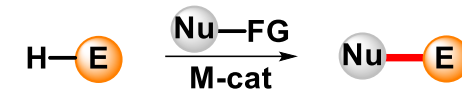


*Chem. Commun.* **2016**, *52*, 13916



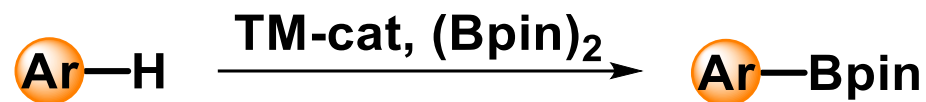
## ➤ C–H activation

- *Meta*-directing groups



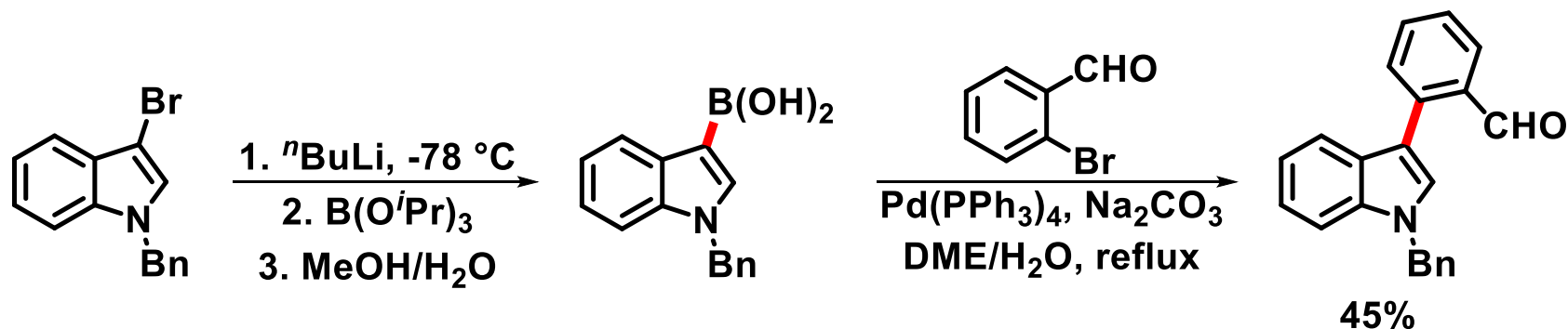
*Angew. Chem., Int. Ed.* 2015, 54, 888

## ➤ C–H activation



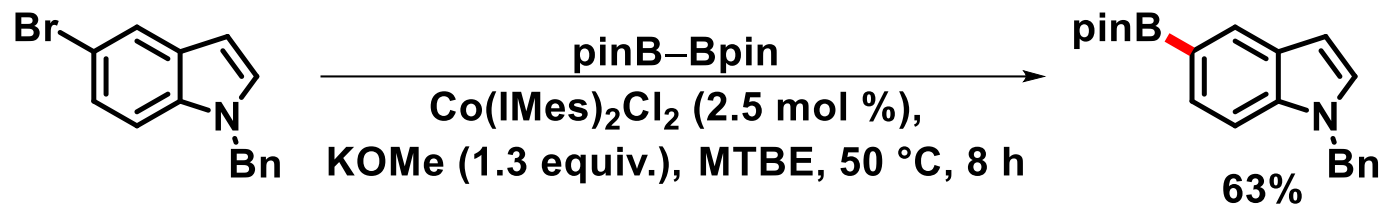
- C–H borylation

### ○ Traditional transition-metal-free borylation



*Tetrahedron* 2008, 64, 9033

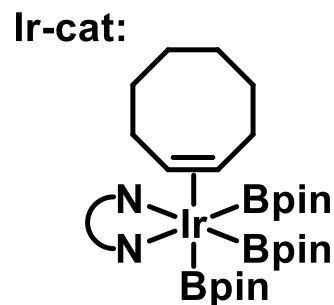
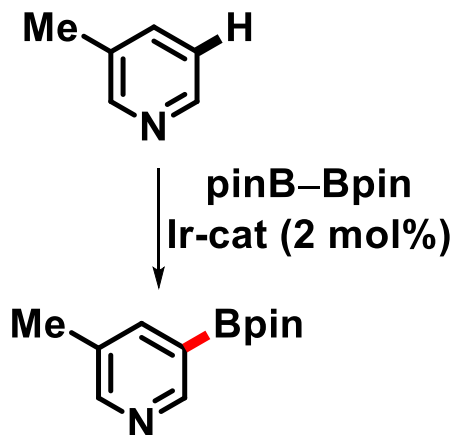
### ○ Miyaura borylation



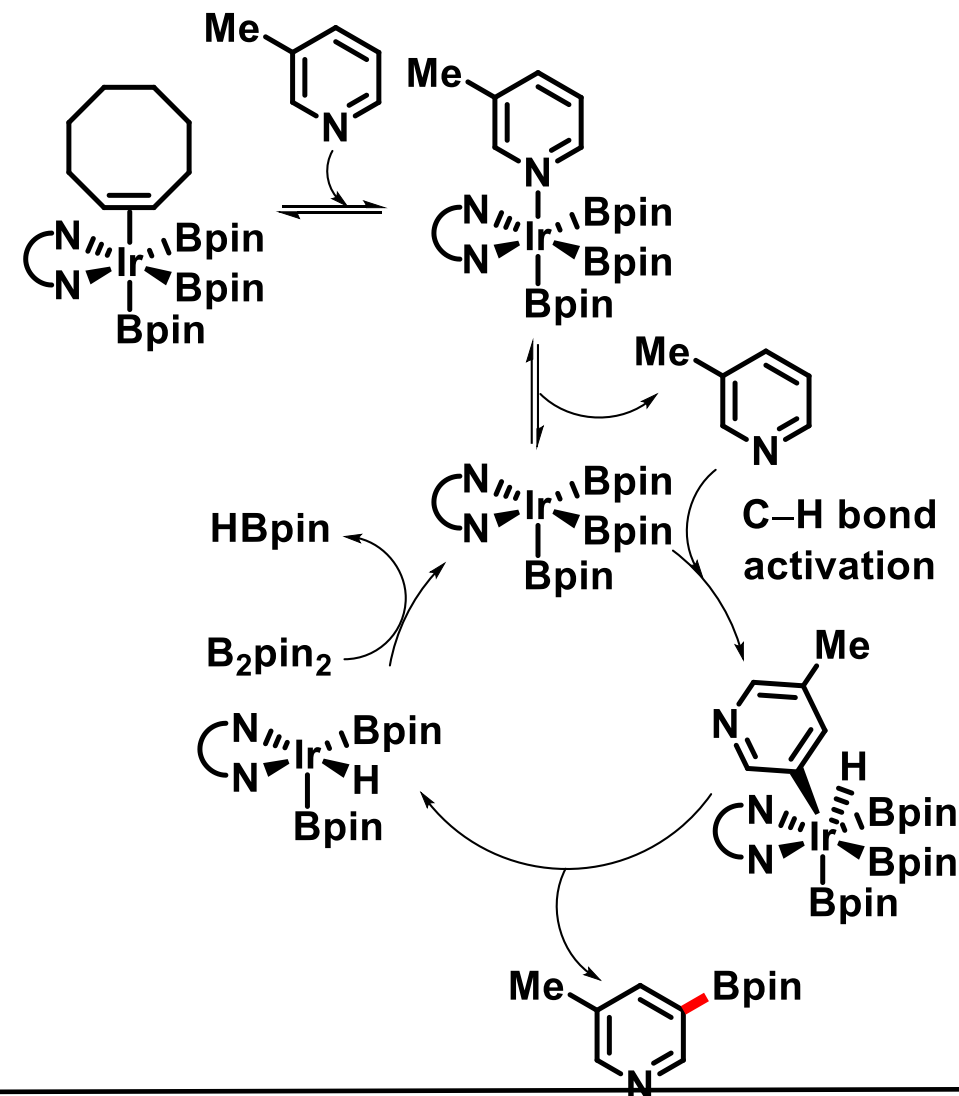
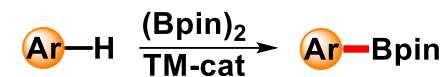
*ACS Catal.* 2018, 8, 4049

## ➤ C–H activation

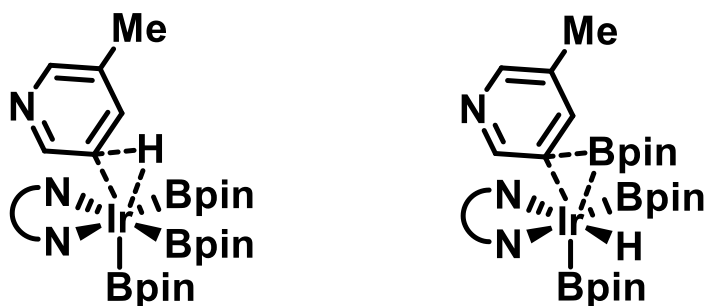
- C–H borylation



## ○ Proposed mechanism

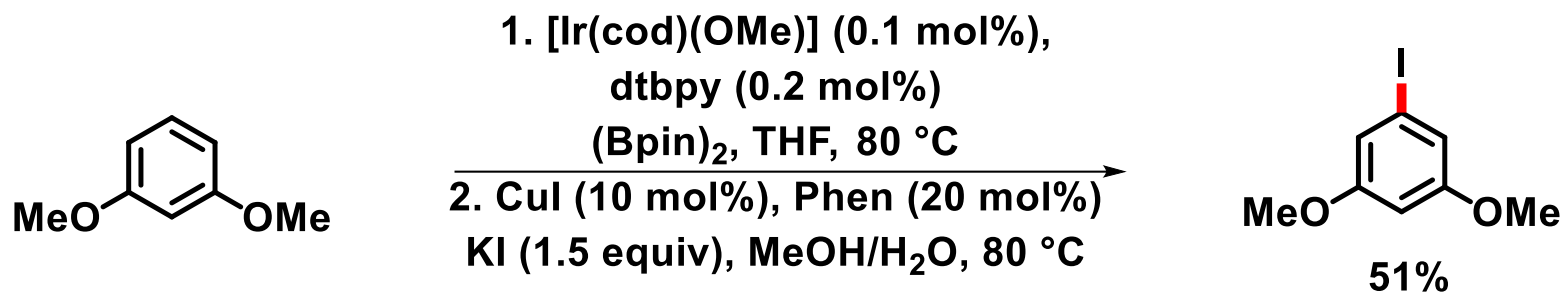
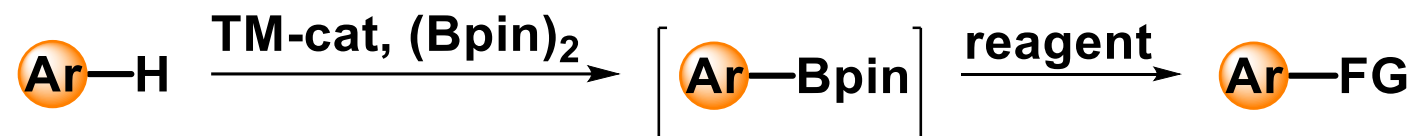
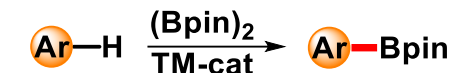


