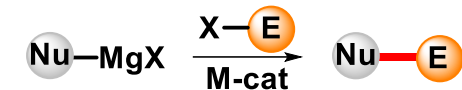


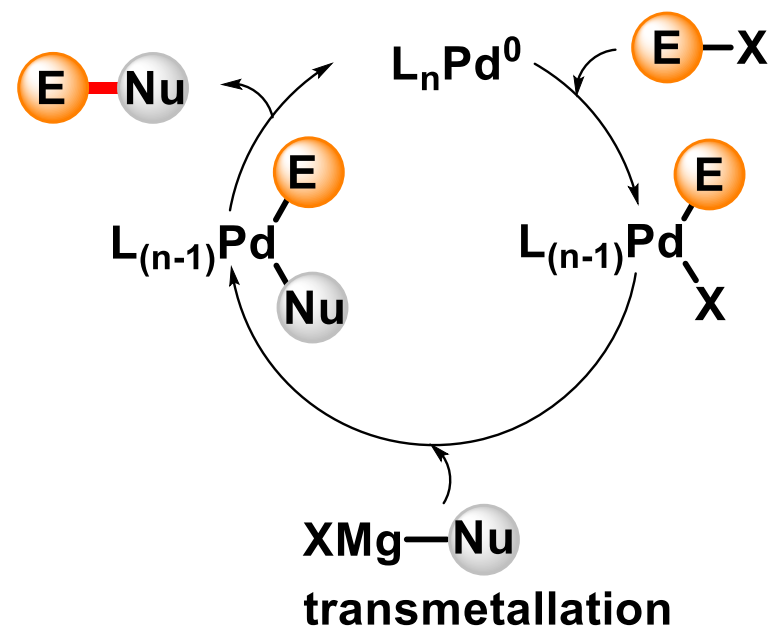
➤ The Kumada-Tamao-Corriu reaction (Kumada reaction)

- Mechanism of the Kumada-Tamao-Corriu reaction



reductive elimination

oxidative addition



➤ The Kumada-Tamao-Corriu reaction (Kumada reaction)



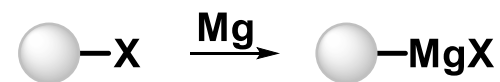
- Synthesis of functionalized Grignard reagents



- Acid–base reaction

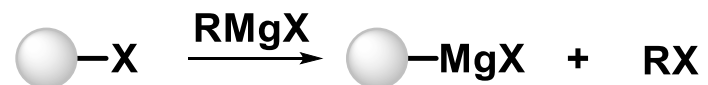


- Reaction with metal



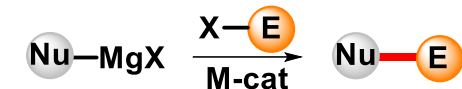
X = Cl, Br, I

- Halogen–Magnesium exchange reaction

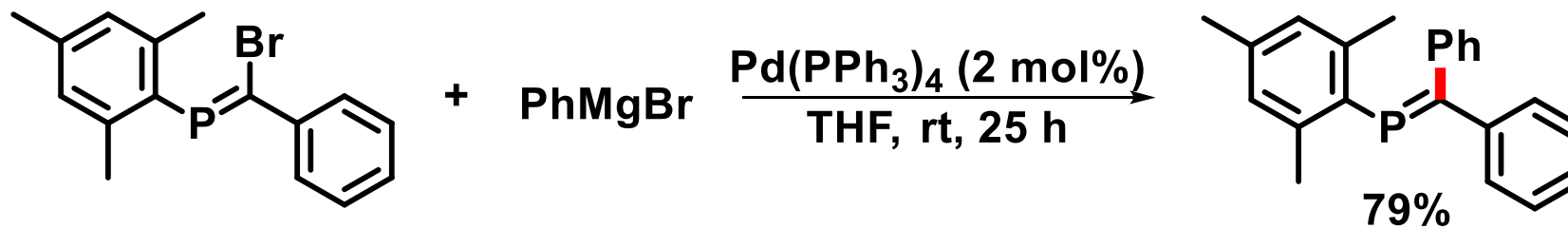


X = Cl, Br, I

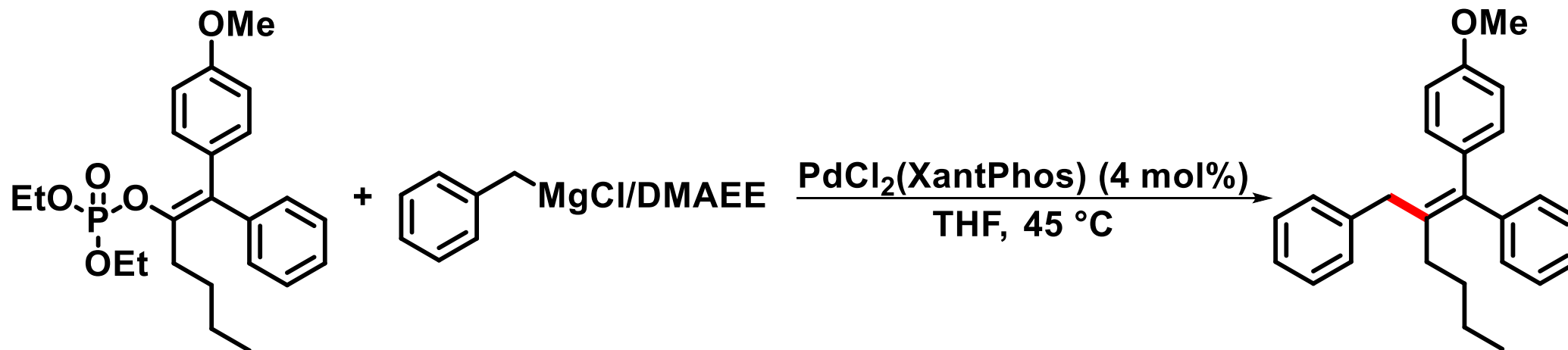
➤ The Kumada-Tamao-Corriu reaction (Kumada reaction)



- Selected examples: Palladium-catalyzed reactions

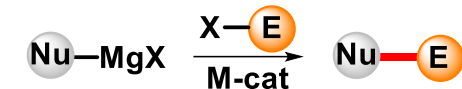


ChemistrySelect 2016, 1, 5260

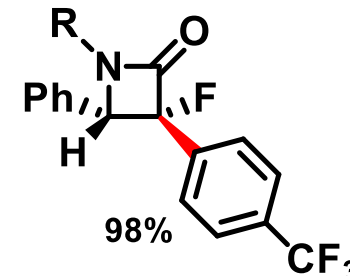
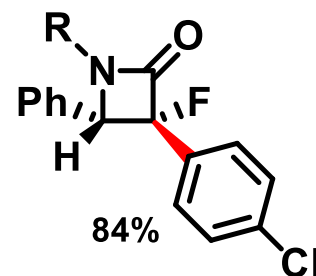
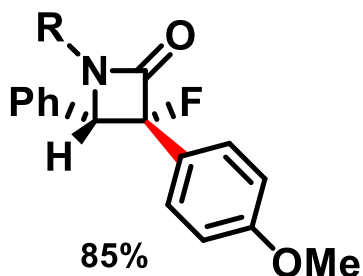
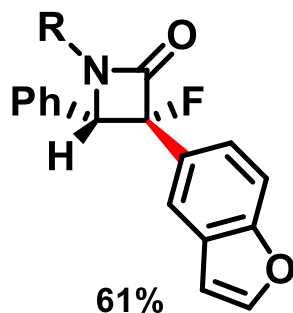
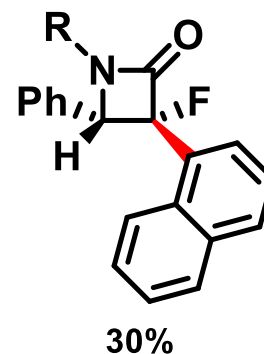
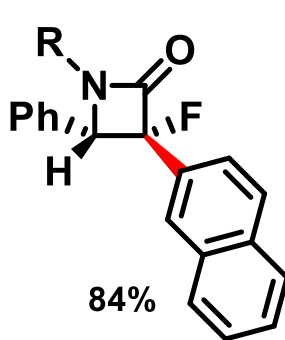
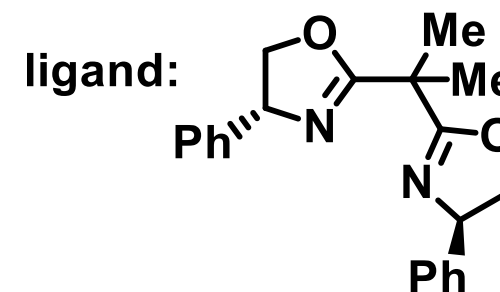
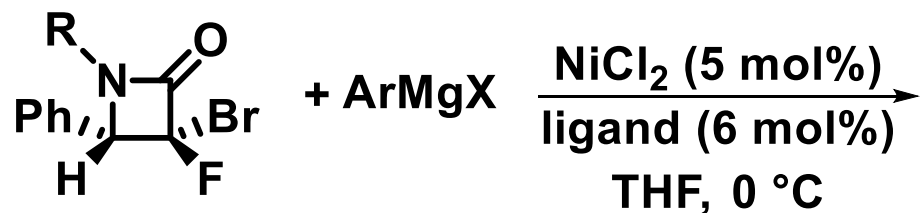


Org. Lett. 2015, 17, 608

➤ The Kumada-Tamao-Corriu reaction (Kumada reaction)