## ○ Calculated transition states for C–H activation



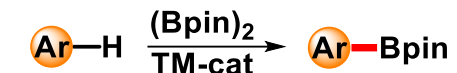
## ➤ C–H activation

- C–H borylation – Selected applications

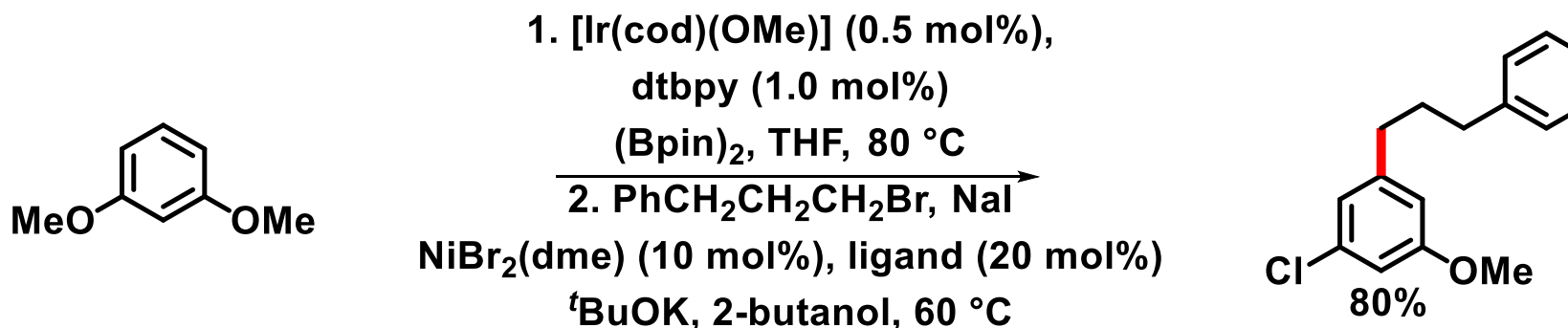
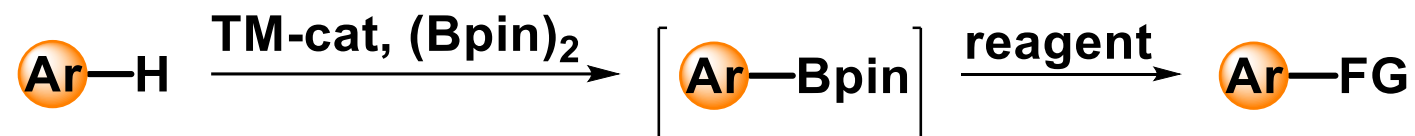


*Org. Lett.* 2013, 15, 140

## ➤ C–H activation

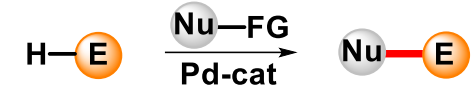


- C–H borylation – Selected applications

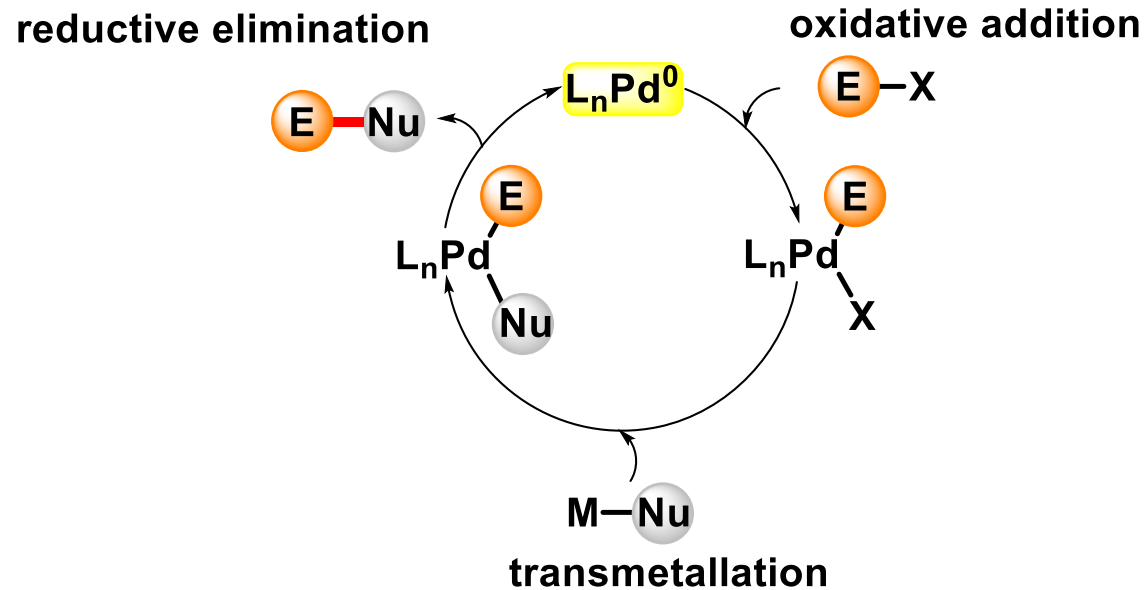


## ➤ C–H activation

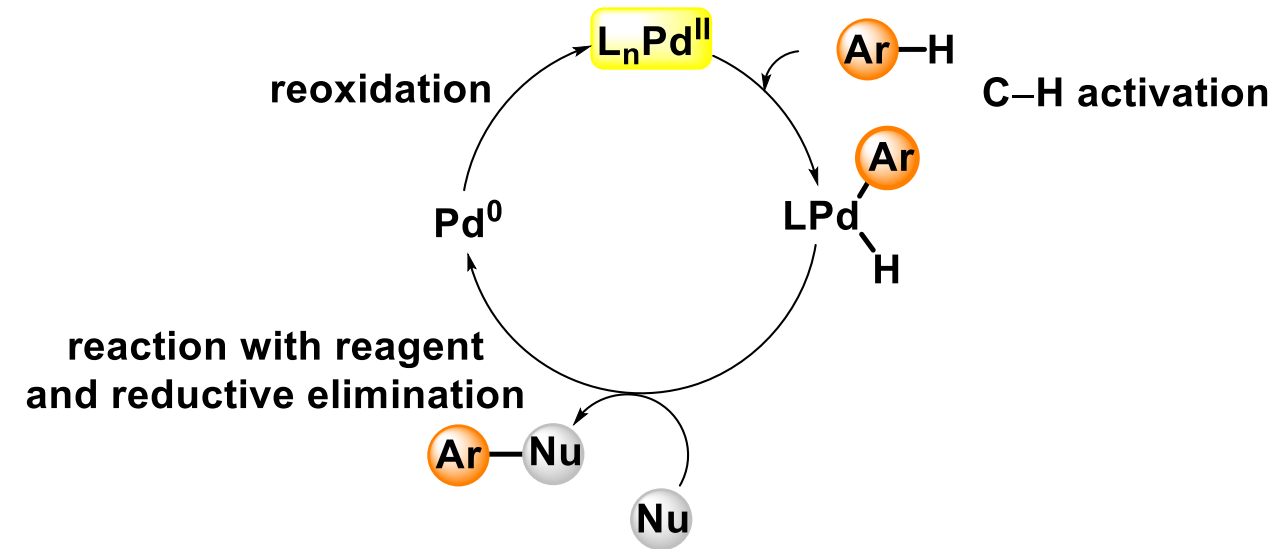
- Palladium catalyzed C–H activation



- Traditional cross-coupling Pd<sup>0</sup>/Pd<sup>II</sup>/Pd<sup>0</sup>:

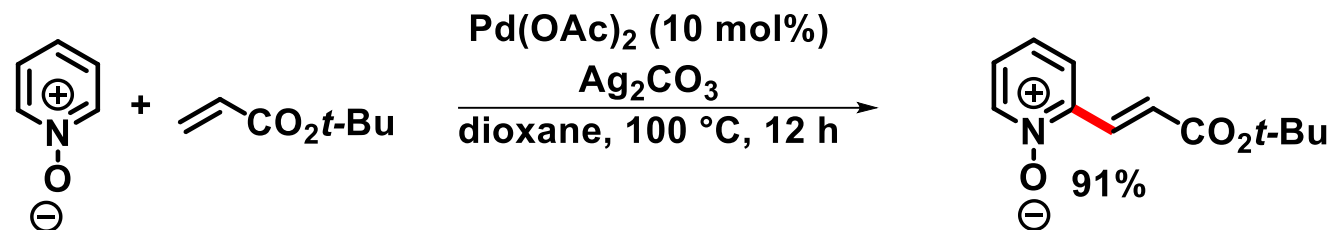
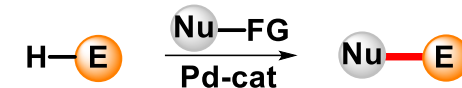


- Oxidative C–H activation Pd<sup>II</sup>/Pd<sup>0</sup>/Pd<sup>II</sup> catalysis:

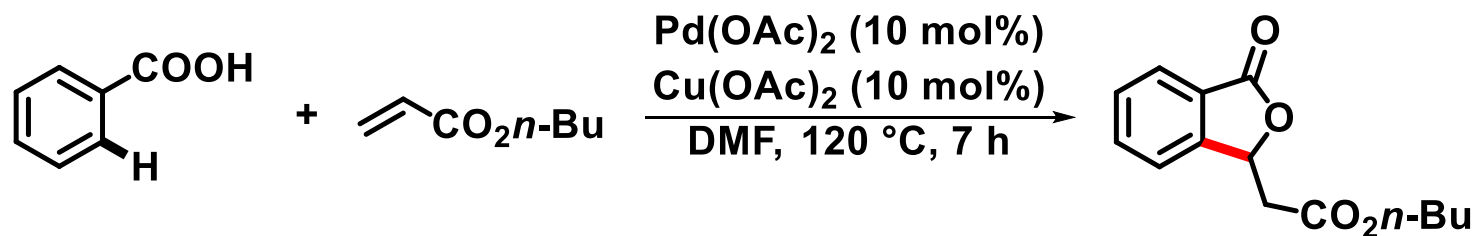


## ➤ C–H activation

- Selected examples for Pd<sup>II</sup>/Pd<sup>0</sup>/Pd<sup>II</sup> cat. cycle



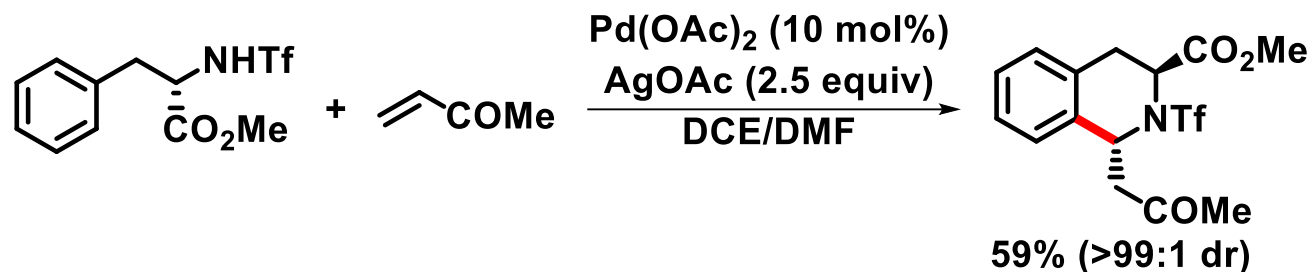
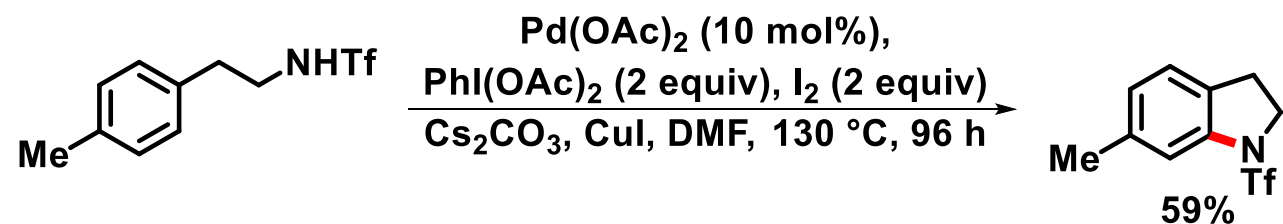
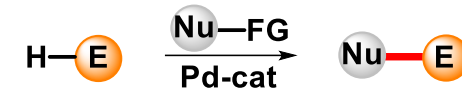
*J. Am. Chem. Soc.* **2008**, 130, 9254



*J. Org. Chem.* **1998**, 63, 5211

## ➤ C–H activation

- Selected examples for Pd<sup>II</sup>/Pd<sup>0</sup>/Pd<sup>II</sup> cat. cycle

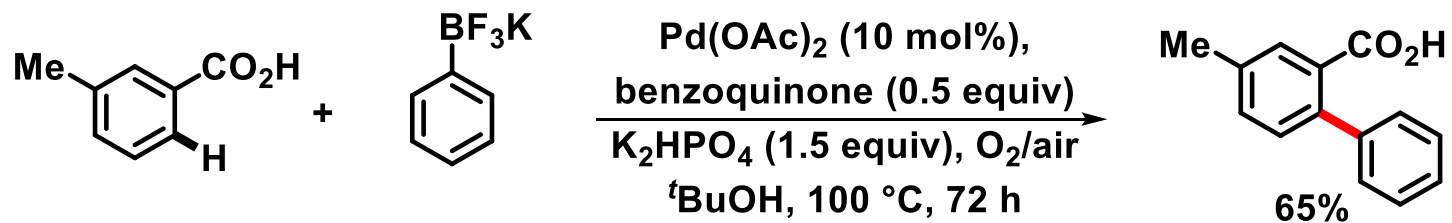


*Angew. Chem. Int. Ed.* **2008**, *47*, 6452

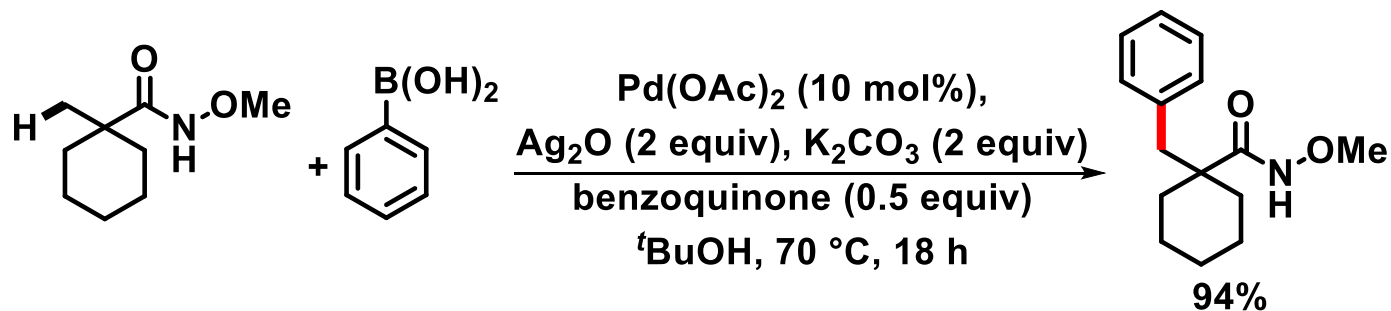


## ➤ C–H activation

- C–H activation with organometallic reagents – Merged catalytic cycle



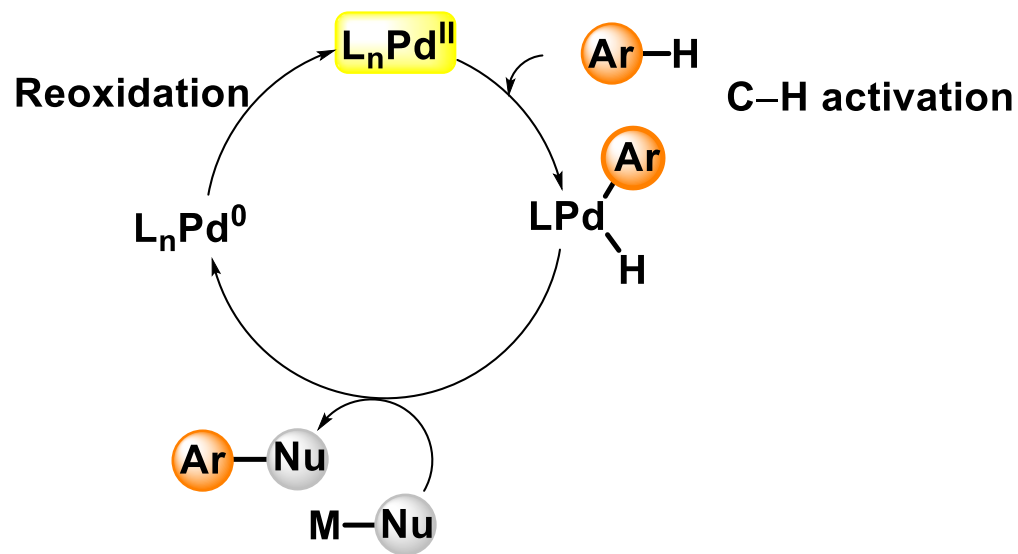
*J. Am. Chem. Soc.* **2008**, *130*, 17676



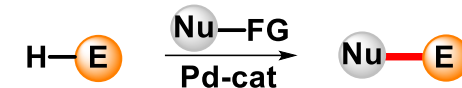
*J. Am. Chem. Soc.* **2008**, *130*, 7190

## ➤ C–H activation

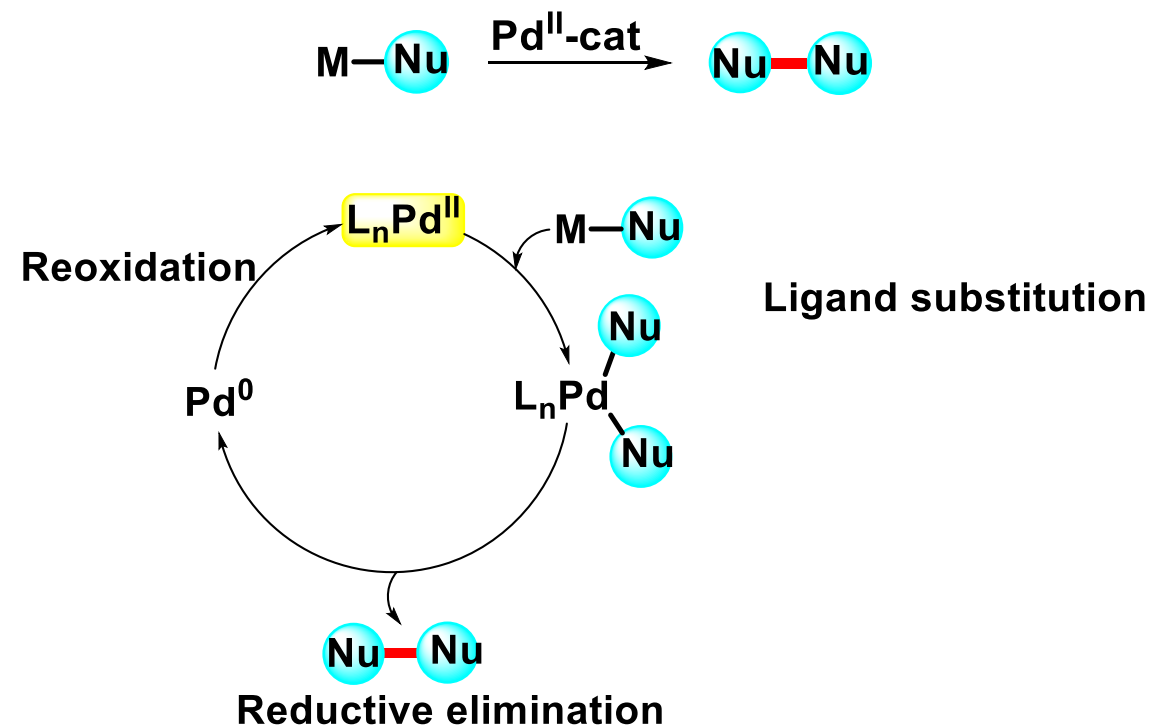
- C–H activation with organometallic reagents
  - Catalytic cycle



Reaction with organometallic reagent ( $M = B(OR)_2, SnR_3, etc$ ) and reductive elimination

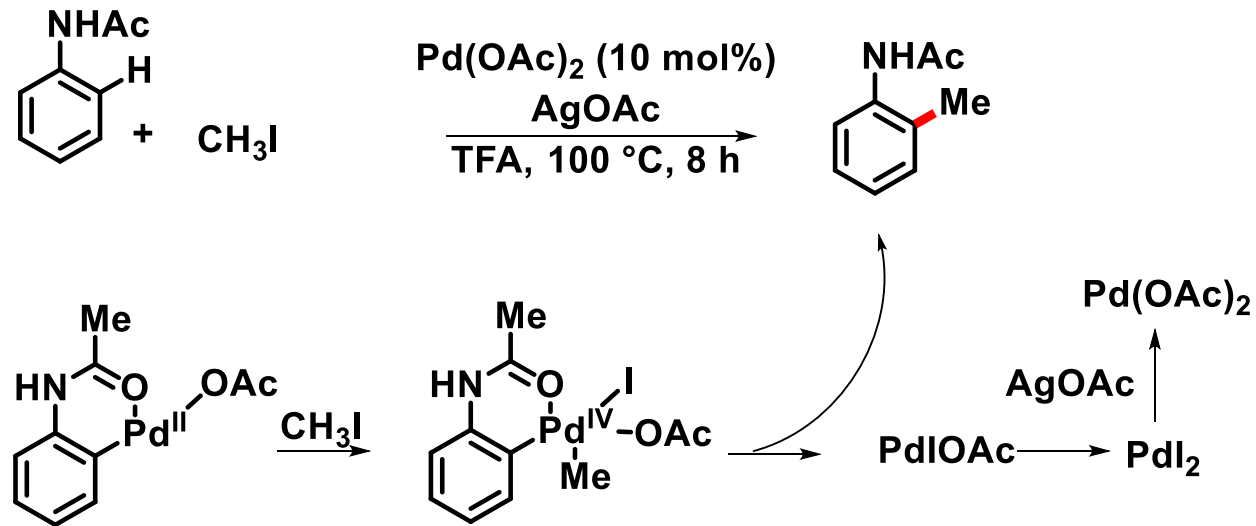


- Possible side reaction

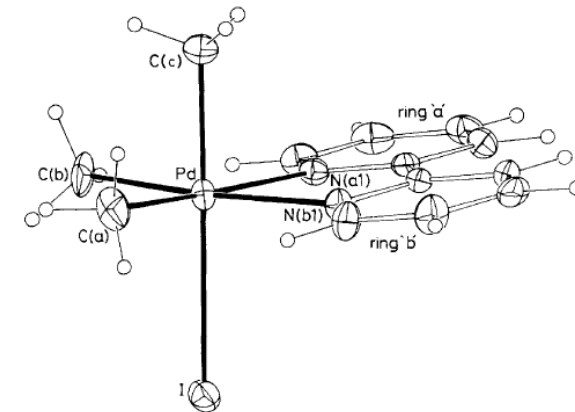
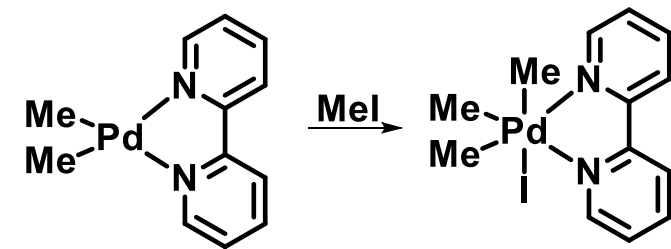
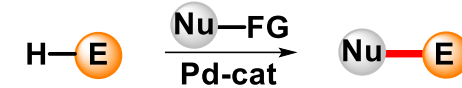