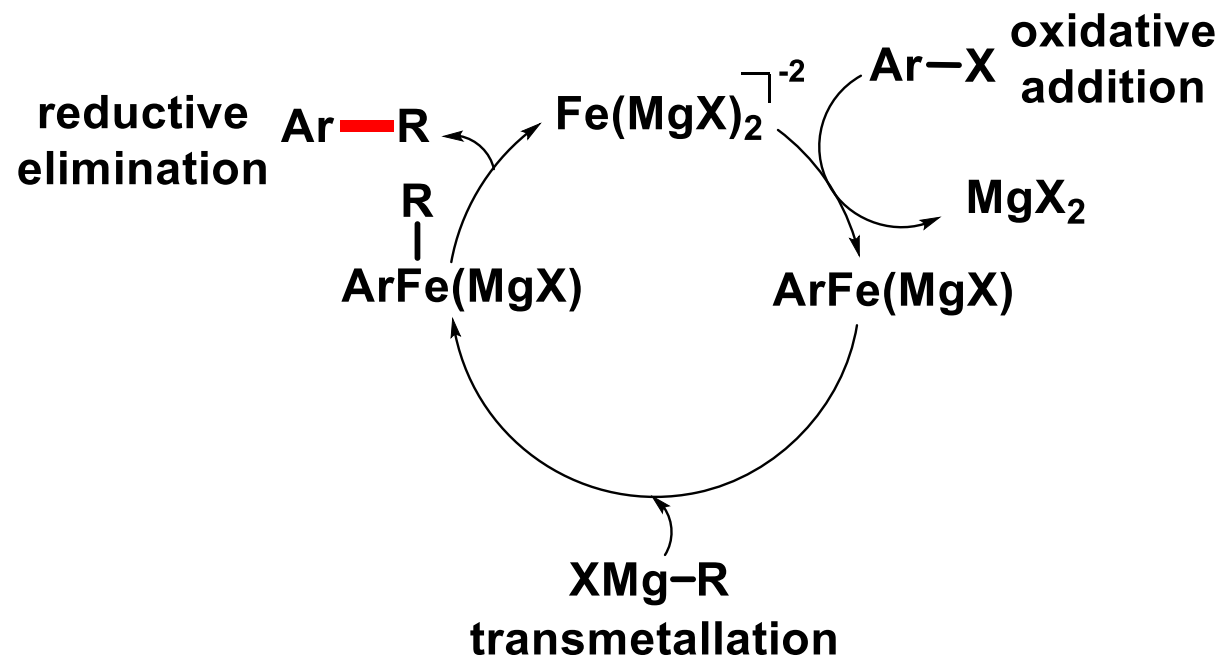
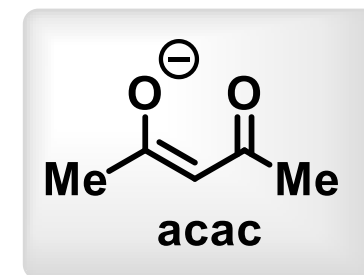
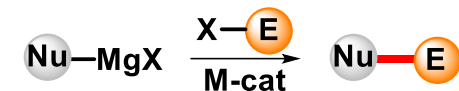
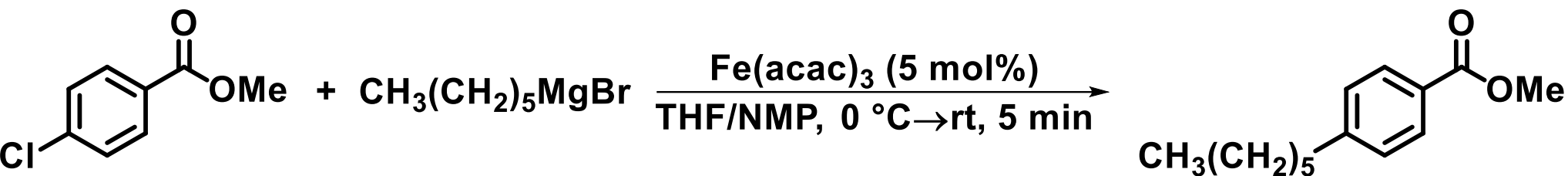


- Selected examples: Nickel-catalyzed reactions



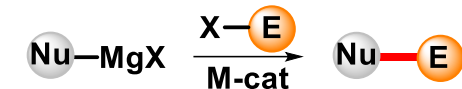
➤ The Kumada-Tamao-Corriu reaction (Kumada reaction)

- Selected examples: Iron-catalyzed reactions and proposed catalytic cycle

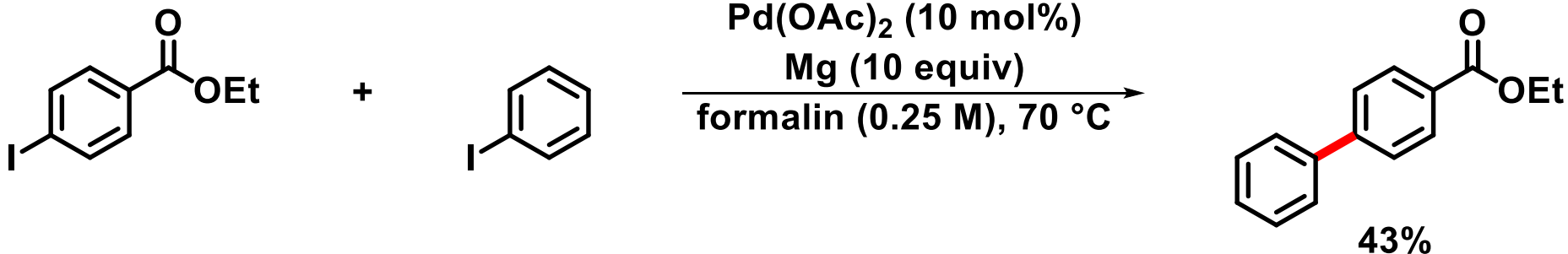


Angew. Chem. Int. Ed. 2002, 41, 609

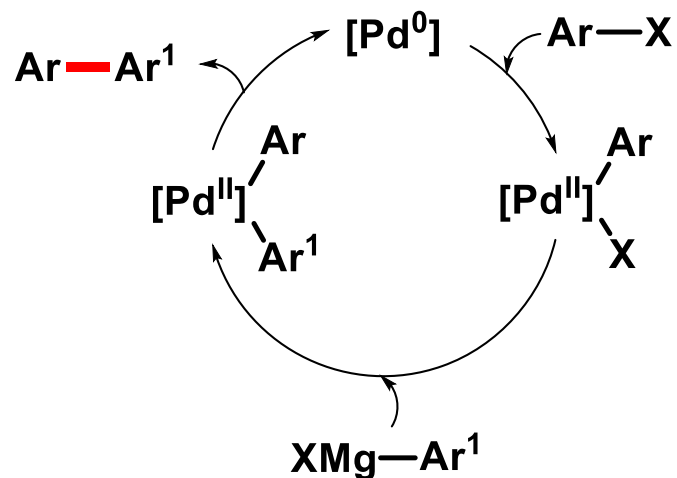
➤ The Kumada-Tasmaso-Corriu reaction (Kumada reaction)



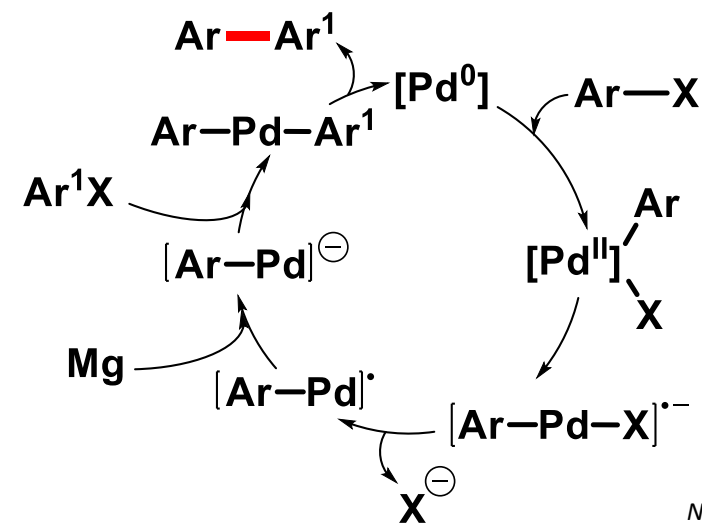
- Kumada reaction „on water“



○ Traditional mechanism



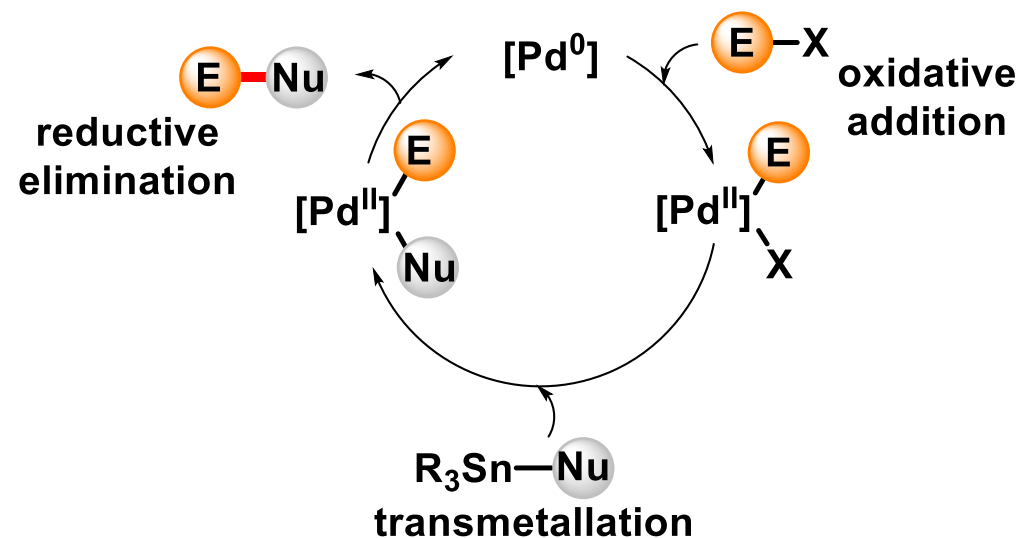
○ Radical mechanism

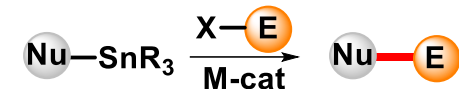


Nat. Commun. 2015, 6, 7401

➤ The Migita-Kosugi-Stille(Stille reaction)

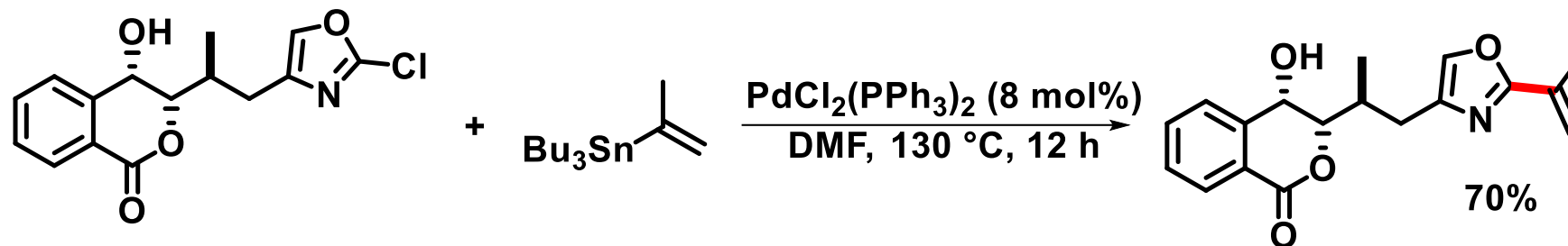
- Mechanism of the Stille reaction



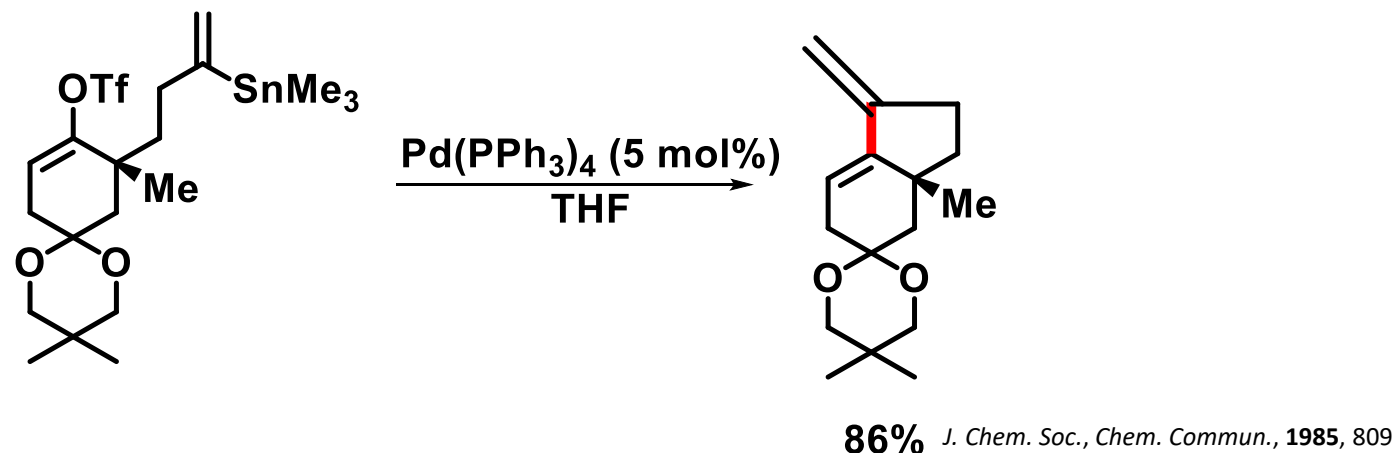


➤ The Stille reaction

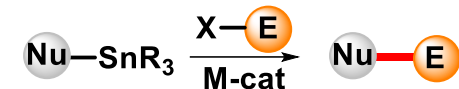
- The Stille reaction – Selected examples



Tetrahedron 2011, 67, 9700

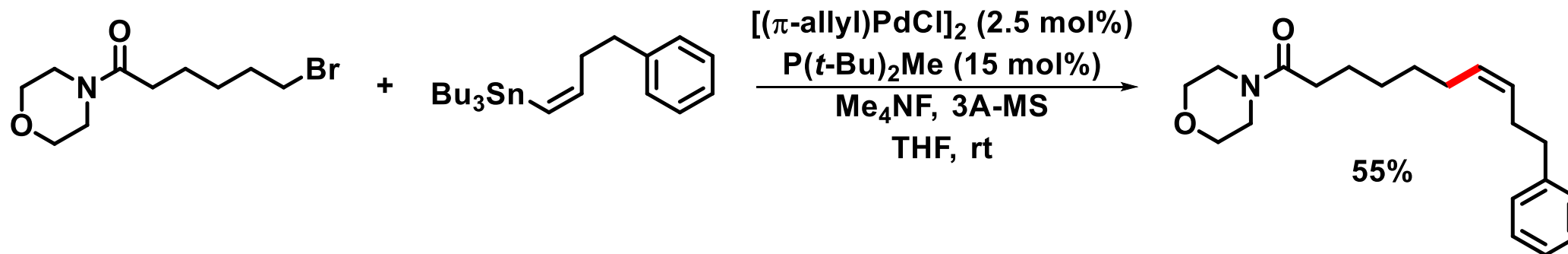
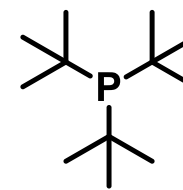
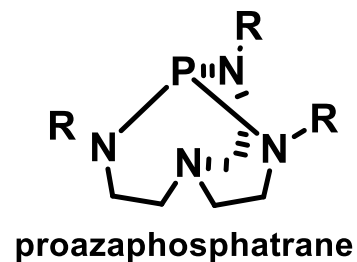
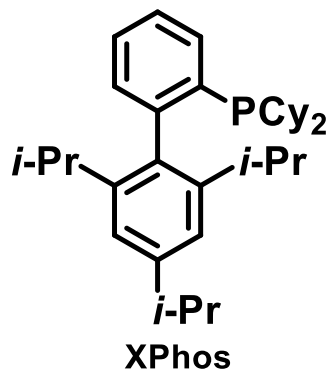


- Relative reactivity of organotributyltins (RSnBu_3) in the Stille reaction:
 alkenyl > aryl > allyl > alkyl