## ➤ C–H activation

- Pd<sup>II</sup>/Pd<sup>IV</sup> catalytic cycle



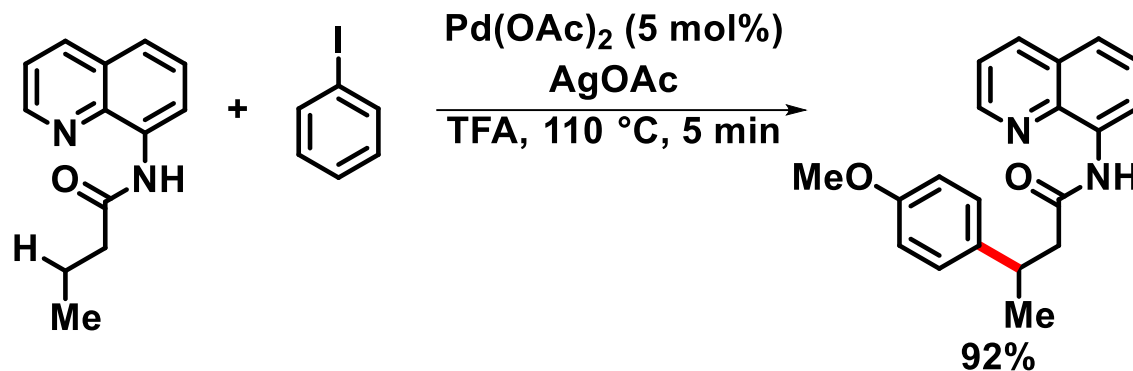
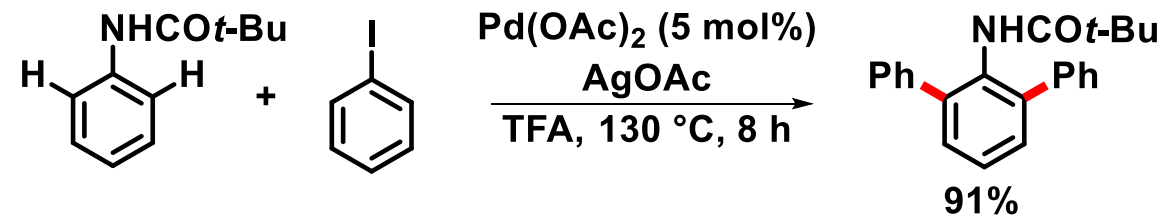
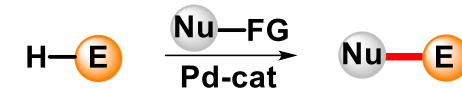
*J. Am. Chem. Soc.* **1984**, 106, 5759



*J. Chem. Soc. Chem. Commun.* **1986**, 1722

## ➤ C–H activation

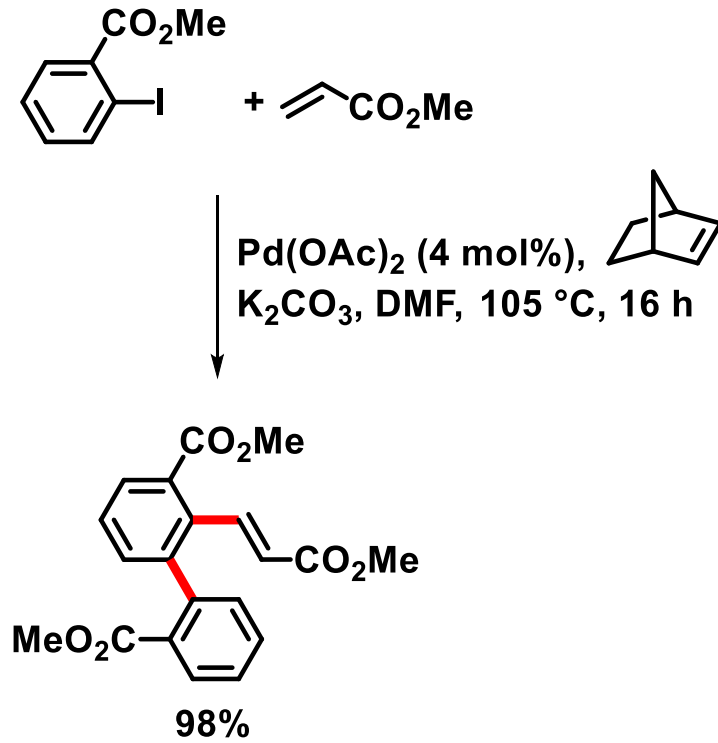
- Pd<sup>II</sup>/Pd<sup>IV</sup> catalytic cycle



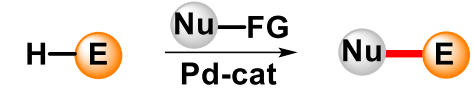
*J. Am. Chem. Soc.* **2005**, *127*, 13154

## ➤ C–H activation

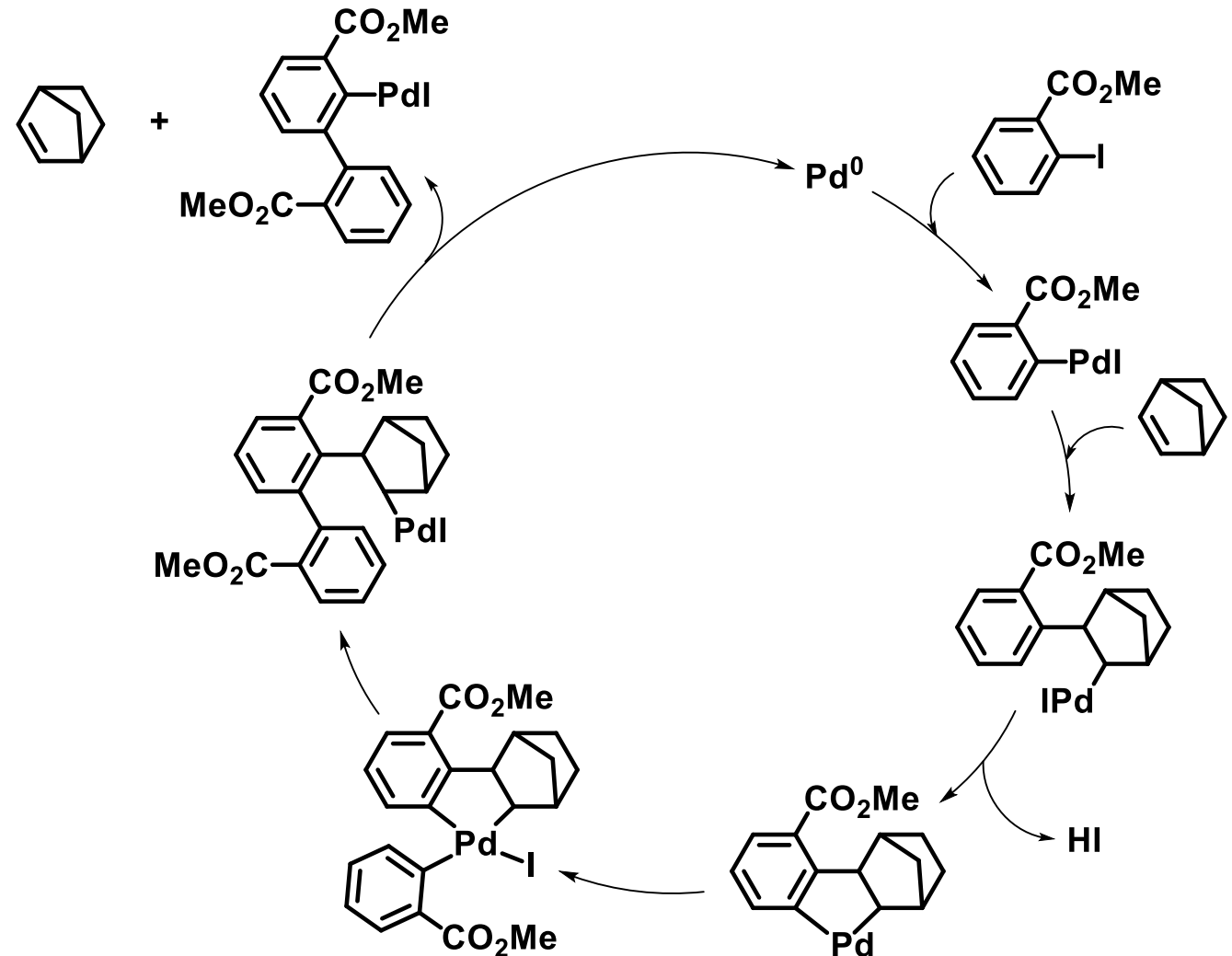
- Pd<sup>II</sup>/Pd<sup>IV</sup> catalytic cycle – Catellani reaction



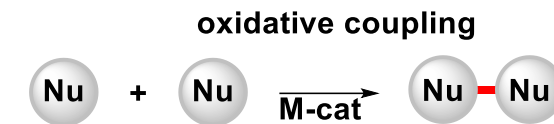
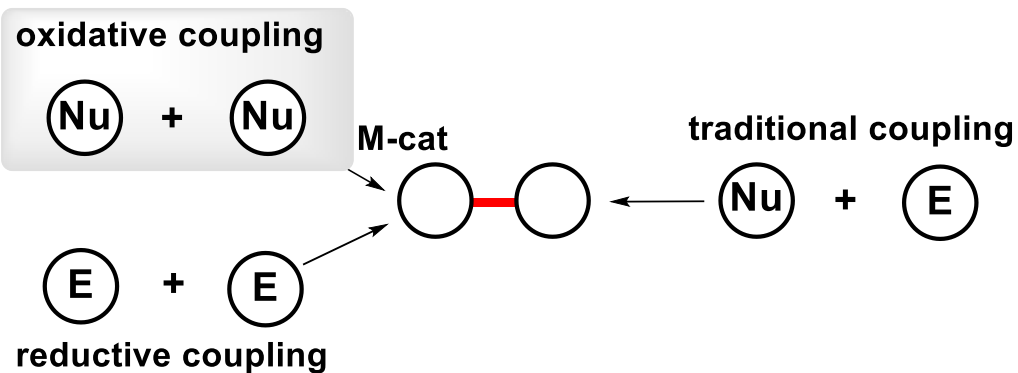
*Synthesis* 2003, 2671



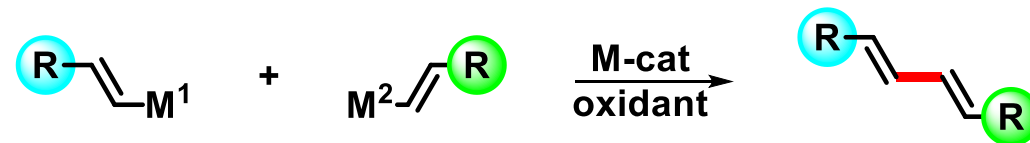
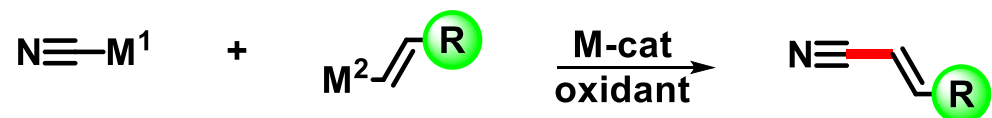
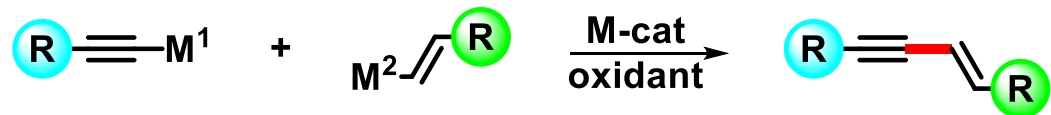
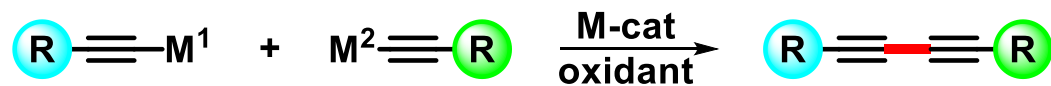
## ○ Proposed mechanism