➤ The Stille reaction

- Electron-rich bulky ligands are the best ligands for the Stille reaction

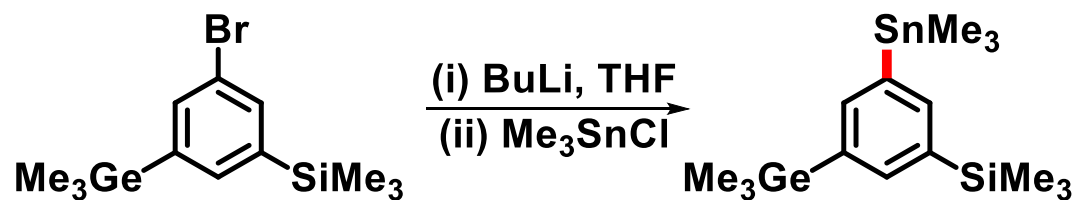


J. Am. Chem. Soc. **2003**, *125*, 3718

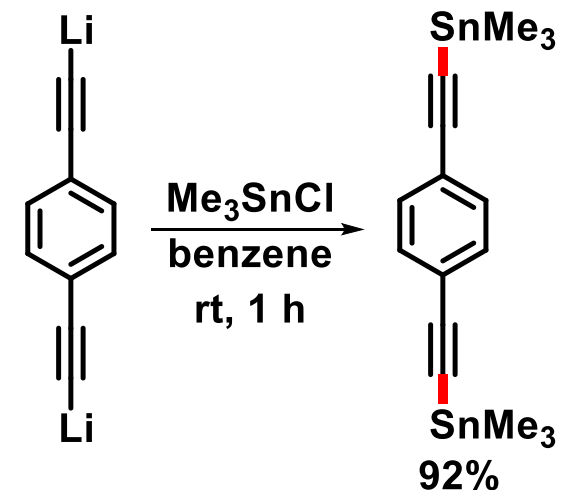
➤ The Stille reaction

- Synthesis of aryl-alkynylstannanes

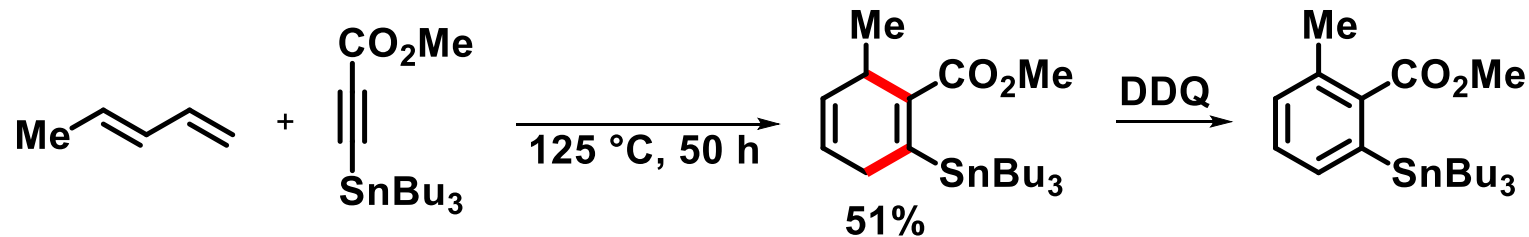
○ Lithiation



J. Organomet. Chem. **1983**, *251*, 149



○ Stannane modification

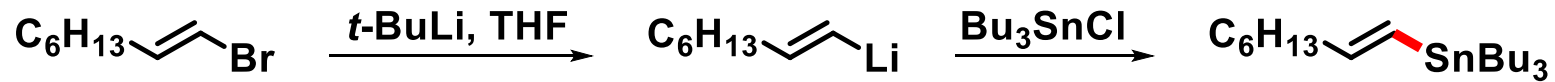


Tetrahedron **1989**, *45*, 1145

➤ The Stille reaction

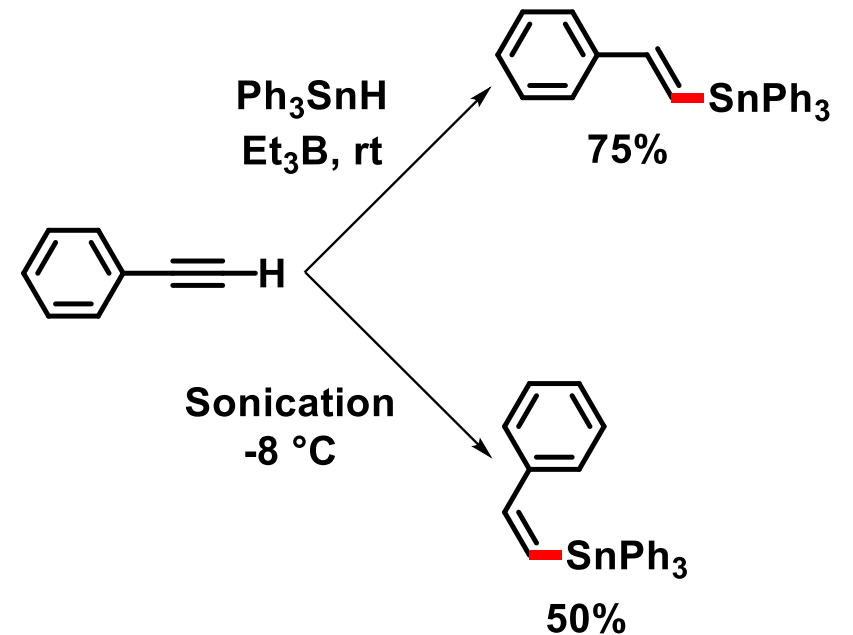
- Synthesis of alkenylstannanes

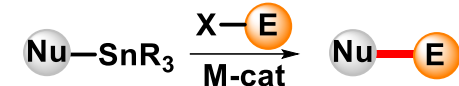
- Lithiation



Organometallics 1991, 10, 2424

- Hydrostannation



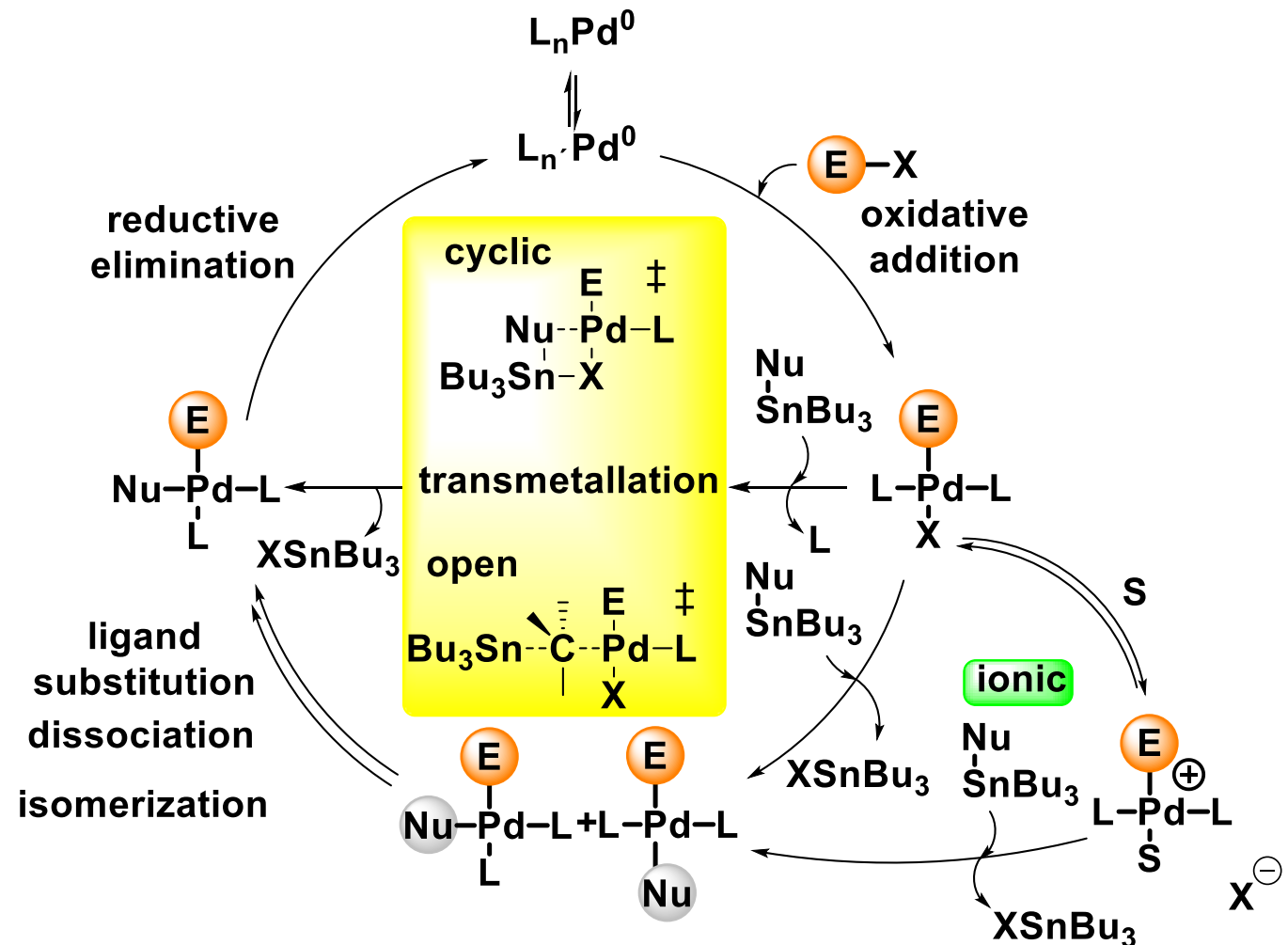
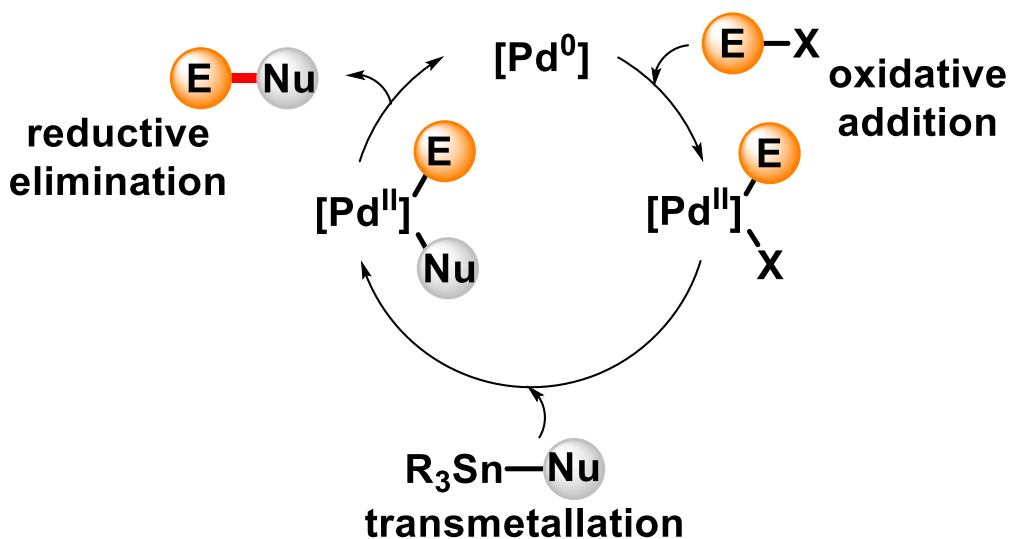


➤ The Stille reaction

- Advanced mechanism of the Stille reaction

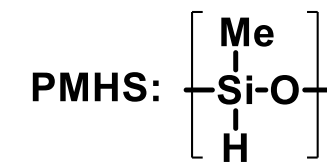
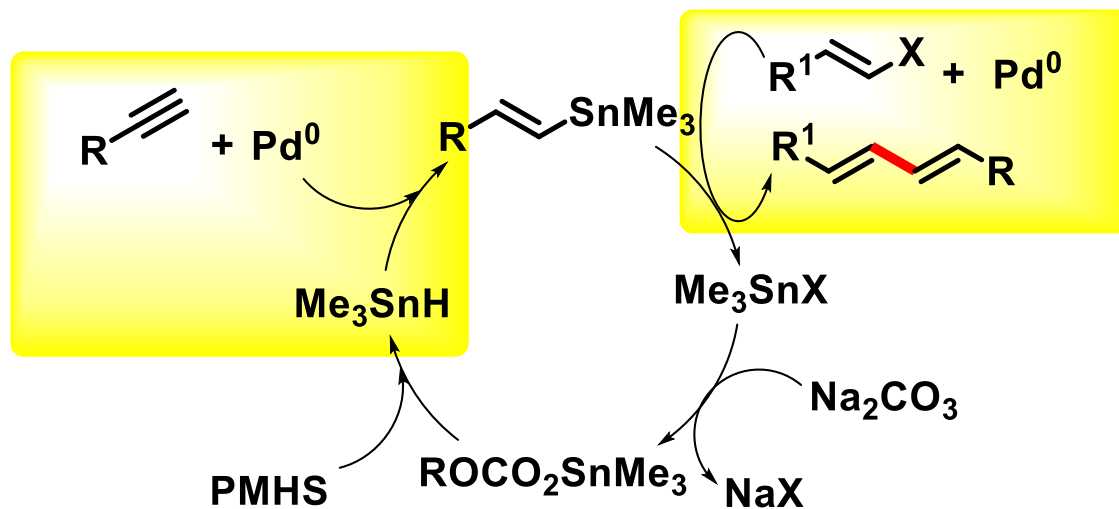
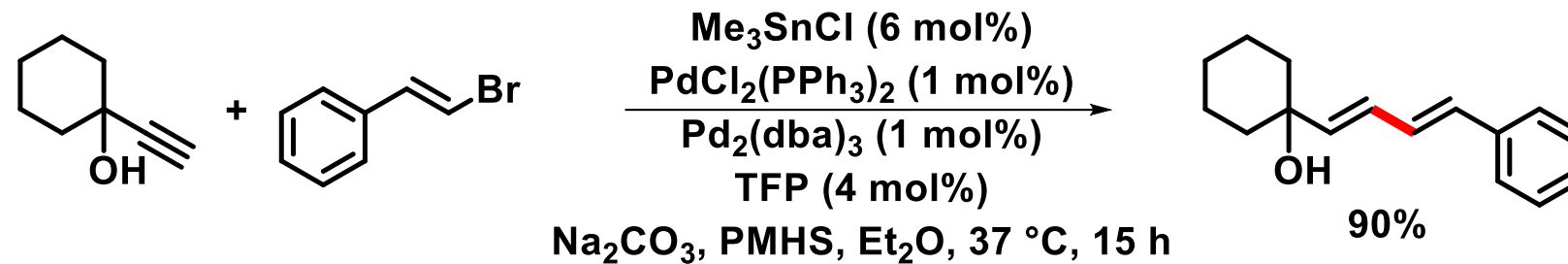
- Advanced mechanism

- Simplified mechanism

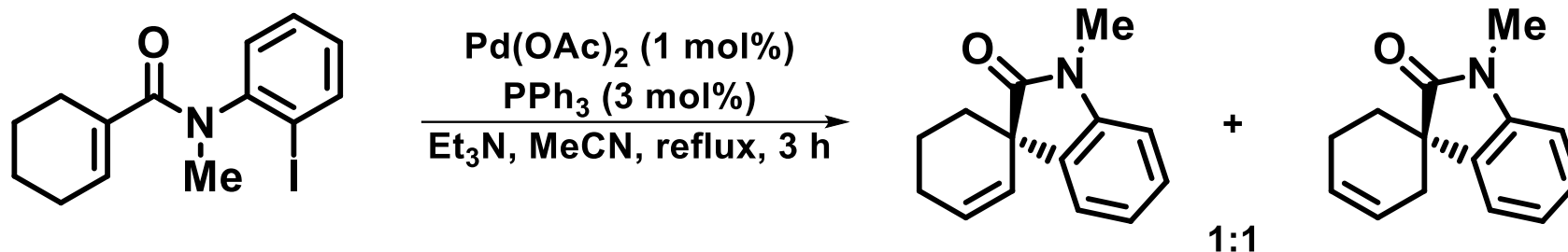
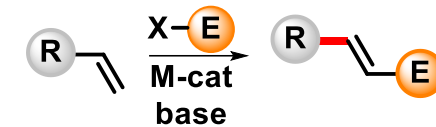


➤ The Stille reaction

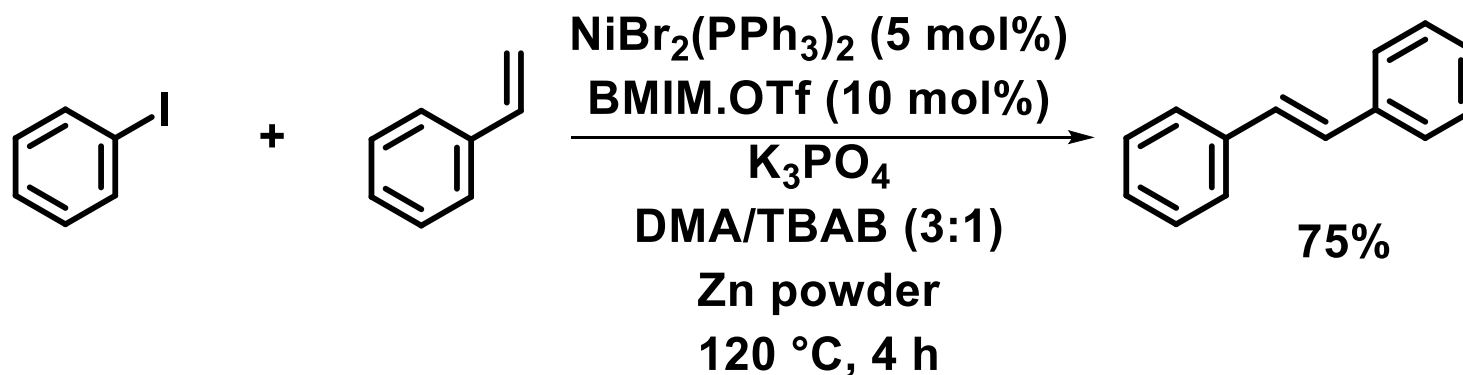
- Greener Stille reaction
 - Catalytic in tin



➤ The Heck reaction

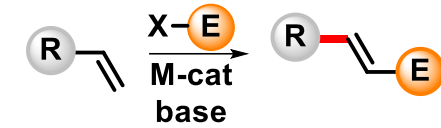


J. Org. Chem. **1987**, *52*, 4133



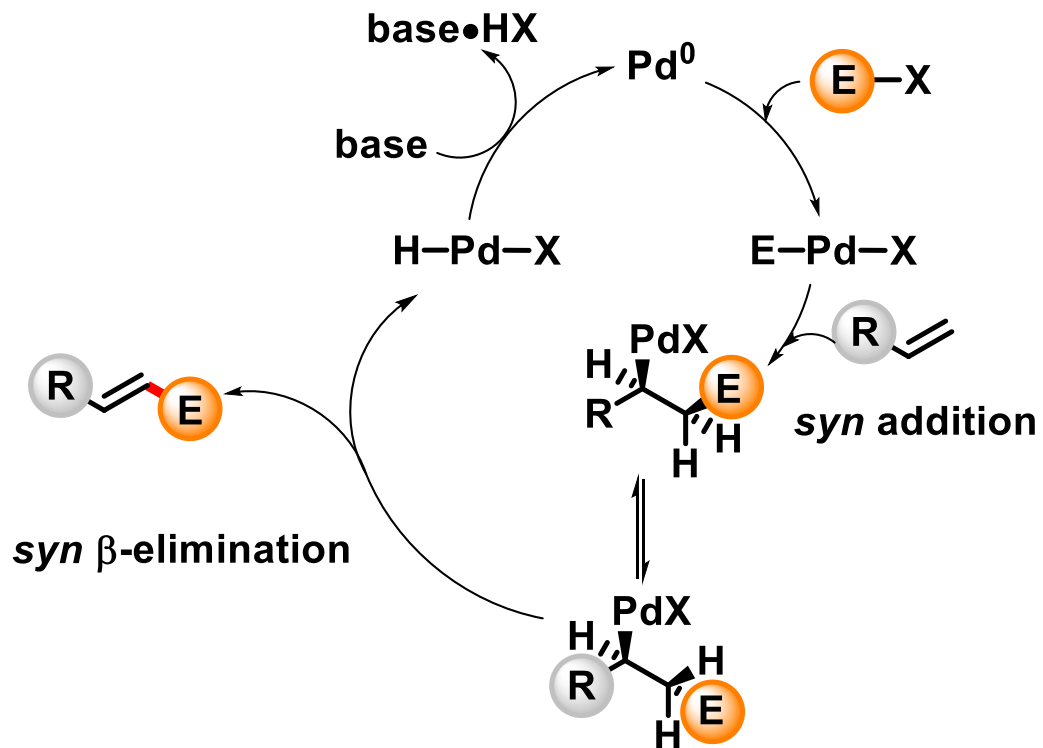
ChemistrySelect **2019**, *4*, 6913

➤ The Heck reaction

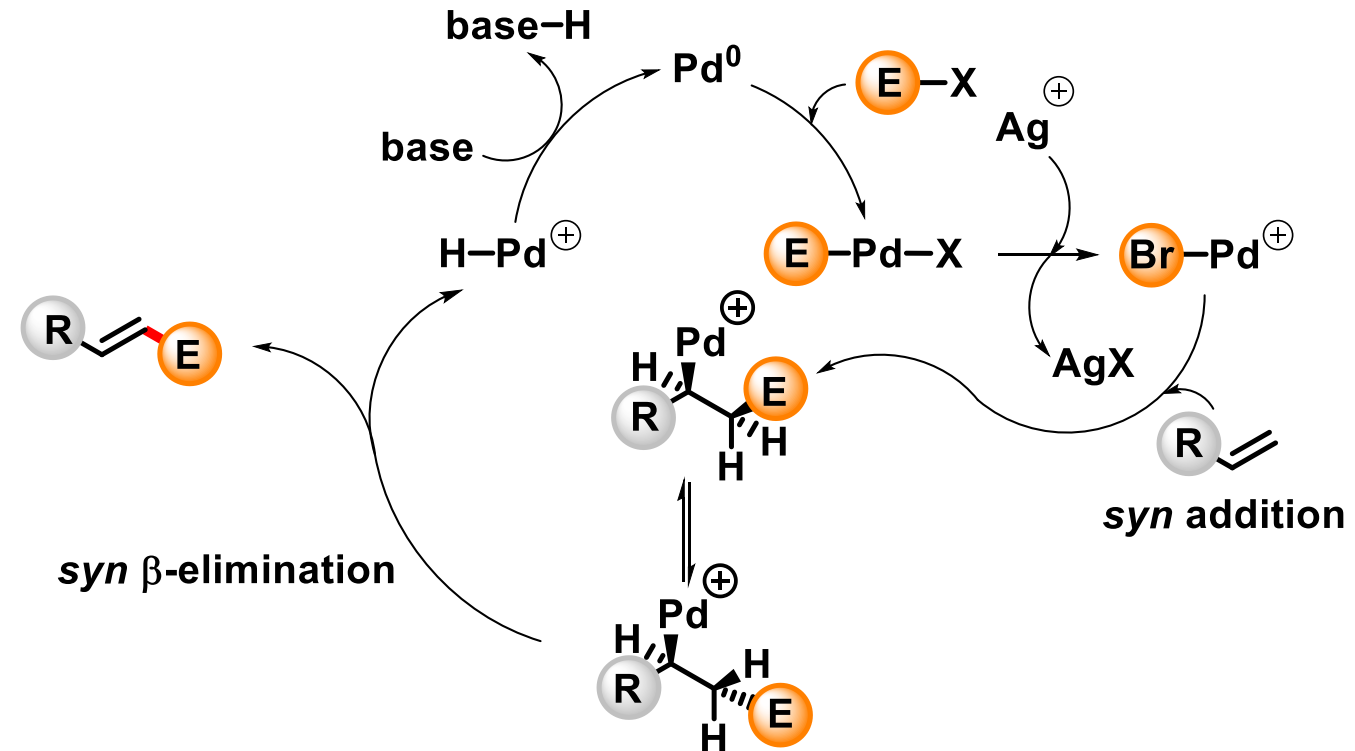


- Mechanisms of the Heck reaction

- Neutral mechanism



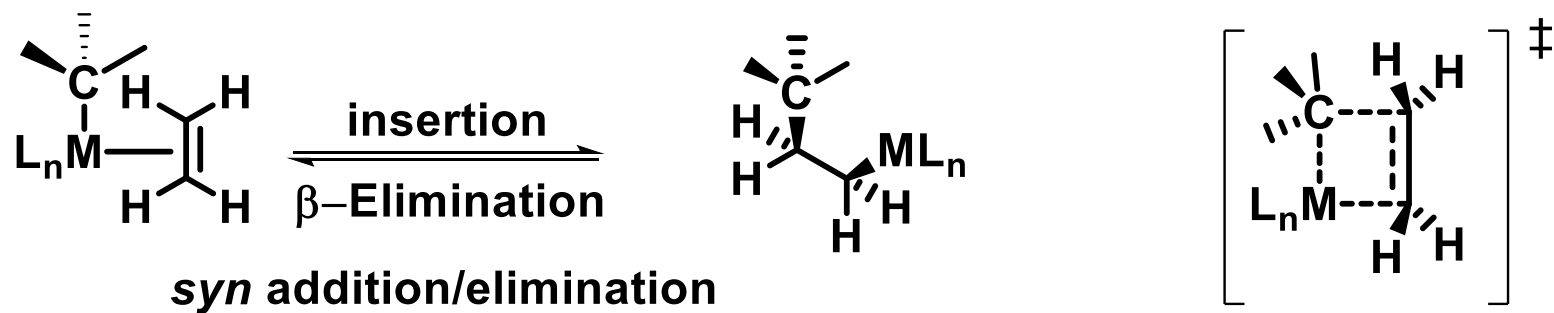
- Cationic mechanism



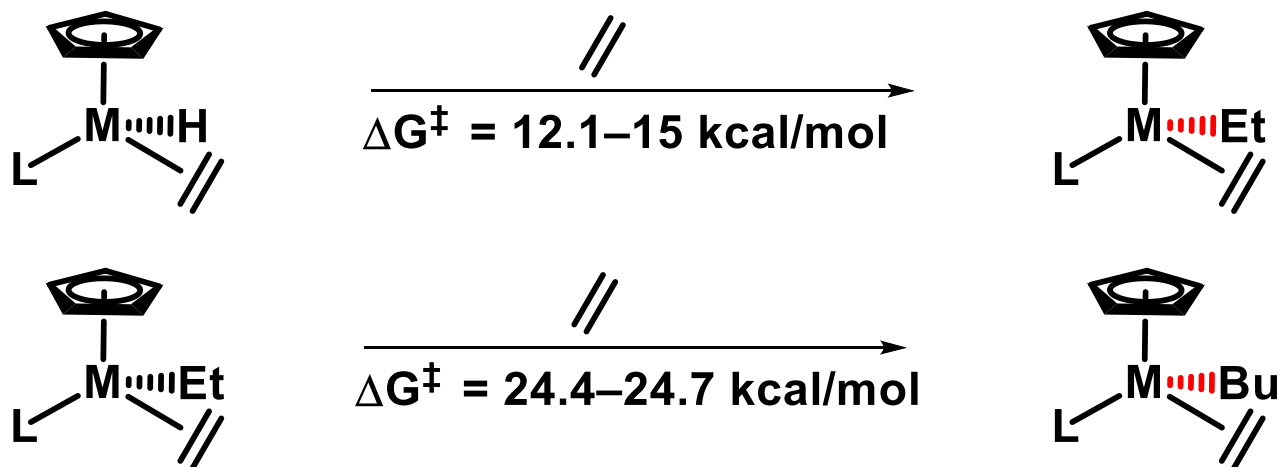
➤ Migratory insertion

- Insertion of alkenes into metal-carbon bonds

- Concerted transition state



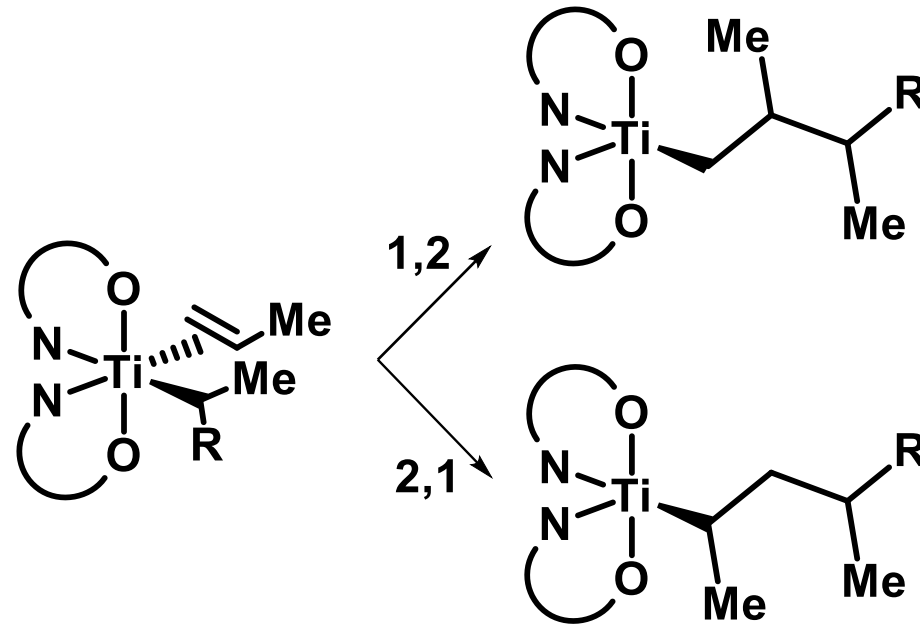
- Thermodynamically favorable but kinetically unfavorable