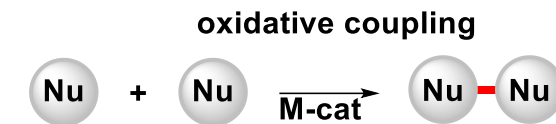
## ➤ Oxidative cross-coupling



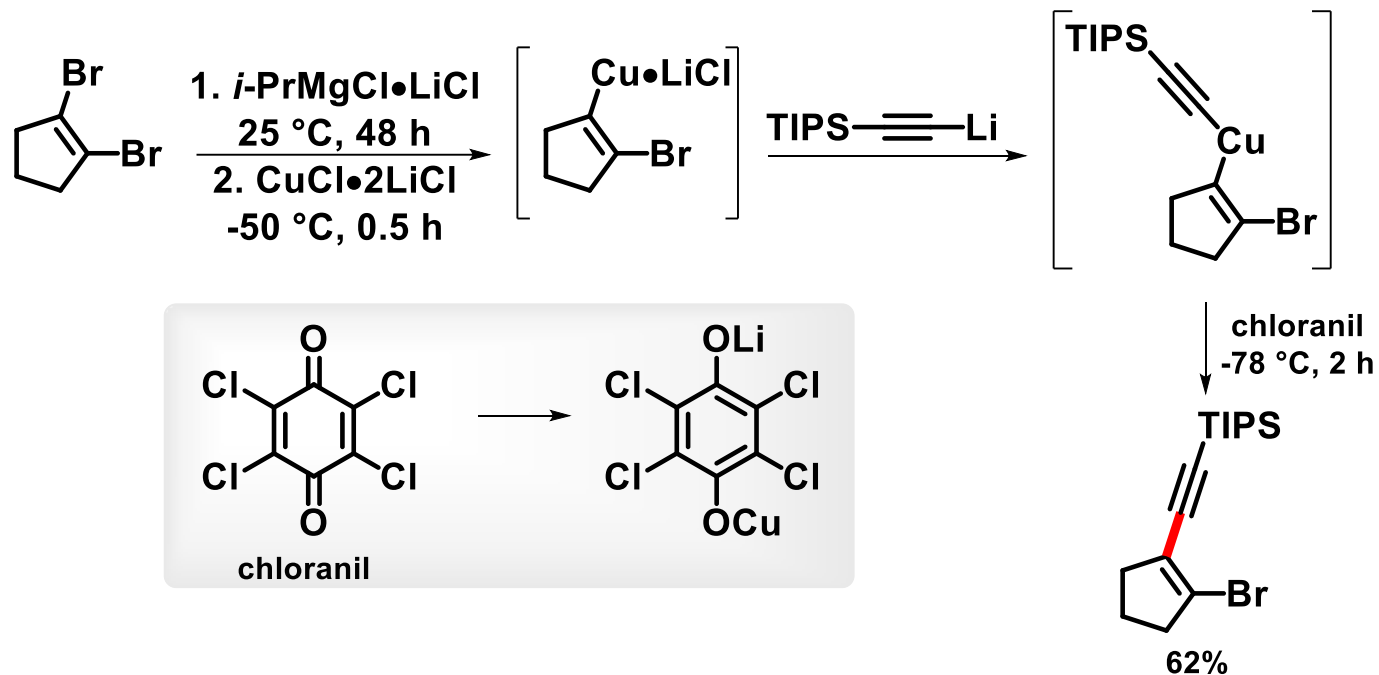
- Common oxidants:  
 Di(*tert*-butyl)peroxide (DTBP)  
*tert*-butylhydroperoxide (TBHP)  
 Benzoquinone/O<sub>2</sub>, O<sub>2</sub>  
 Cu(OAc)<sub>2</sub>, AgOAc



## ➤ Oxidative cross-coupling



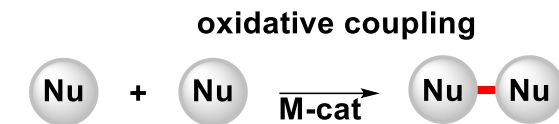
- Selected examples



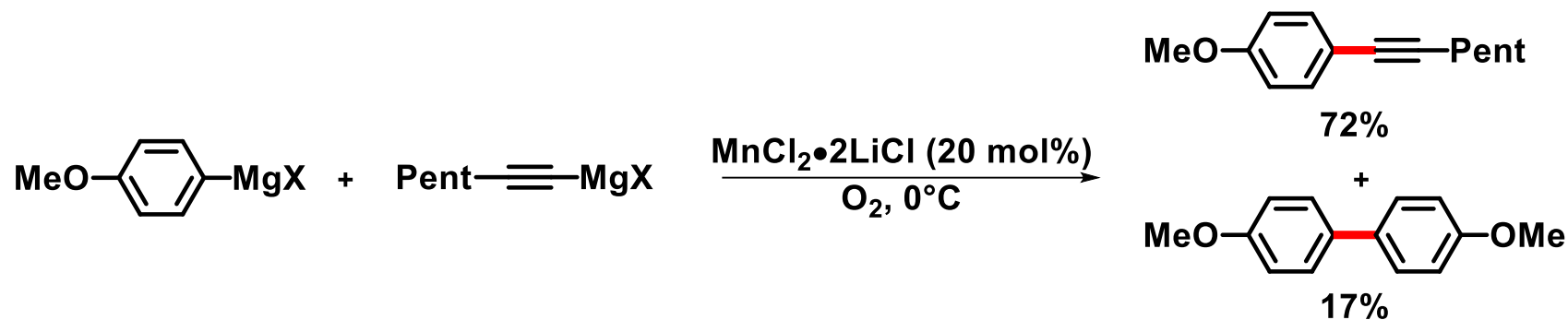
- Stoichiometric reaction

Angew. Chem. Int. Ed. 2007, 46, 9093

## ➤ Oxidative cross-coupling



- Selected examples – catalytic reactions



*Angew. Chem., Int. Ed.* **2009**, *48*, 6731

- Other organometals

**ArZnCl / R<sub>3</sub>In**  
**PdCl<sub>2</sub>(MeCN)<sub>2</sub>/DPEPhos**  
**desyl chloride**

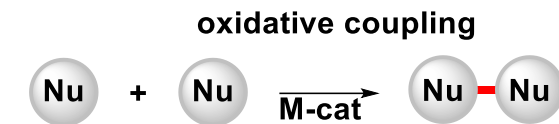
*Angew. Chem., Int. Ed.* **2009**, *48*, 2969

**ArZnCl / RC≡CSnBu<sub>3</sub>**  
**Pd(dba)<sub>2</sub> (2.5 mol%)**  
**desyl chloride**

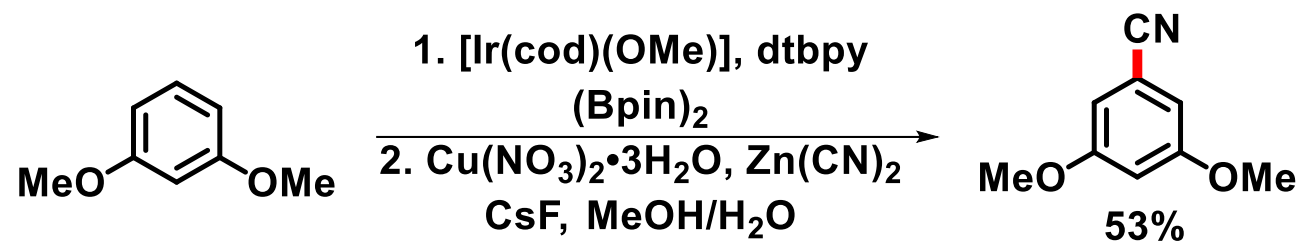
*J. Am. Chem. Soc.* **2006**, *128*, 15048



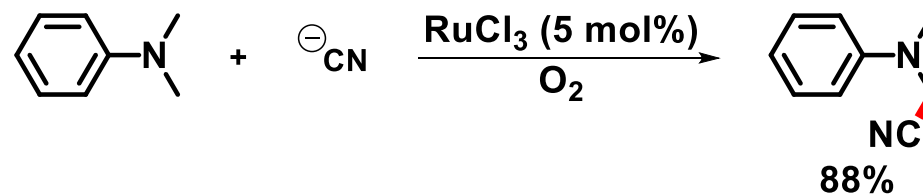
## ➤ Oxidative cross-coupling



- Selected examples – catalytic reactions



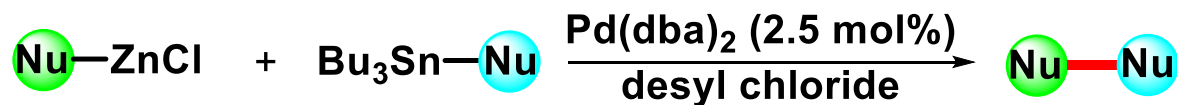
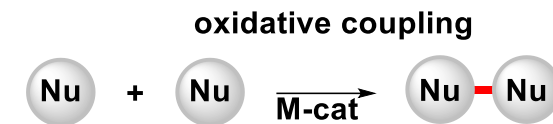
*J. Am. Chem. Soc.* **2010**, *132*, 11389



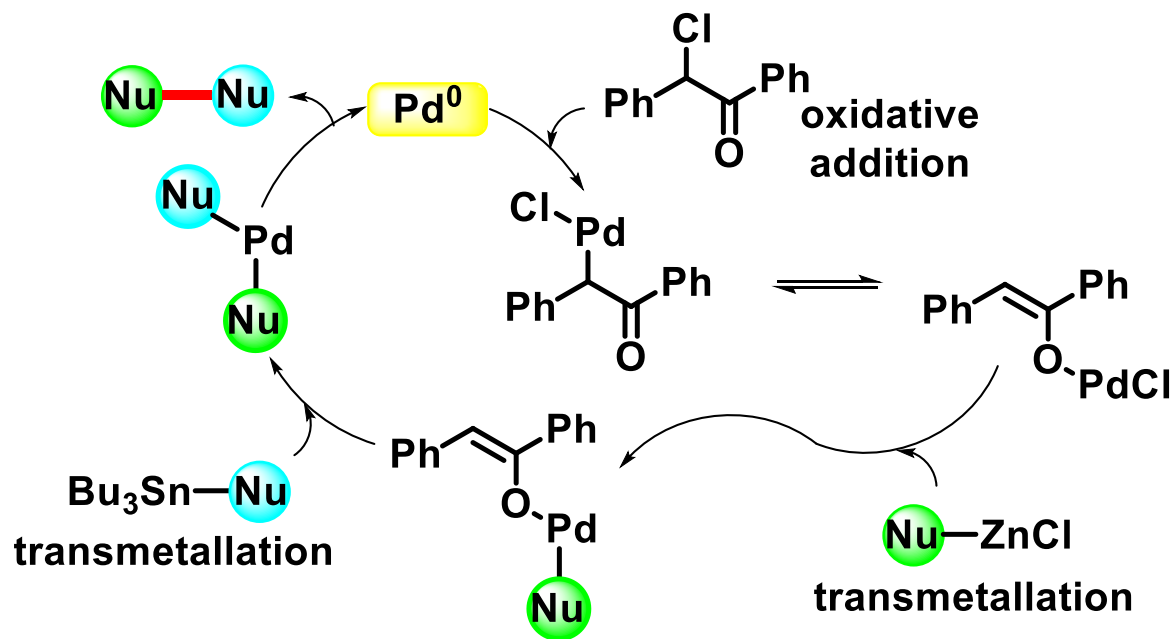
*J. Am. Chem. Soc.* **2003**, *125*, 15312