J. Am. Chem. Soc. **1992**, *114*, 10394

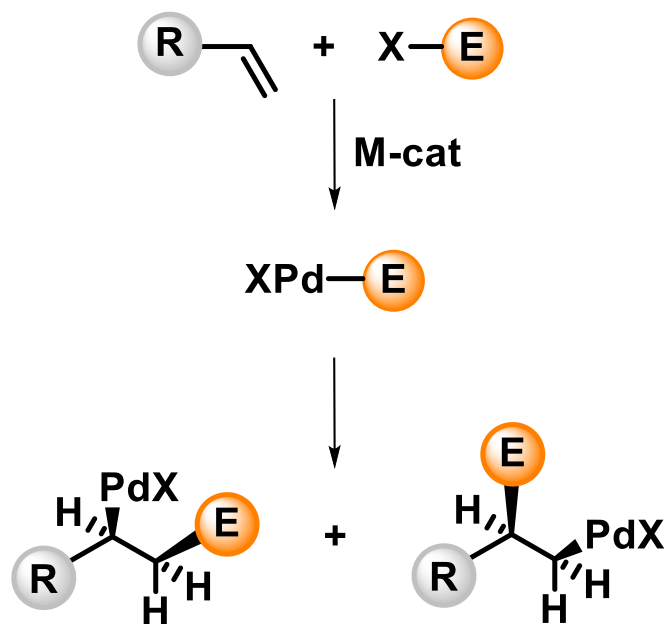
➤ Migratory insertion

- Insertion of alkenes into metal-carbon bonds
 - Markovnikov (2,1) or anti-Markovnikov (1,2) regioselectivity



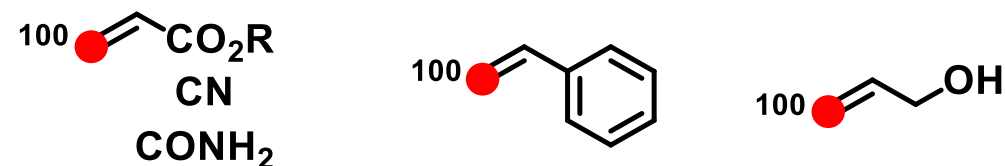
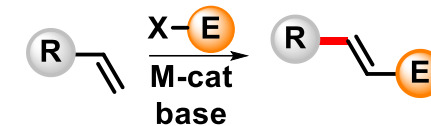
➤ The Heck reaction

- Regioselectivity of the Heck reaction

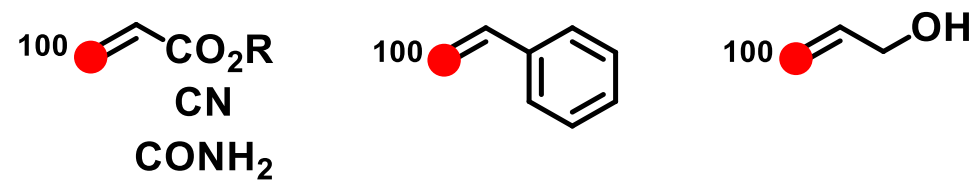


two possible regioisomers

- Affected by steric reasons



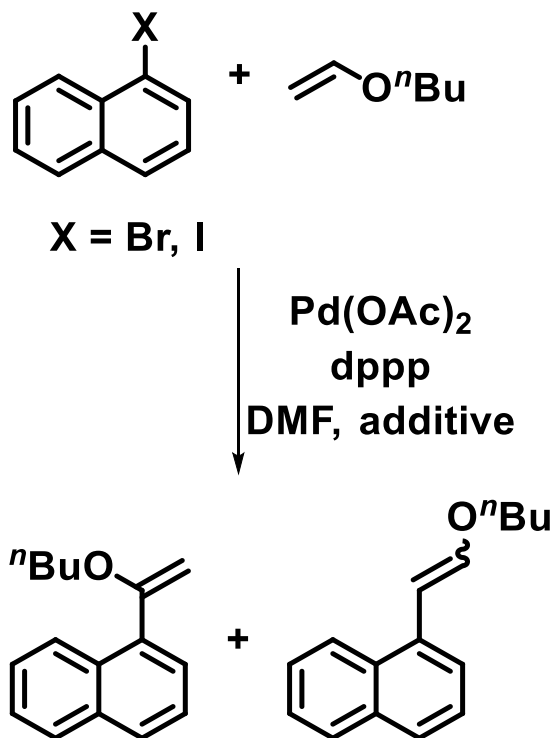
- Affected by electronic reasons



Acc. Chem. Res. 1995, 28, 2
J. Org. Chem. 1992, 57, 1481

➤ The Heck reaction

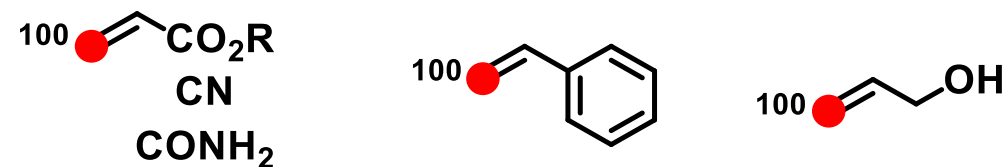
- Regioselectivity of the Heck reaction



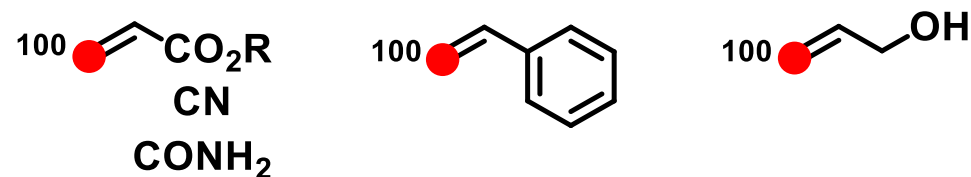
$\text{Et}_3\text{N, } 100\text{ }^\circ\text{C}$ 67/33

$\text{TIOAc, } 100\text{ }^\circ\text{C}$ >99/1

- Affected by steric reasons



- Affected by electronic reasons

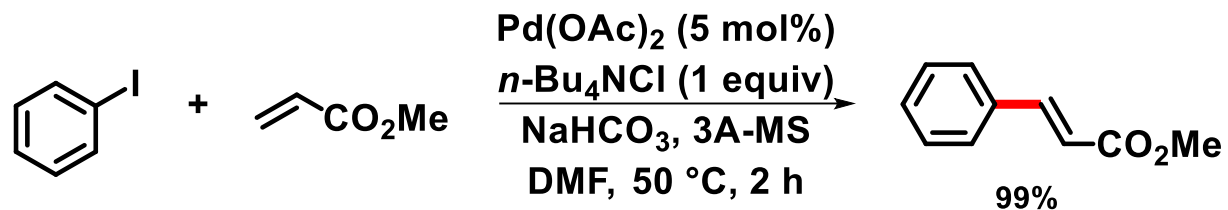


Acc. Chem. Res. 1995, 28, 2
J. Org. Chem. 1992, 57, 1481

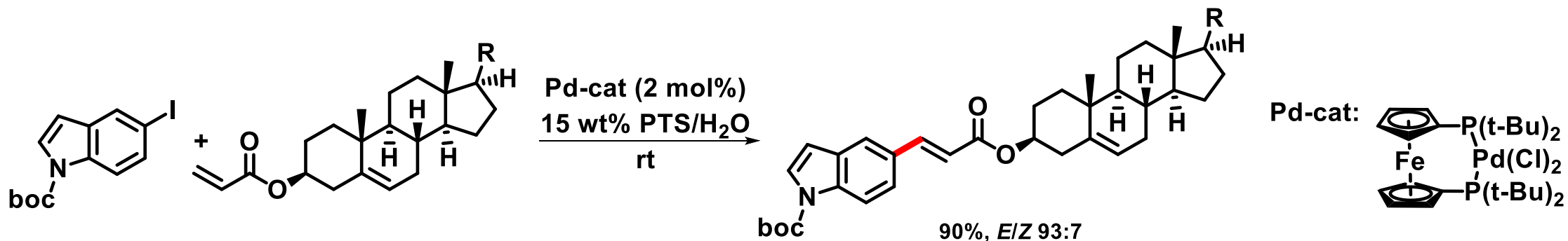
➤ The Heck reaction

- The Heck reaction under mild reaction conditions

- Jeffery conditions – use of ammonium salts



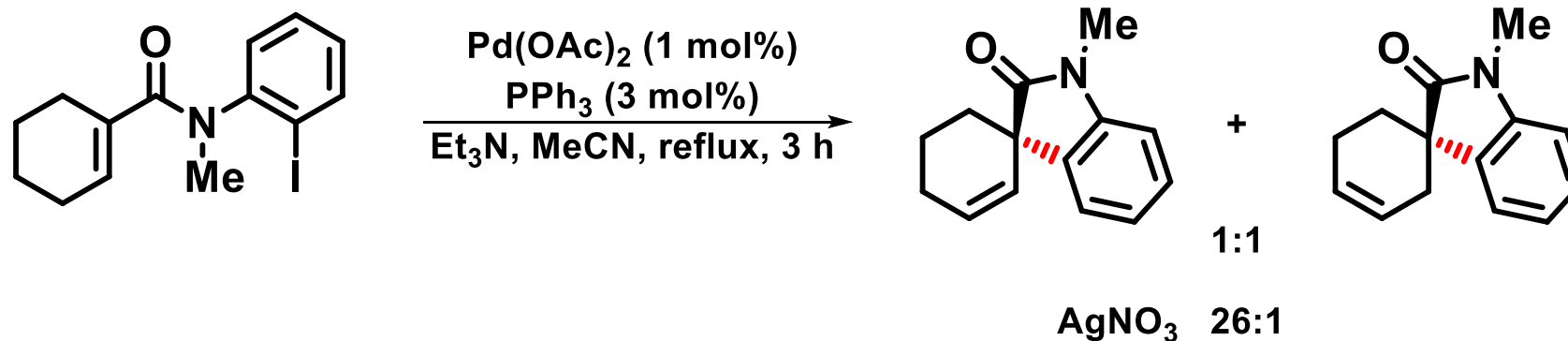
Tetrahedron 1996, 52, 10113



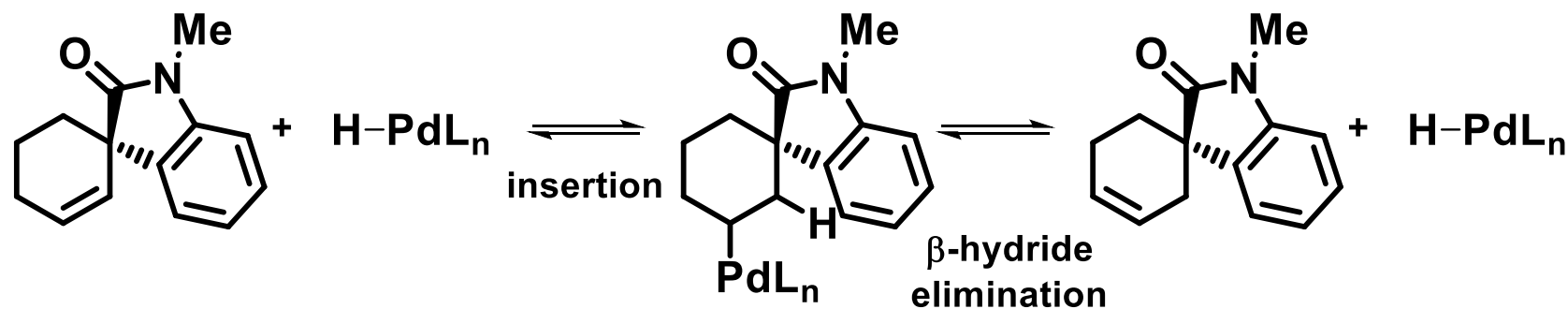
Org. Lett. 2008, 10, 1329

➤ The Heck reaction

- Double bond isomerization

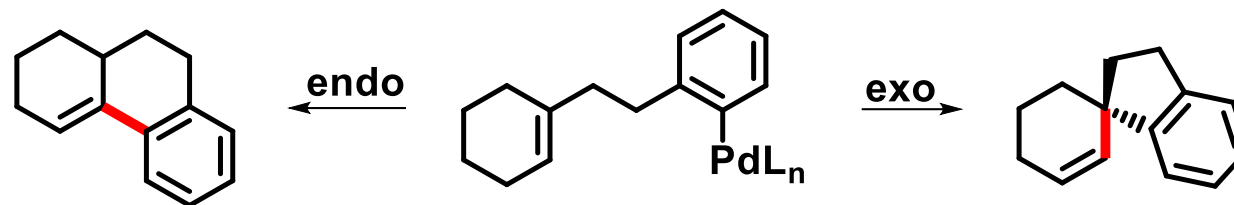


J. Org. Chem. 1987, 52, 4133

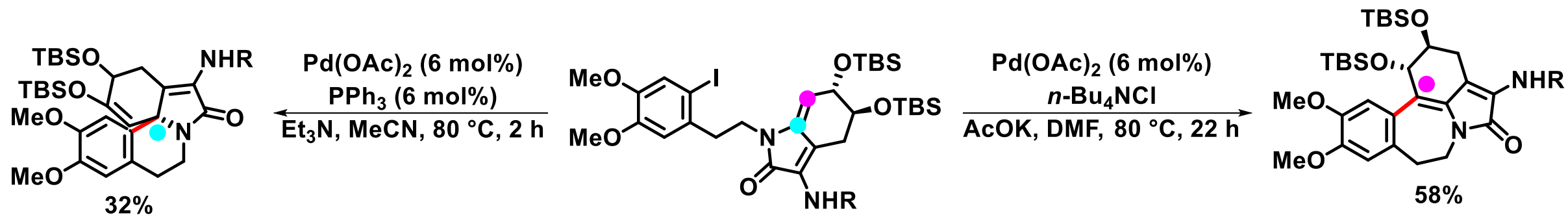


➤ The Heck reaction

- Intramolecular Heck reaction – exo x endo cyclization



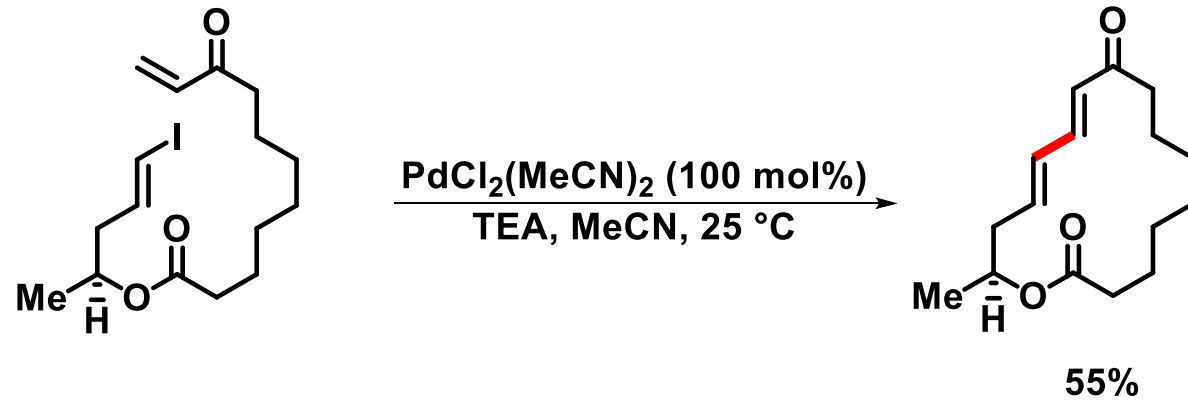
- Influenced by reaction conditions and substrate structure



J. Am. Chem. Soc. **1995**, *117*, 7834

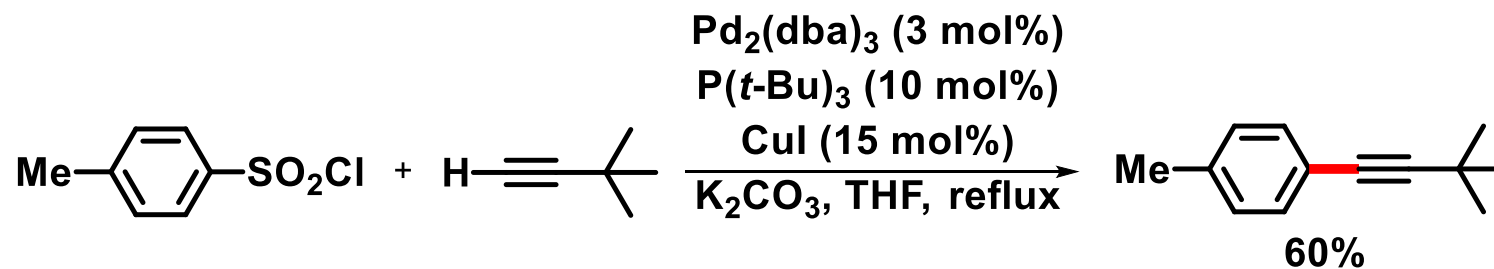
➤ The Heck reaction

- Intramolecular Heck reaction – excellent tool for macrolactonization

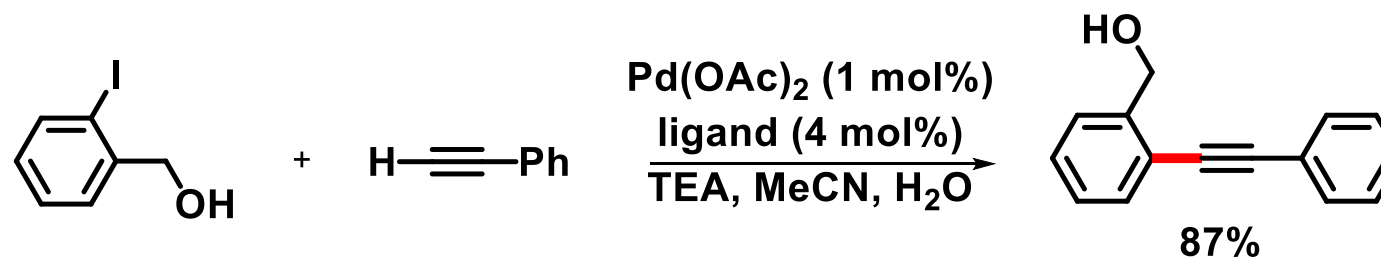


Tetrahedron **1981**, 37, 4035

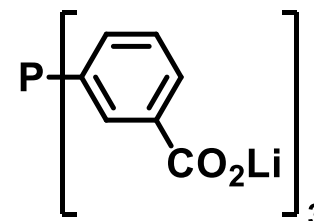
➤ The Sonogashira reaction



Adv. Synth. Catal. **2004**, 346, 1793



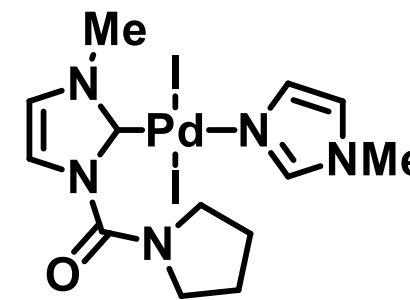
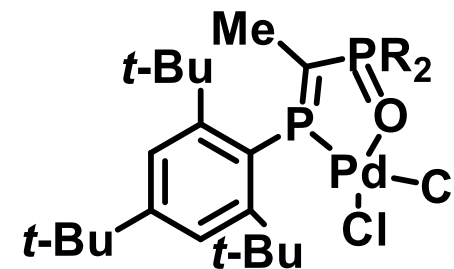
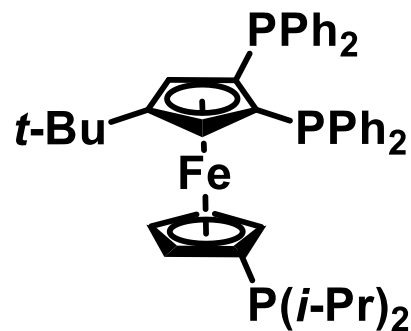
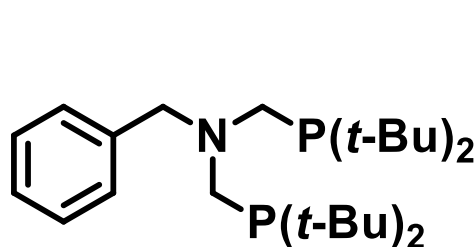
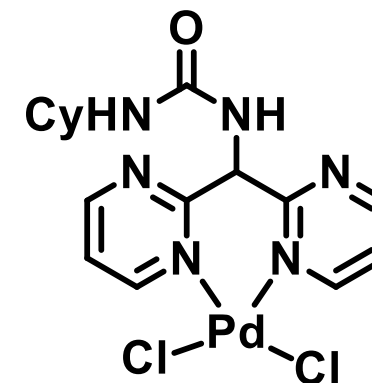
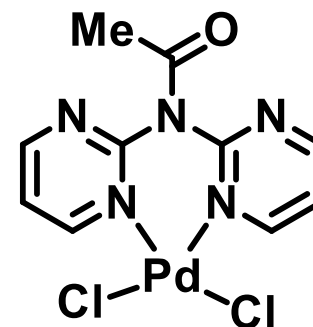
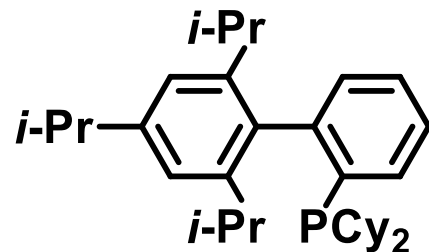
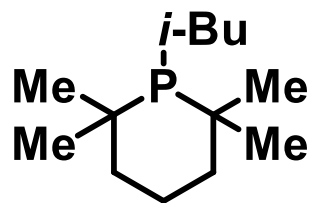
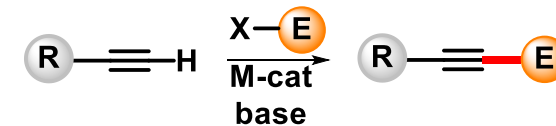
ligand:



Adv. Synth. Catal. **2004**, 346, 1733

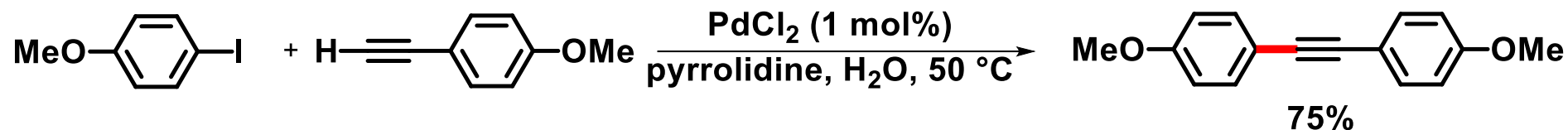
➤ The Sonogashira reaction

- Ligands for the Sonogashira reaction

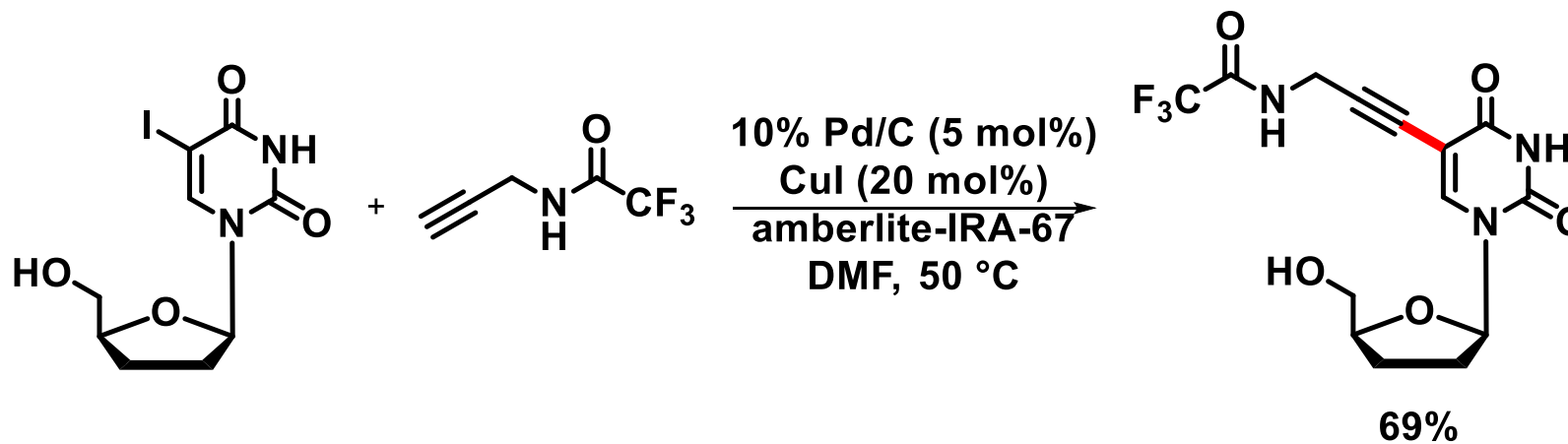


➤ The Sonogashira reaction

- Ligands-free Sonogashira reaction



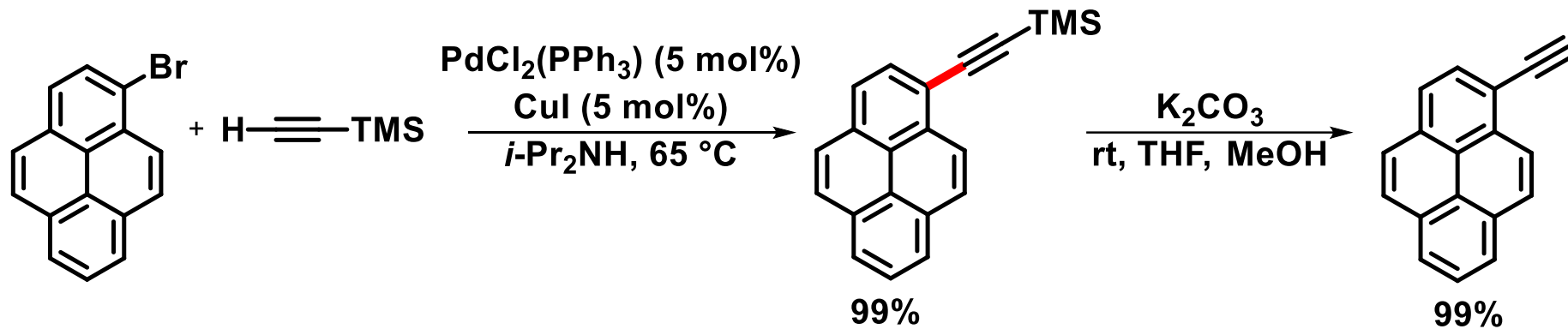
J. Org. Chem. **2005**, *70*, 391



Chem. Commun. **2005**, 4551

➤ The Sonogashira reaction

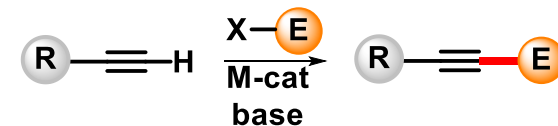
- Synthesis of terminal alkynes



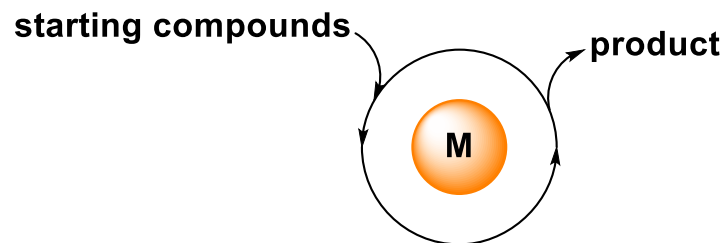
Chem. Commun. 2015, 12088

➤ The Sonogashira reaction

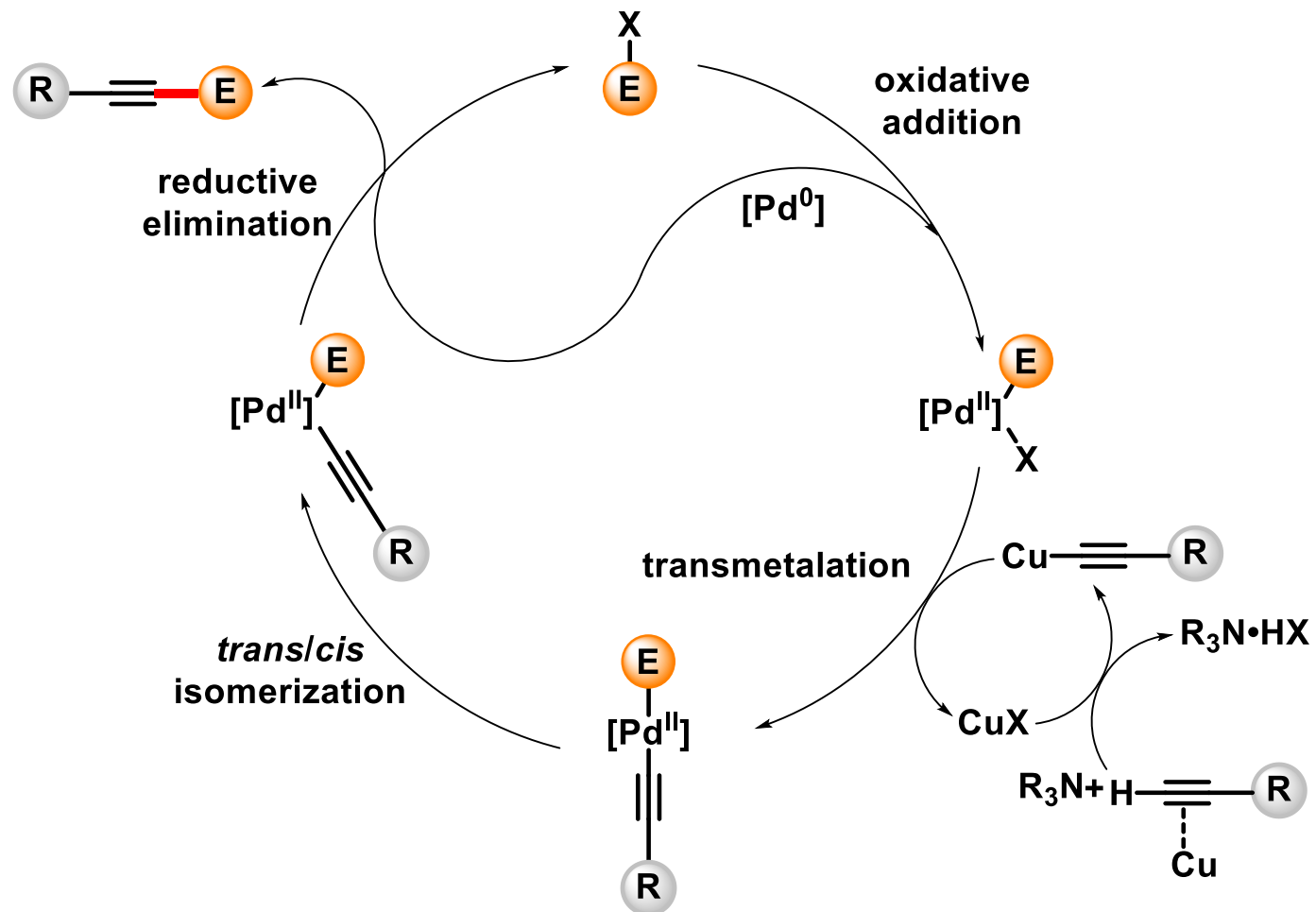
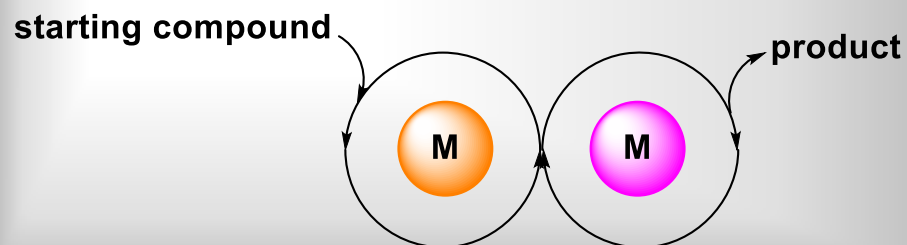
- Mechanism of the Sonogashira reaction



Monocatalysis



Cooperative catalysis



➤ The Sonogashira reaction

- Mechanism of the copper-free Sonogashira reaction