## ➤ Oxidative cross-coupling

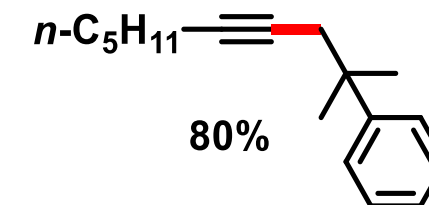
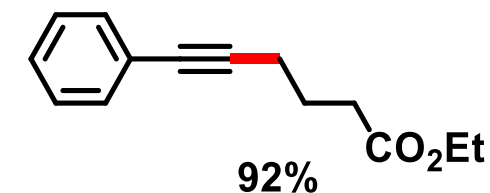
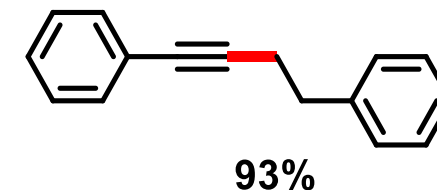
- Mechanism of oxidative cross-coupling



reductive elimination



## ○ Selected examples

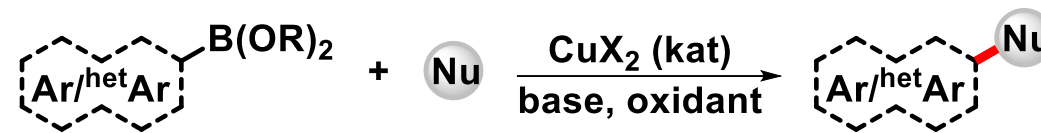


*J. Am. Chem. Soc.* 2006, 128, 15048

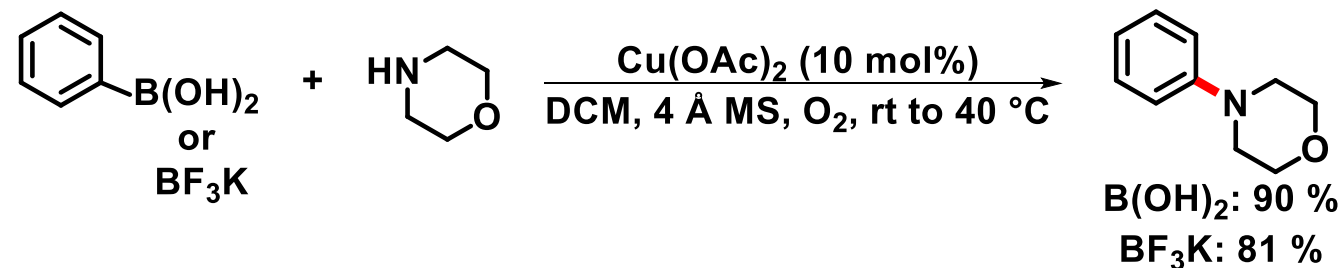
## ➤ Oxidative cross-coupling

- Chan-Lam reaction

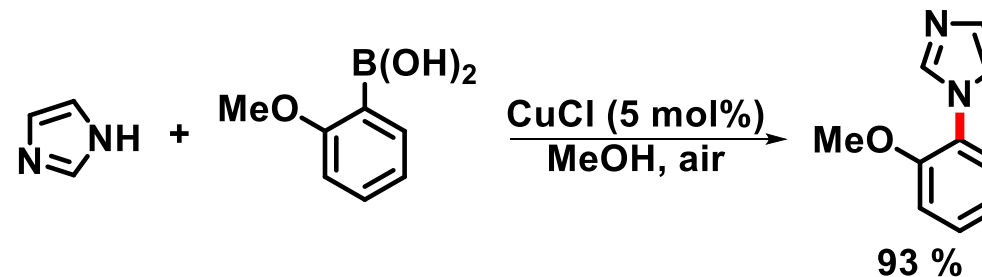
oxidative coupling



Nu = amine, alcohol



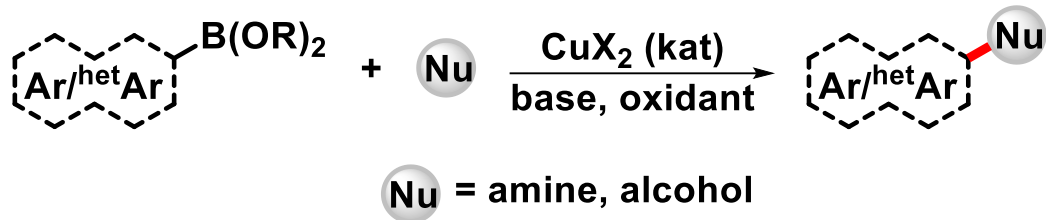
*Org. Lett.* 2003, 5, 4397



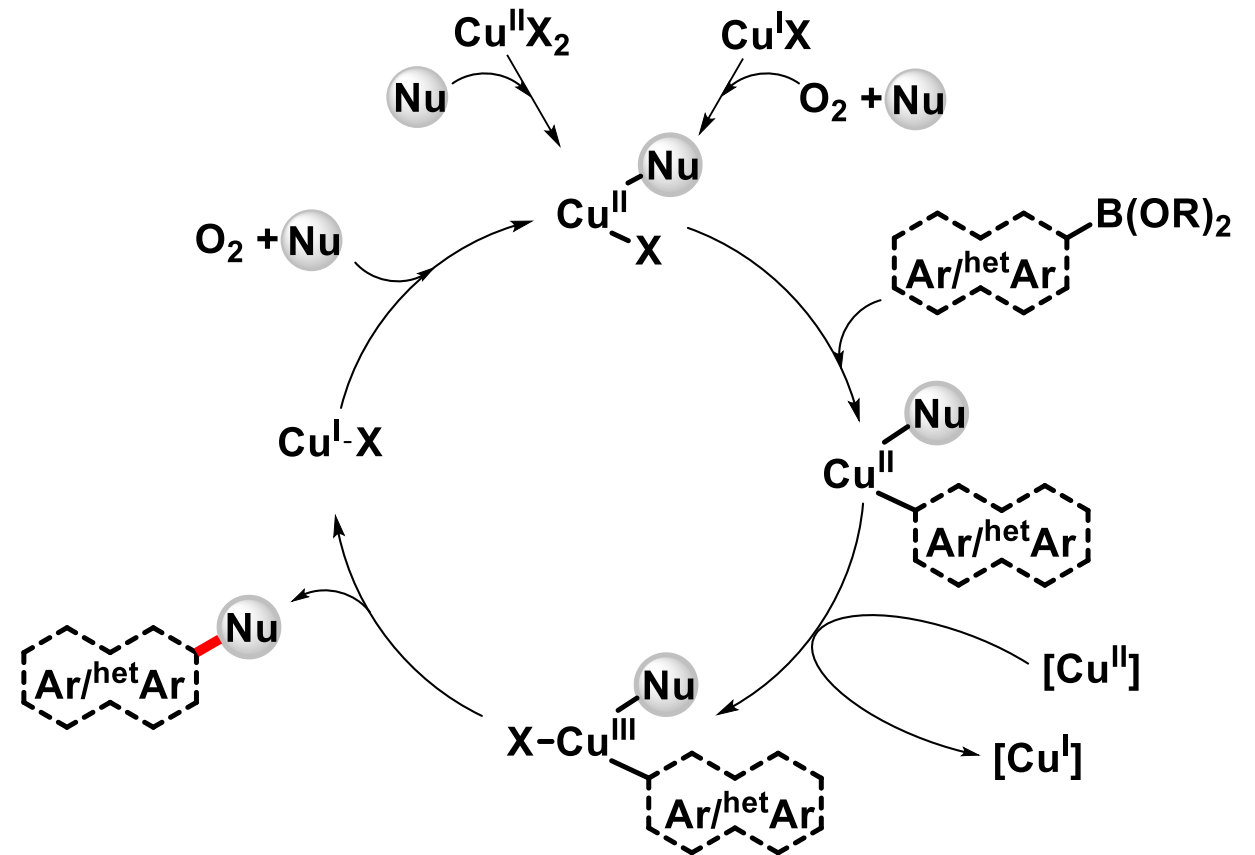
*Chem. Commun.* 2004, 188

## ➤ Oxidative cross-coupling

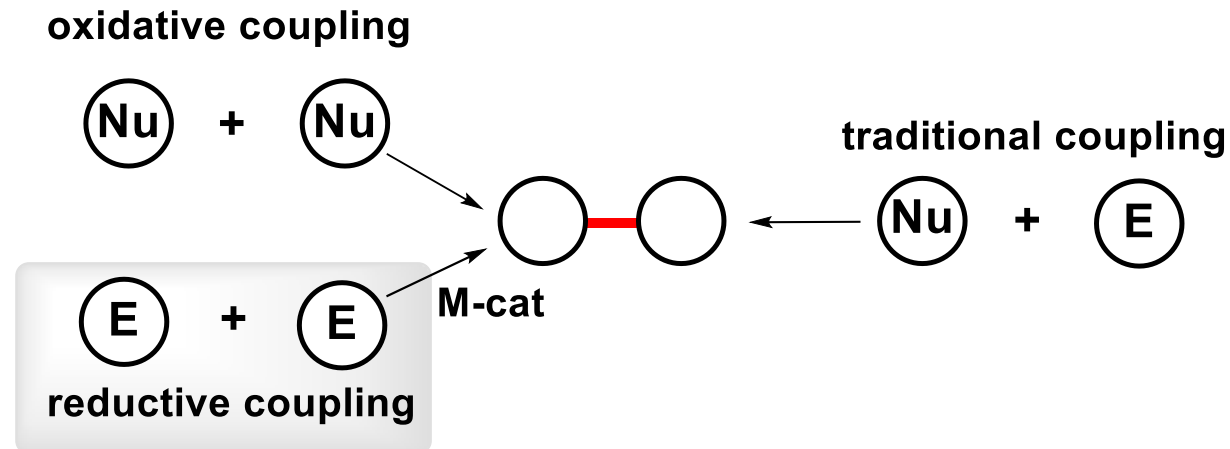
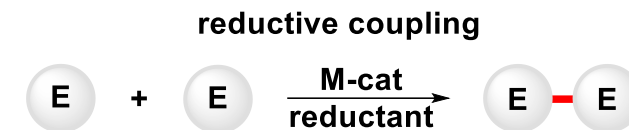
- Chan-Lam reaction
  - Mechanism



oxidative coupling



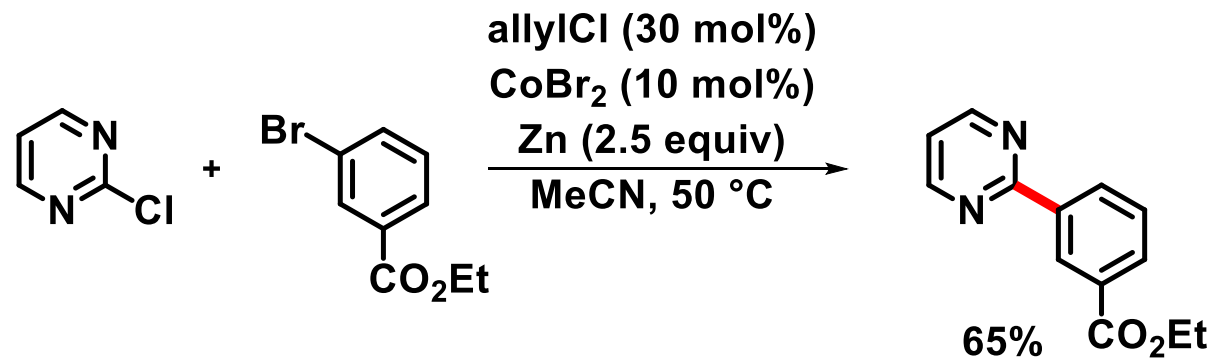
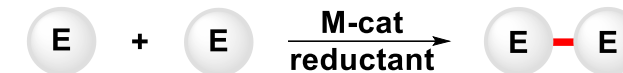
## ➤ Reductive cross-coupling



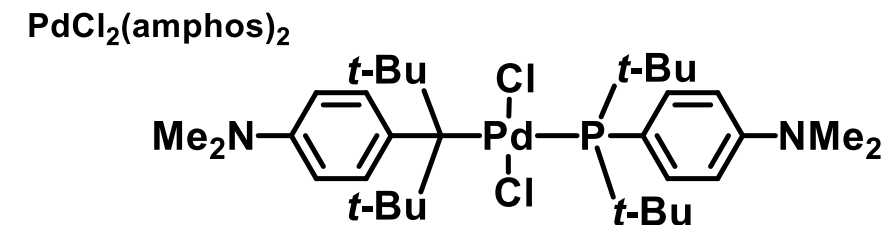
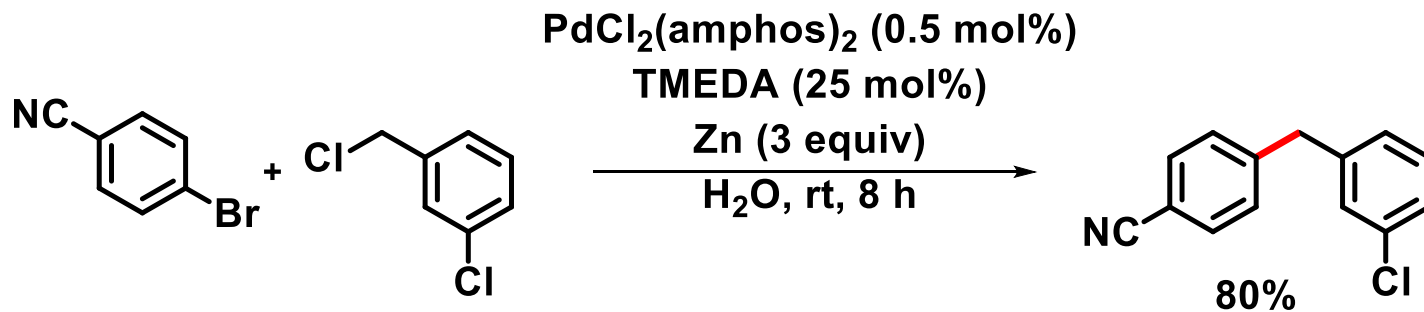
## ➤ Reductive cross-coupling

- Selected examples

reductive coupling



*J. Org. Chem.* 2009, 74, 3221

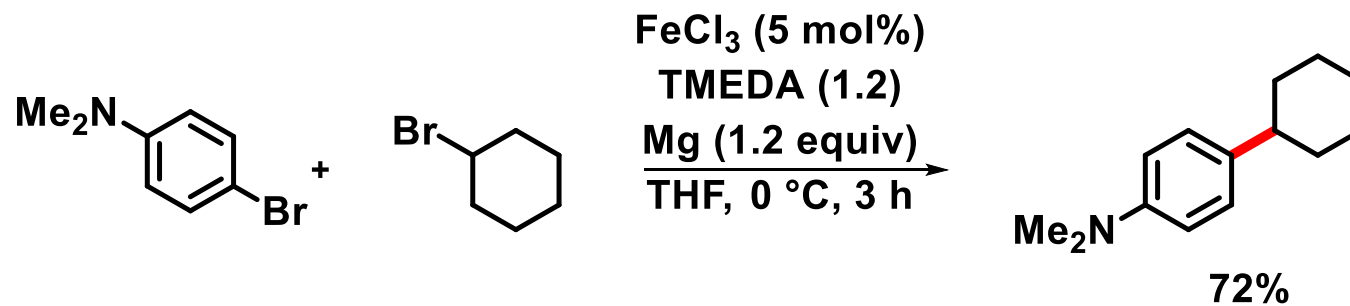
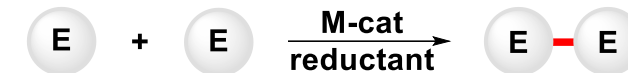


*Chem. Commun.* 2010, 46, 562

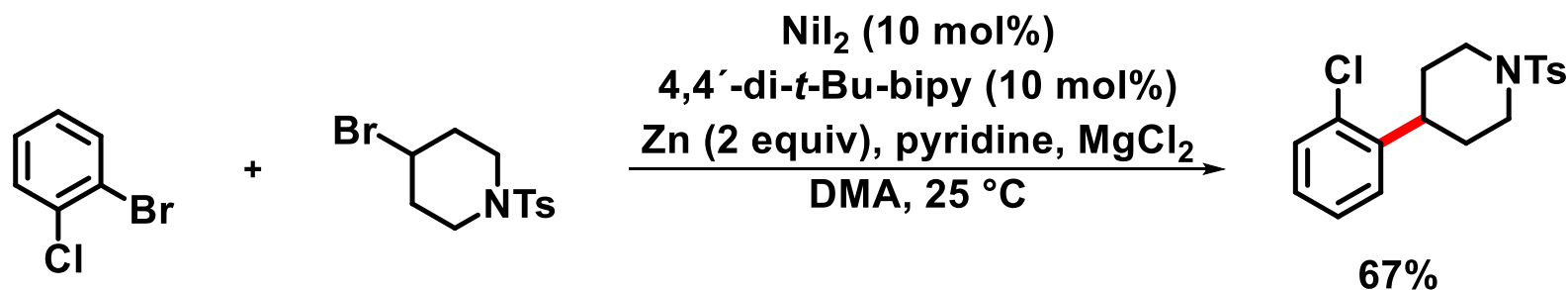
## ➤ Reductive cross-coupling

- Selected examples

reductive coupling



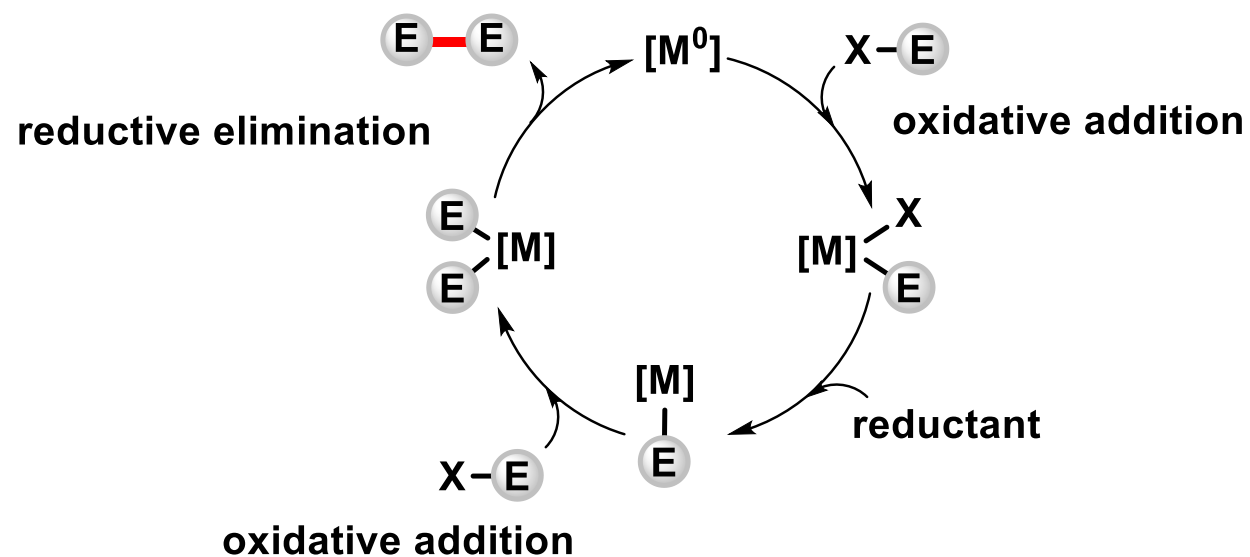
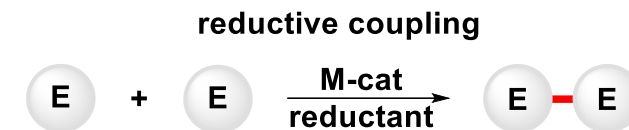
*Angew. Chem. Int. Ed.* **2009**, 48, 607



*Org. Lett.* **2012**, 14, 3352

## ➤ Reductive cross-coupling

- General mechanism of reductive cross-coupling



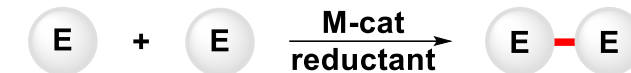
- Typical reductant: Zn, Mg, Cu



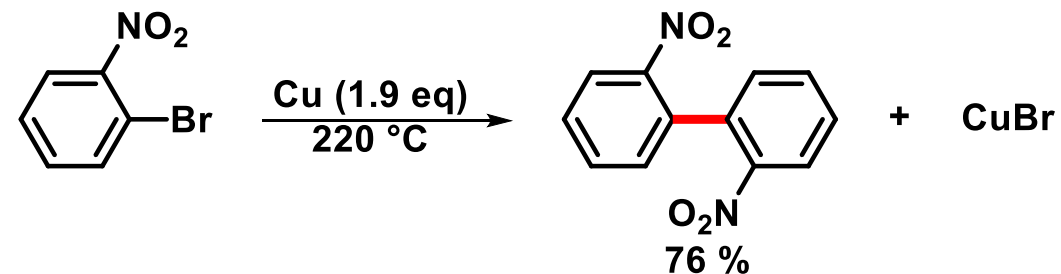
## ➤ Reductive cross-coupling

- Ullmann reaction

reductive coupling



- The first dimerization was published in 1901

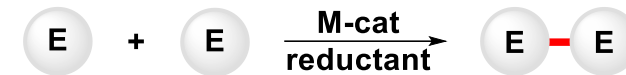


*Ber. Dtsch. Chem. Ges.* **1901**, 34, 2174

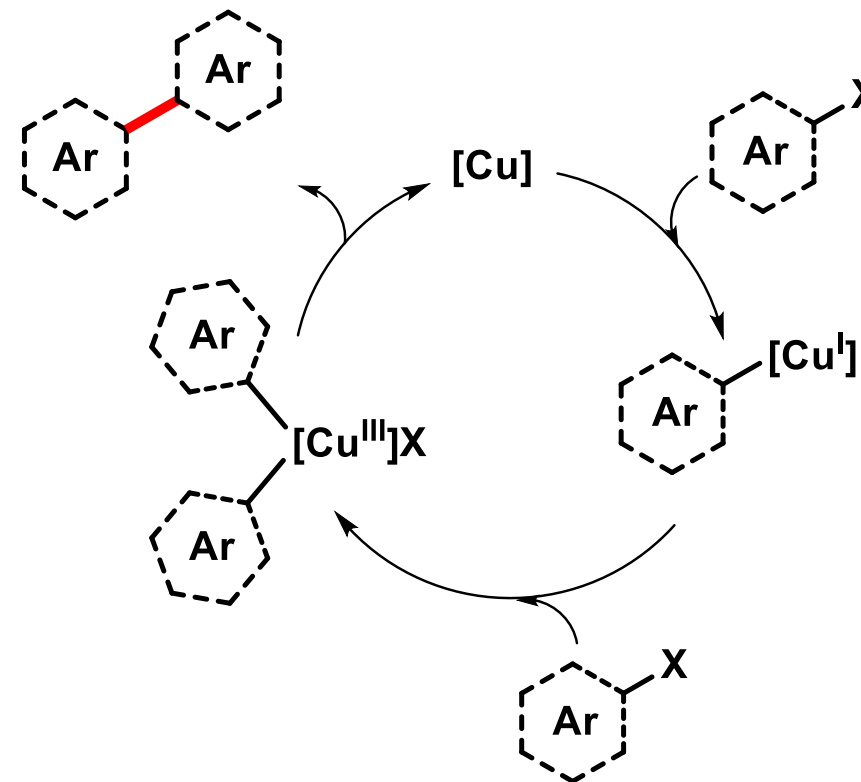
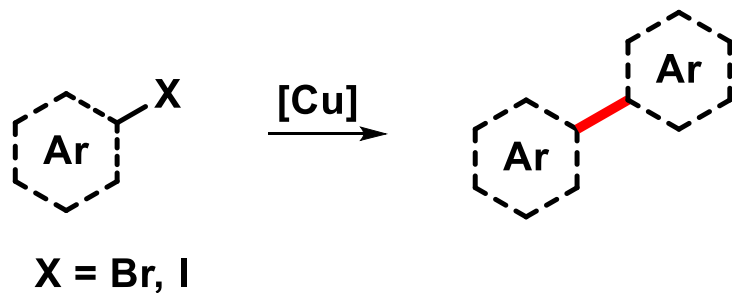
## ➤ Reductive cross-coupling

- Ullmann reaction

reductive coupling



- General reaction scheme



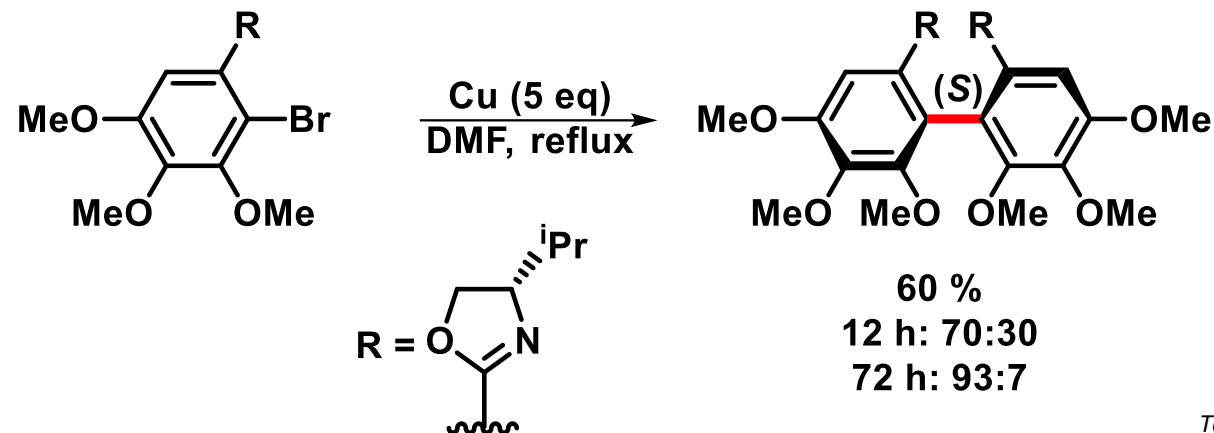
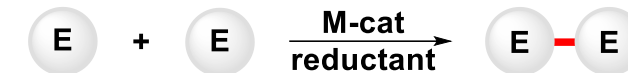
*Synthesis* 1974, 20, 9

## ➤ Reductive cross-coupling

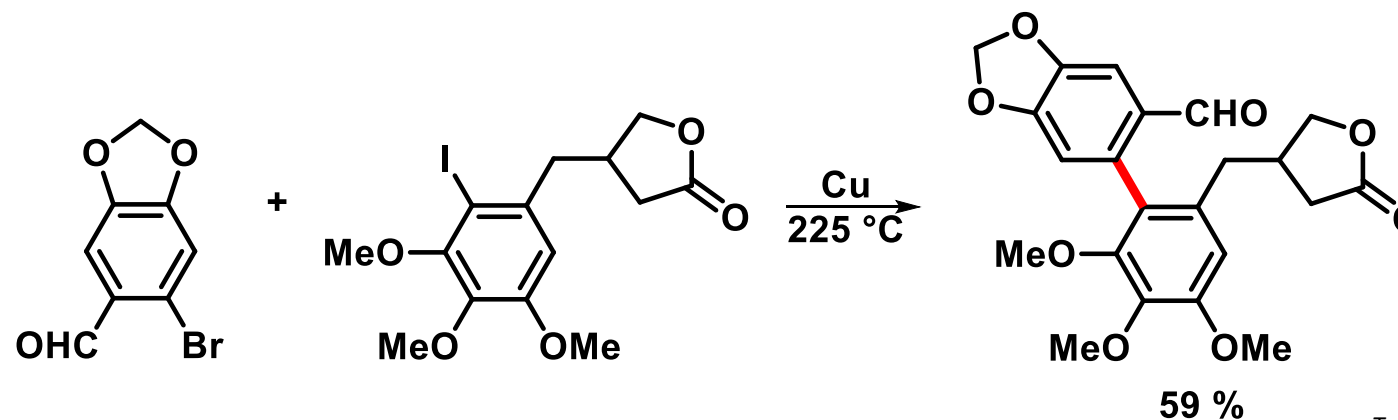
- Ullmann reaction

- Selected examples

reductive coupling



*Tetrahedron Lett.* 1993, 34, 3061

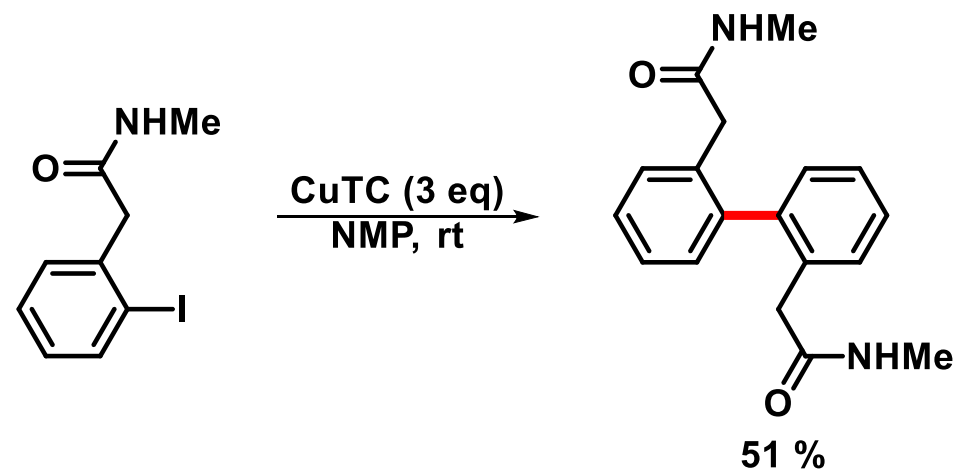
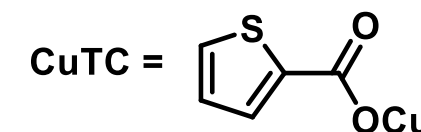
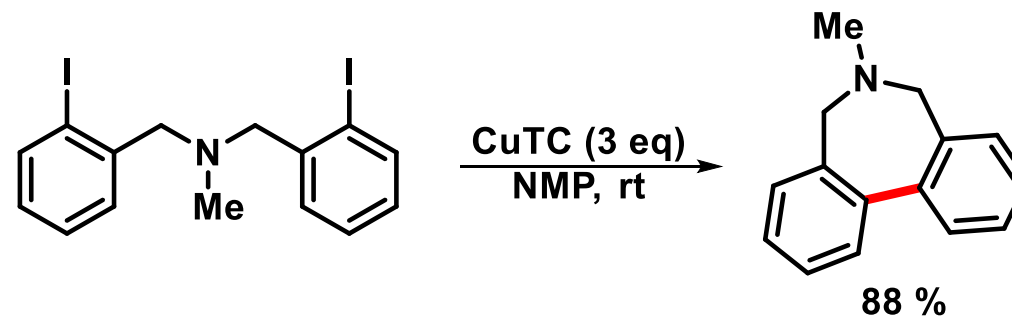
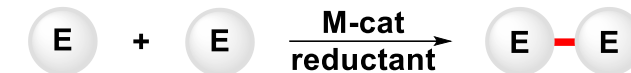


*Tetrahedron Lett.* 1979, 20, 773

## ➤ Reductive cross-coupling

- Ullmann reaction
  - Selected examples

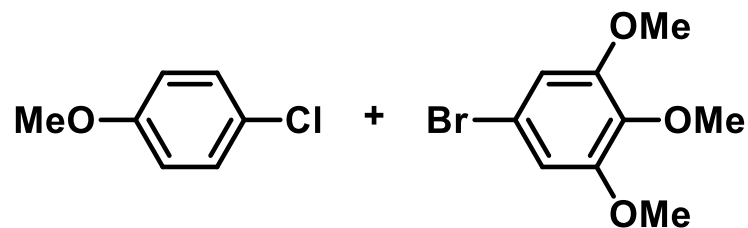
reductive coupling



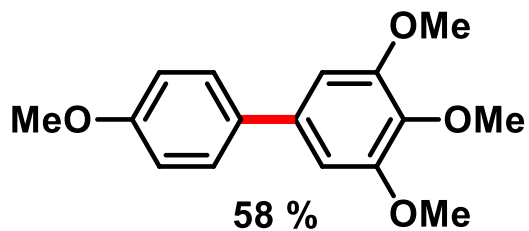
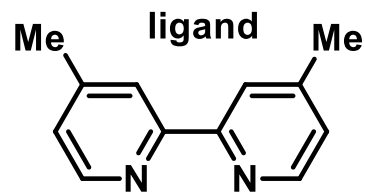
*J. Org. Chem.* **1997**, *62*, 2312

## ➤ Reductive cross-coupling

- Nickel-catalyzed reactions
  - Selected examples



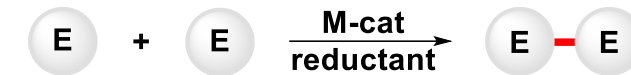
$\text{NiI}_2$  (10 mol%)  
 ligand (10 mol%)  
 Zn (2 eq)  
 pyridine (1 eq)  
 $\text{Bu}_4\text{NI}$  (1 eq)  
 DMA, 25 °C



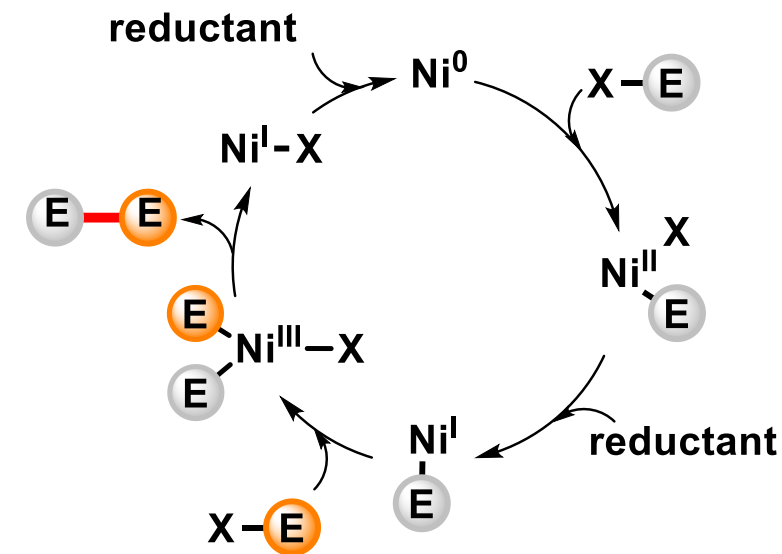
58 %

*Synlett* 2013 , 24 , 619

reductive coupling

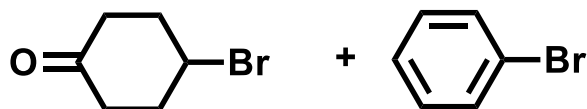


- Typical mechanism

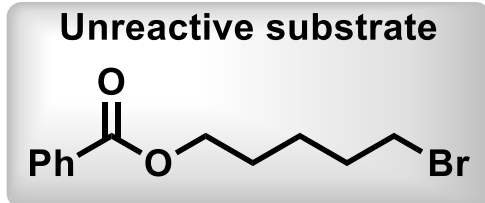
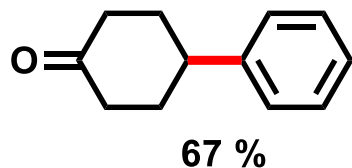
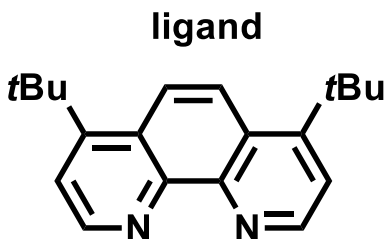


## ➤ Reductive cross-coupling

- Nickel-catalyzed reactions
  - Selected examples

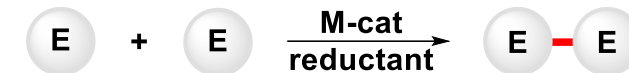


$\text{Ni}^0$  (10 mol%)  
 ligand (10 mol%)  
 $\text{MgCl}_2$  (1.5 eq)  
 $\text{Zn}$  (2 eq), pyridine (1 eq)  
 DMA, 25 °C, 12 h



*Org. Lett.* 2012, 14, 3352

reductive coupling



- Typical mechanism

